

Investigating the effects of atmospheric aging on the direct radiative properties and climate impacts of black- and brown- carbon aerosol



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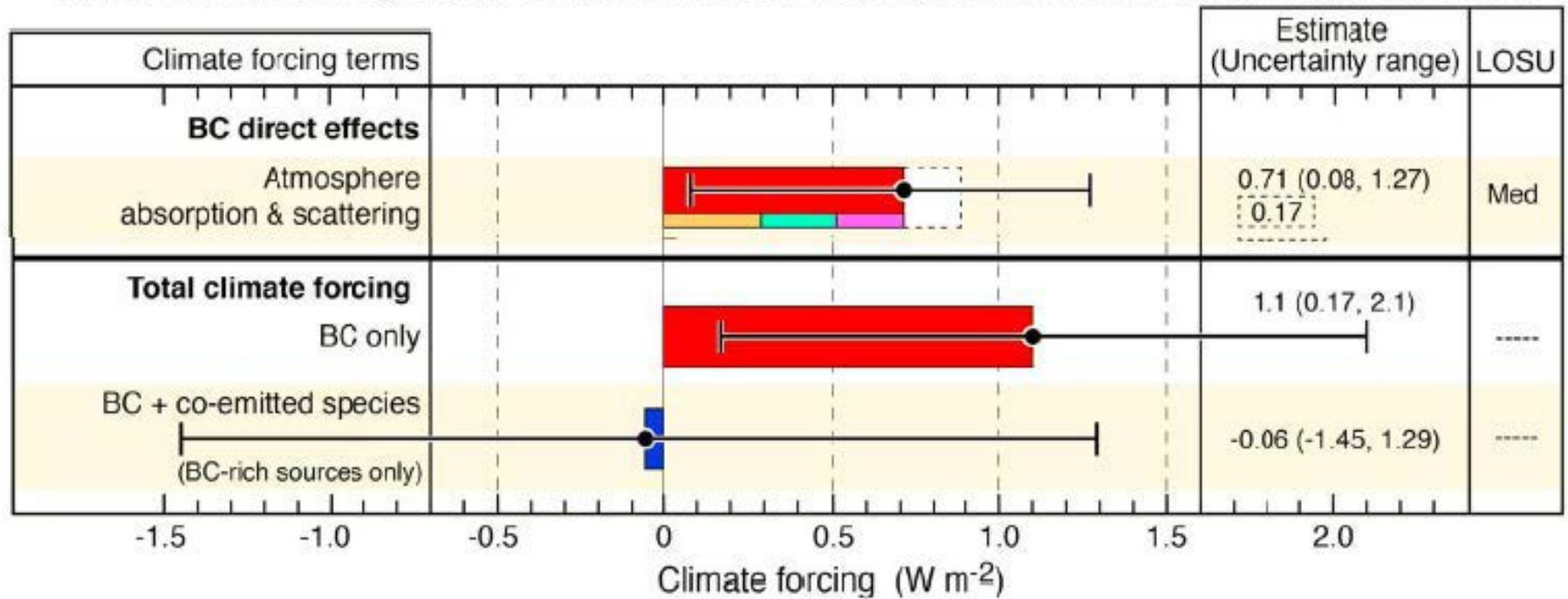
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