

P2 Assessment of Polymers

**A Discussion of
Physical-Chemical Properties,
Environmental Fate,
Aquatic Toxicity,
and Non-Cancer Human Health Effects
of Polymers**

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P2 Assessment of Polymers

- **Definition of polymer for P2 assessment purposes**
 - A polymer is a chemical made up of covalently linked repeating units, generally with MWn >1,000
 - < 25% < 1000
 - < 10% < 500
- **(Q)SAR assessment of polymers may not be possible with some Sustainable Futures models and methods due to:**
 - Limited data sets
 - Large molecular weight (>1,000)
 - The presence of multiple species (mixtures)

P2 Assessment of Polymers

- **Polymer assessment based on combination of**
 - QSAR prediction
 - SAR - Read-across methods
 - Professional judgment
- **Primary reference (PDF Version included in hand outs)**
 - Boethling, Robert S. and Nabholz, J. Vincent
“Environmental Assessment of Polymers under the U.S. Toxic Substances Control Act”, pp. 187-234, in Ecological Assessment of Polymers Strategies for Product Stewardship and Regulatory Programs, Hamilton, John D. and Sutcliffe, Roger (eds.), (1997) Van Nostrand Reinhold.

Polymer Assessment Goals

- **Screening Level Assessment**
 - **Qualitative and/or Quantitative**
- **Looking for a snapshot of how the polymer will act (fate and toxicity)**
 - **Important to distinguishing between low concern, and NOT low concern**

Properties Affecting Polymer Assessment

- **To assess polymer, you need to know:**
 - **MW_n and % of Low Molecular Weight (LMW) components (%<1000, %<500)**
 - LMW components may need to be assessed separately
 - **Polymer Composition (Monomers)**
 - **Polymer Charge**
 - Neutral, Anionic, Cationic, Amphoteric
 - **Structural Features**
 - Reactive Functional Groups (RFGs)
 - Particle Size and Inhalability

Molecular Weight (MW) and Low Molecular Weight (LMW) Components

Three Categories of Polymers Identified

- **Category 1: $MW_n < 1,000$**
 - May be assessed as a “discrete” chemical (with representative structure)
 - **Category 2: $MW_n > 1,000$; $\geq 25\%$ with $MW < 1,000$ and $\geq 10\%$ with $MW < 500$**
 - Assess polymer and LMW materials (SAR modeling with representative structure)
 - **Category 3: $MW_n > 1,000$; $< 25\%$ with $MW < 1,000$ and $< 10\%$ with $MW < 500$**
 - Assess polymer using nearest analog approach or by estimation
- * Human health concerns for polymers of $MW_n > 10,000$ are discussed in the human health section

Polymer Charge

- **Nonionic, Anionic, Cationic, Amphoteric**
 - **Charge affects many aspects of the assessment**
 - **Physical properties**
 - **Fate**
 - **ecotoxicity**
 - **Human toxicity**

Cationic Polymers

- **Carbon-based polymer backbone**
- **Silicone-based backbone**
- **Natural-based backbone: chitin (glucosamine), tannin, starch**

Anionic Polymers

- **Poly(aromatic acids):**
bisphenolsulfones, cresols,
phenol, biphenylsulfones,
biphenylethers, naphthalene,
benzene
- **Poly(aliphatic acids)**

Example: Dividing up Dyes

- **Nonionic**
- **Cationic**
 - delocalized charge, localized charge +1, +2, +3, +4, etc; then, triphenylmethanes, acridines, phenothiazines, thiazoliums, azo, anthraquinones, phthalocyanines
- **Anionic:**
 - number of acids 1,2,3,4, etc; then, aminoanilines, anthraquinone, anilines, phenols, benzothiazoles, FWAs, chelated: Cu, Cr, Co, Fe.
- **Amphoteric**

Example: Dividing up Surfactants via Hydrophile

- **Nonionic**
 - ethoxylates; polyalcohols; alpha,omega-dialkyl-ethoxylates; TWEENS
- **Cationic**
 - N, P, S, number of dominant alkyls, ETHOMEENS, N-ethoxylates, guanidines
- **Anionic**
 - type of acid, ethoxylated
- **Amphoteric**

Structural Features

- **Reactive Functional Groups (RFGs)**
 - **Examples include, but are not limited to: acrylates/methacrylates, epoxides, phenols, sulphonates, Isocyanates, etc.**
- **Physical Features**
 - **Inhalability/particle size**
 - **Swellability**
 - **Fibrous properties**
- **Primarily affects mammalian toxicity**

Physical Properties Assessment

- **Based primarily on size of polymer**
- **Charge and structural features also play a role**
- **Most polymers will fit a general trend**
 - **See Interpretive Assistance Document for Polymers provided in hand-out material**

Physical Properties (cont.)

- **Vapor pressure – generally very low (10^{-8} mm Hg)**
- **Henry's Law constant – generally very low (10^{-8} mm Hg)**
- **Water solubility**
 - **Neutral – usually insoluble**
 - **Ionic – may be dispersible**

Physical Properties (cont.)

- **Soluble**
- **Dispersible**
- **Micro emulsions / Macro emulsions**
- **Dispersed solid particles**
- **Gels**
- **Micro micelles / Macro micelles**
 - **Surfactants**

Environmental Fate

- **Environmental Fate Assessment**
 - based on size, charge, and polymer make up (monomers and end groups)
- **The goal is to establish how the chemical will behave in the environment**
 - Partitioning – where it will go?
 - Persistence – how long it will last?
- **Screening Level Assessment**

Environmental Fate - Partitioning

- **POTW Removal**
 - **POTW removal is based on MW_n and charge**
 - **Cationic, Amphoteric, Nonionic, and Insoluble and Non-dispersible Anionic**
 - **Ranges from 50% at MW_n of <500 – 90% at MW_n of >1,000**
 - **Soluble or Dispersible Anionic**
 - **Ranges from 0% at MW_n of <5,000 – 90% at MW_n of >50,000**

Environmental Fate - Partitioning

- **Soil Mobility**
 - **Polymers tend to have poor mobility in soil**
- **Volatilization from water**
 - **Polymers tend to be insoluble in water, but do not volatilize from water**
- **Bioconcentration Factor (BCF)**
 - **MW_n <1,000, use EPI Suite**
 - **MW_n >1,000, Low BCF Concern (100 can be used for modeling purposes)**

Environmental Fate - Partitioning

- **Overall Partitioning Picture**
 - **Generally polymers will partition to**
 - **Soil, suspended particles, sediments, and sludge**
 - **Soluble and/or dispersible polymers may remain partially in water**
 - **Partitioning to air only as particulate (dust), not usually significant**

Environmental Fate - Degradation

- **Hydrolysis**
 - Hydrolysis of susceptible groups solubility dependant
 - Not usually a major removal route
- **Air oxidation**
 - Poor partitioning to air
 - Presence of polymer as particle in air reduces potential removal rate
 - Not a major route of removal

Environmental Fate - Degradation

- **Spontaneous Degradation**
 - This will be polymer specific
 - In most cases it will be a known property
- **Polymerization**
- **Biodegradation**
 - In most cases polymers will be resistant to biodegradation
 - Due to size and hydrophobicity
 - Exceptions are usually polymers designed for rapid biodegradation

Environmental Fate - Overall Picture

- **General Trends**
 - **Polymers will tend to partition to**
 - **Soil, suspended particles, sediments, and sludge**
 - **High persistence concern**
 - **Low concern for bioconcentration (BCF)**
- **Exceptions will exist**
 - **High solubility**
 - **Polymers designed for degradation**
 - **Polymers with low MW_n**
 - **Insoluble polymers in a solvent**

Aquatic Toxicity

- **Assessment method varies**
 - **Main grouping is based on polymer charge (Neutral, Anionic, Cationic, and Amphoteric)**
- **Insoluble or non-dispersible polymers generally have low aquatic toxicity hazard concern (regardless of charge)**
 - **Not soluble or bioavailable**
 - **Exceptions may include finely divided particles**
- **For polymer with MW_n of $<1,000$ (category 1) or those with significant amounts of LMW components (category 2), ECOSAR may be used**

Aquatic Toxicity - Neutral Polymers

- **Nonionic polymers tend toward low hazard concern**
- **Exception is neutral polymers that are blocked for use as a surfactant or dispersant, these may exhibit toxicity**
 - **Use nearest analog approach**
 - **Or SAR**

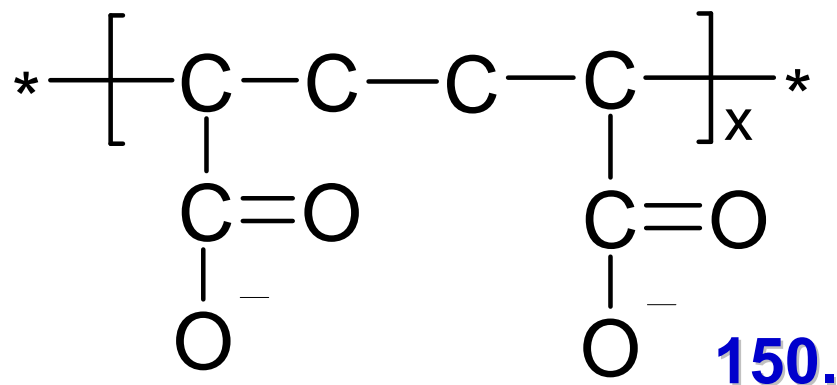
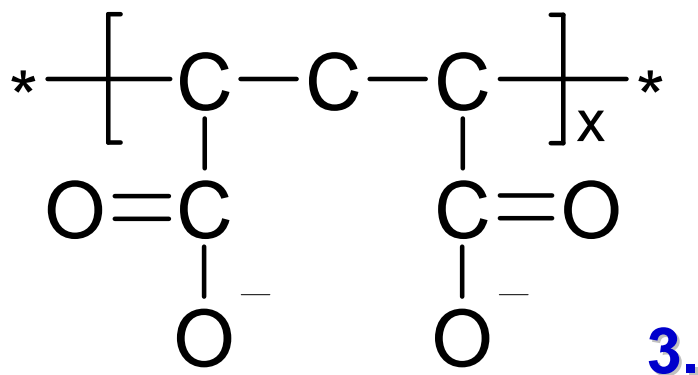
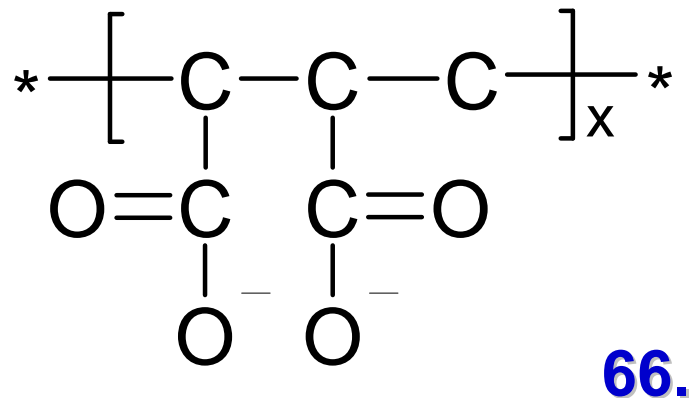
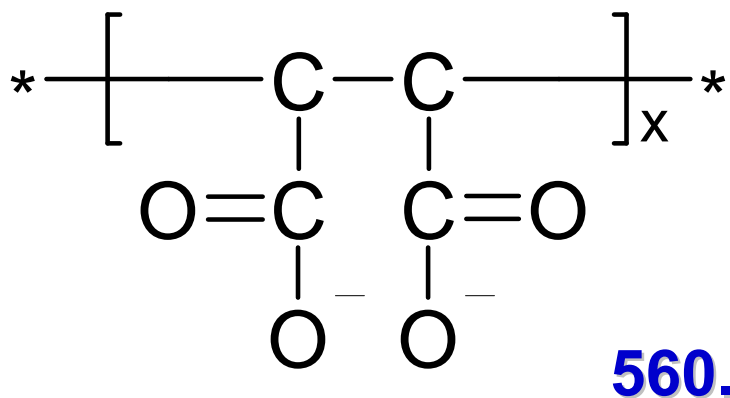
Aquatic Toxicity - Anionic Polymers

- **Polyanionic polymers that are soluble or dispersible may exhibit ecotoxicity**
- **Two main classes**
 - **Poly(aromatic acids)**
 - **Poly(aliphatic acids)**
- **Nearest analog approach**
 - **Tables with many analogs are collected in the “Environmental Assessment of Polymers under the U.S. Toxic Substances Control Act” chapter**

Polyanionic Polymers

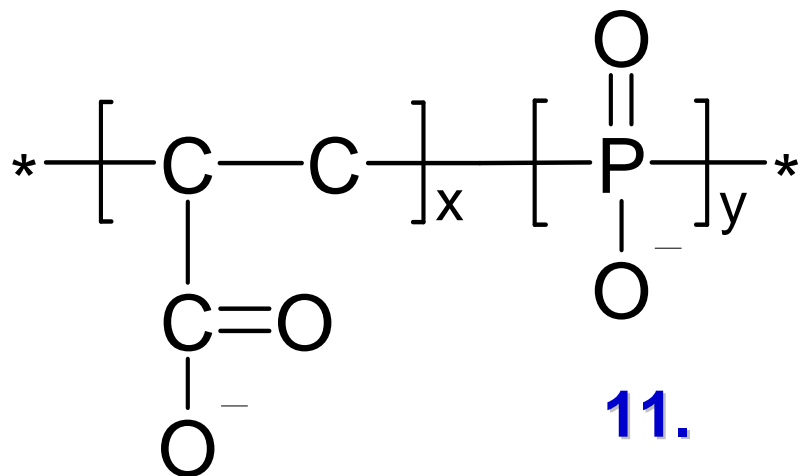
Poly (Carboxylic Acids)

Structure and Green Algal 96-h EC50 (mg/L)

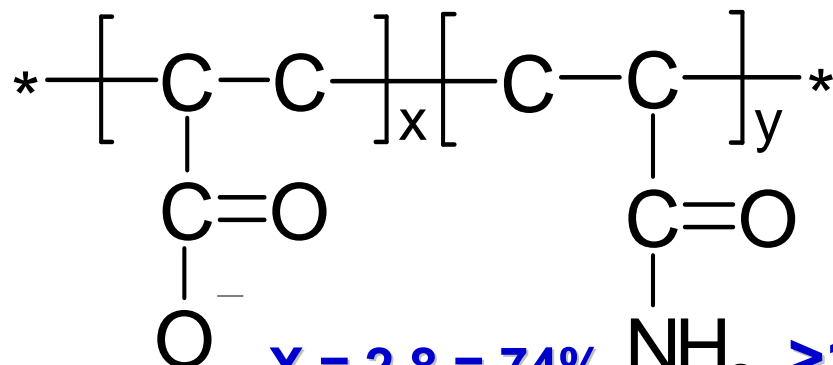


Polyanionic Polymers Poly (Carboxylic Acids)

Structure and Green Algal 96-h EC50 (mg/L)

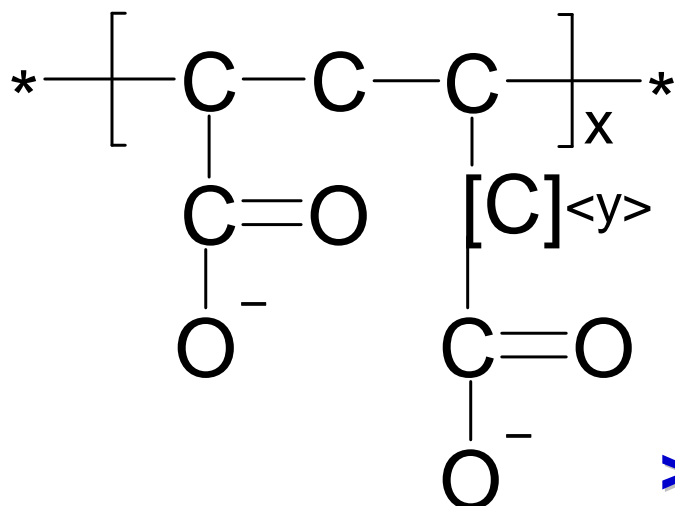


11.



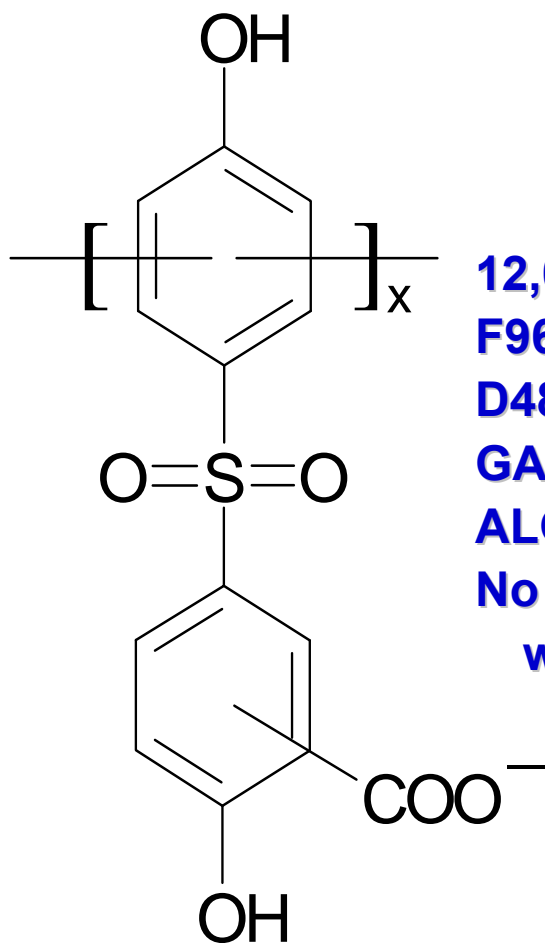
$x = 2.8 = 74\%$
 $y = 1.0 = 26\%$

$\text{NH}_2 > 100.$

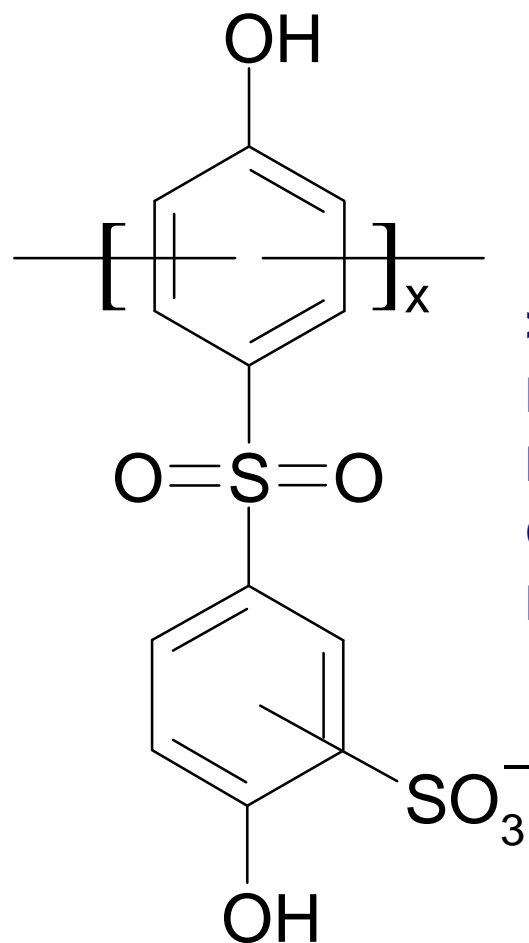


$> 100.$

Polyanionic Polymers Poly (Aromatic Sulfonates)

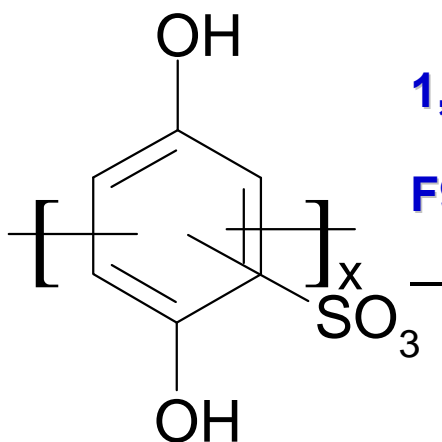


12,000 / 0 / 0
F96 = 70. mg/L
D48 = 90.
GA96 = 40.
ALGALICIDAL
No Mitigation
with Ca²⁺



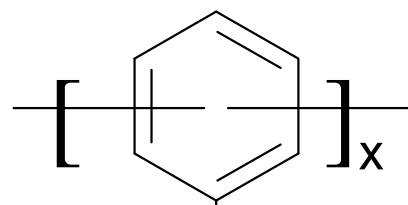
> 1,000 / 40 / 15
F96 = 5. mg/L
D48 = 145.
GA96 = 20.
No Mitigation
with Mg²⁺

Polyanionic Polymers Poly (Aromatic Sulfonates)



1,000 / ? / ?

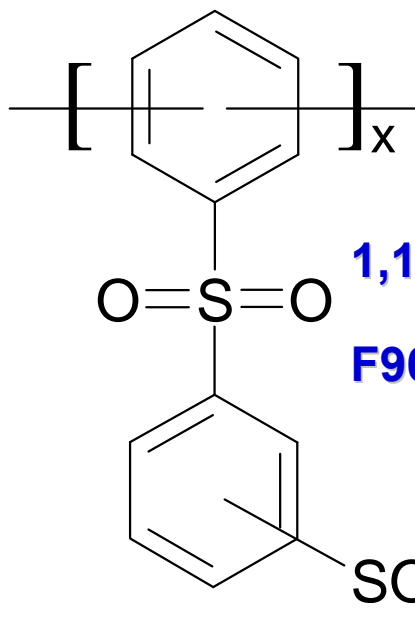
F96 = 2. mg/L



600 / 90 / 40

F96 = 30. mg/L

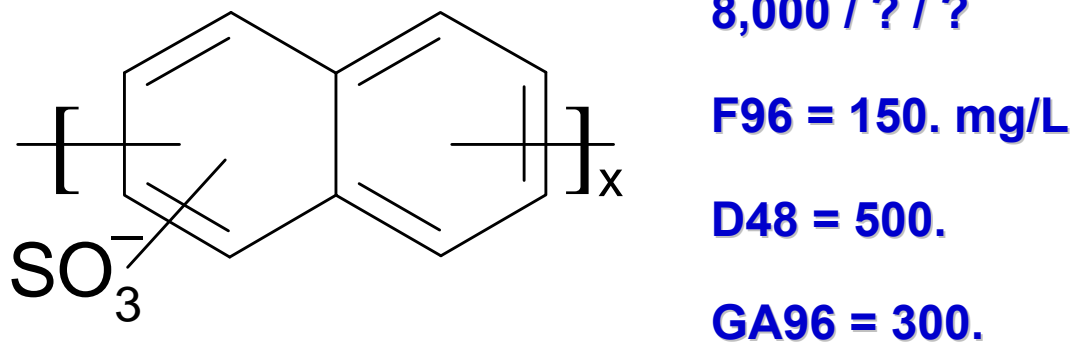
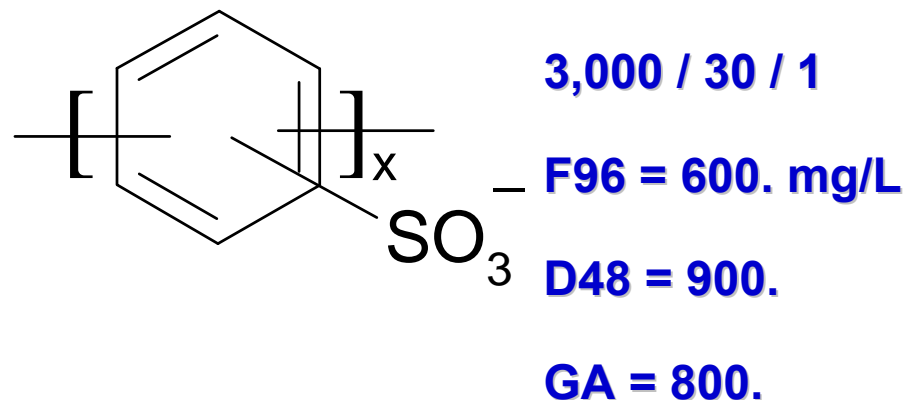
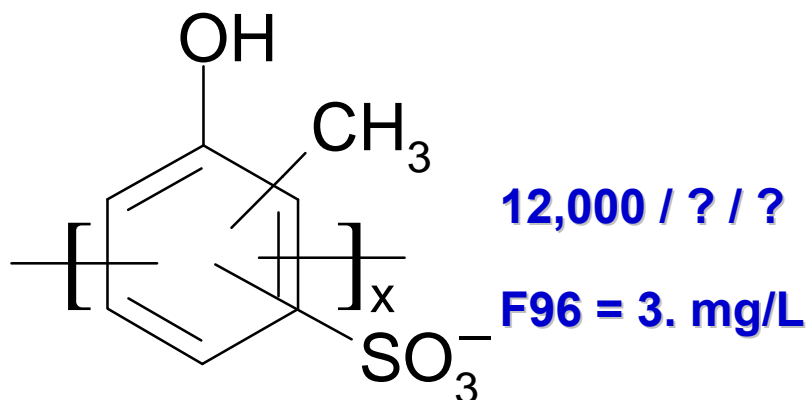
D48 = 120.



1,100 / ? / 0

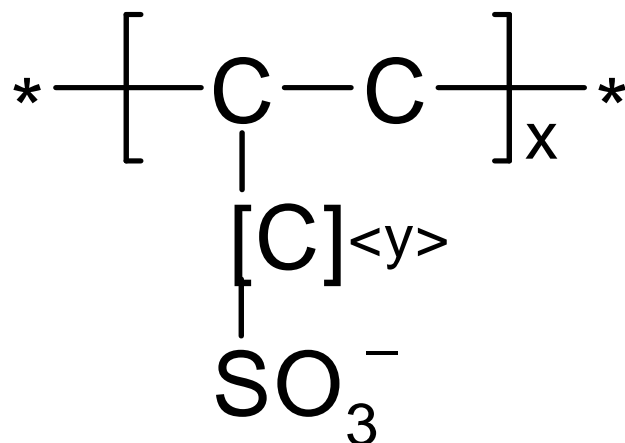
F96 = 3. mg/L

Polyanionic Polymers Poly (Aromatic Sulfonates)



SAR POLYMERS

Polyanionic Polymers Poly (Aliphatic Sulfonates)



3,000 / 30 / 10

F96 > 500. mg/L

D48 = 800.

GA96 = 800.

Aquatic Toxicity - Cationic and Amphoteric Polymers

- **Cationic polymers that have a net positive charge or that may become positive may pose a hazard concern for ecotoxicity**
 - **Cationic atoms of concern include (but are not limited to): Nitrogen, phosphorus, and sulfur**
 - **Nitrogen is the cationic group in 99% of cases**
 - **Nitrogens in or on an aromatic ring, amides, nitriles, nitro groups, and carbo diimides are not considered**

Aquatic Toxicity - Cationic and Amphoteric Polymers (cont.)

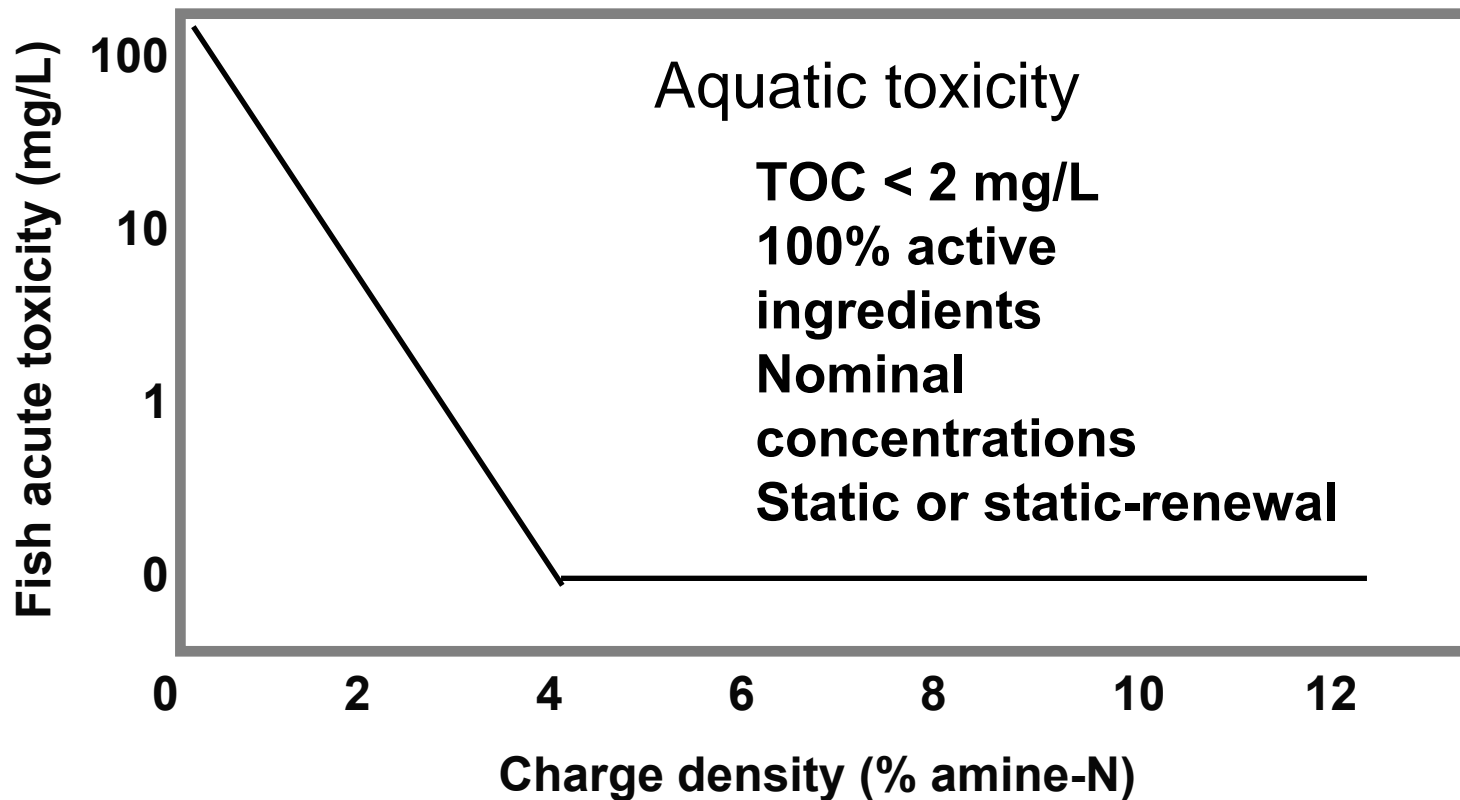
- **Factors in ecotoxicity estimation**
 - **Percentage of amine nitrogen (%A-N) or other cation by weight**
 - Nitrogens in or on an aromatic ring, amides, nitriles, nitro groups, and carbo diimides are not considered
 - **Amphoteric polymers**
 - %A-N is adjusted based on cation-to-anion ratio (CAR)
 - **Backbone**
 - SARs available for carbon based, silicon based, and naturally occurring polymer backbones

Aquatic Toxicity - Cationic and Amphoteric Polymers (cont.)

- **Toxicity may be mitigated by water hardness**
 - Mitigation Factor (MF) equations included, also based on %A-N
- **Toxicity is estimated by:**
 - Choosing correct SAR for backbone
 - Calculating %A-N
 - Calculating base toxicity
 - Calculating MF
 - Adjusting toxicity based on MF to give final endpoint

Polycationic Polymers

Polyamine Polymers



Applying Mitigation Factors

- **Cationic and Amphoteric Polymers: Mitigation of Toxicity**
 - **Standard aquatic hazard testing media (OECD) usually has a low total organic content (TOC) which may result in artificially high toxicity of polycationic and amphoteric polymers in those media.**

Mitigation Factors

- **To correct for TOC in actual surface water a mitigation factor (MF) has been calculated, based on testing in standard media**
- **The MF is dependent on the overall charge density (%A-N) for the polymer**
- **Several conditions and/or structural features have been shown to affect the mitigation factor**
- **See page 8 of Interpretive Assistance Document for Polymers which provides further details**

Non-Cancer Human Health Concerns

- **Traditional U.S. EPA Human Health Effects Assessment based on:**
 - **Nearest analog approach**
 - **OncoLogic**
 - **U.S. EPA Chemical Categories Report**
<http://www.epa.gov/oppt/newchemicals/pubs/chemcat.htm>
 - **Structural Features**

Non-Cancer Human Health Concerns

- **Traditional Health Assessment approach is relevant for polymers**
 - **Chemical Categories**
 - **Nearest analog approach**
 - **LMW components and residual monomer(s) may need to be considered**
- **Special considerations for polymers**
 - **Large inhalable polymers**
 - **OncoLogic is run differently**

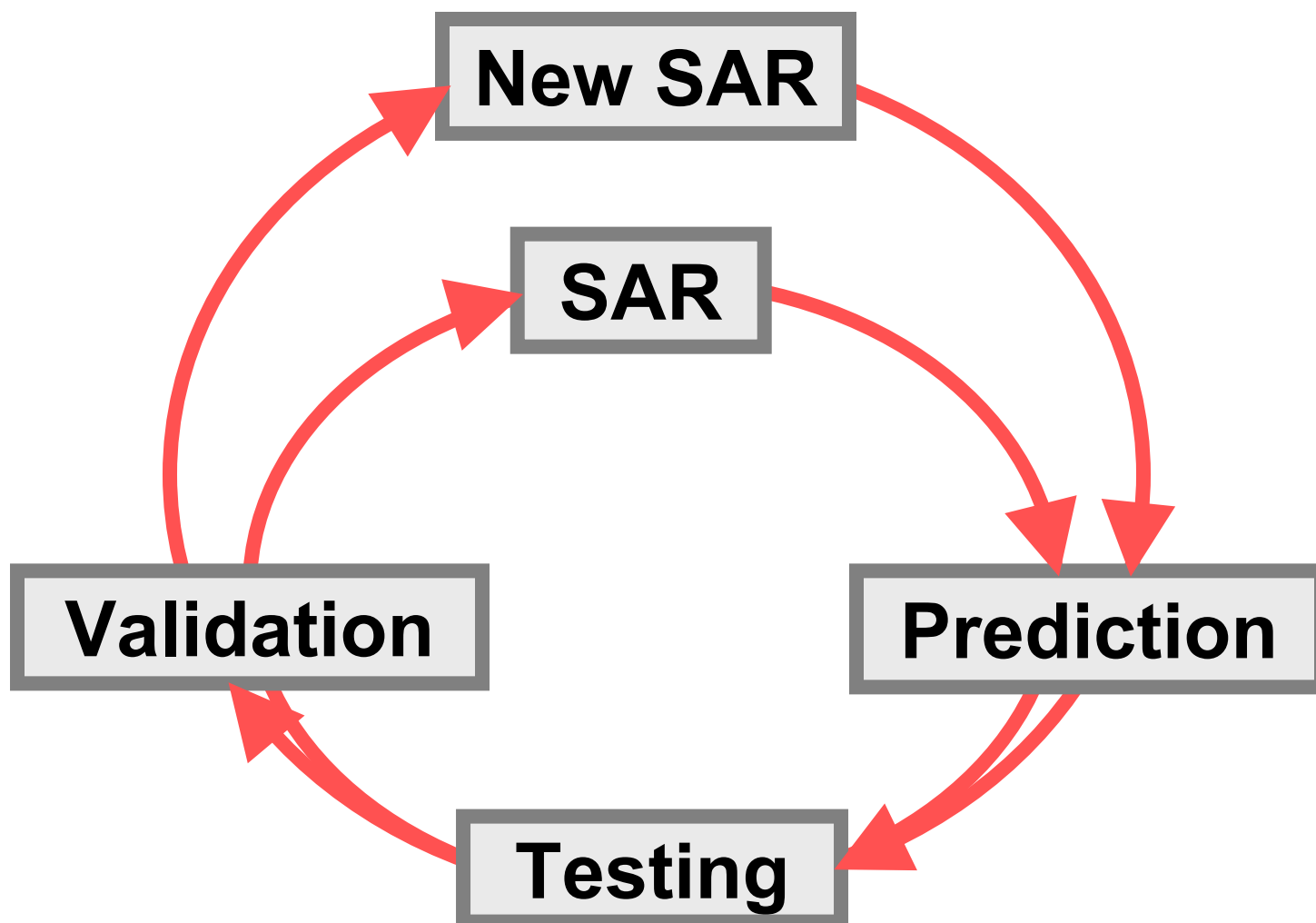
Non-Cancer Human Health Concerns

- **Large Inhalable Polymers**
 - **Polymers with MW_n of $>10,000$ are generally of concern only for lung effects**
- **There are three further distinctions**
 - **Soluble – Not generally a concern**
 - **Insoluble – Concern may exist**
 - **Swellable – Concern may exist**

Non-Cancer Human Health Concerns

- **Cationic/Amphoteric Binding to Lungs**
 - **Binding to Lung Membrane**
 - Charge
 - Reaction
 - Alkoxysilanes
- **Waterproofing of lung membranes**
 - **Anti-stain aerosols**
 - **Anti-stain polymers**
 - Polysilicones

**SARs, QSAR Models, and Assessment Methods for Polymers
Continually Updated as New Information is Available**



References

- **Boethling, Robert S. and Nabholz, J. Vincent**
“Environmental Assessment of Polymers under the U.S. Toxic Substances Control Act”, pp. 187-234, in Ecological Assessment of Polymers Strategies for Product Stewardship and Regulatory Programs, Hamilton, John D. and Sutcliffe, Roger (eds.), (1997) Van Nostrand Reinhold.
 - Included on Sustainable Futures Workshop CD
- **Interpretive Assistance Document for Polymers**
 - Available on the EPA website:
<http://www.epa.gov/oppt/sf/meetings/train.htm#materials>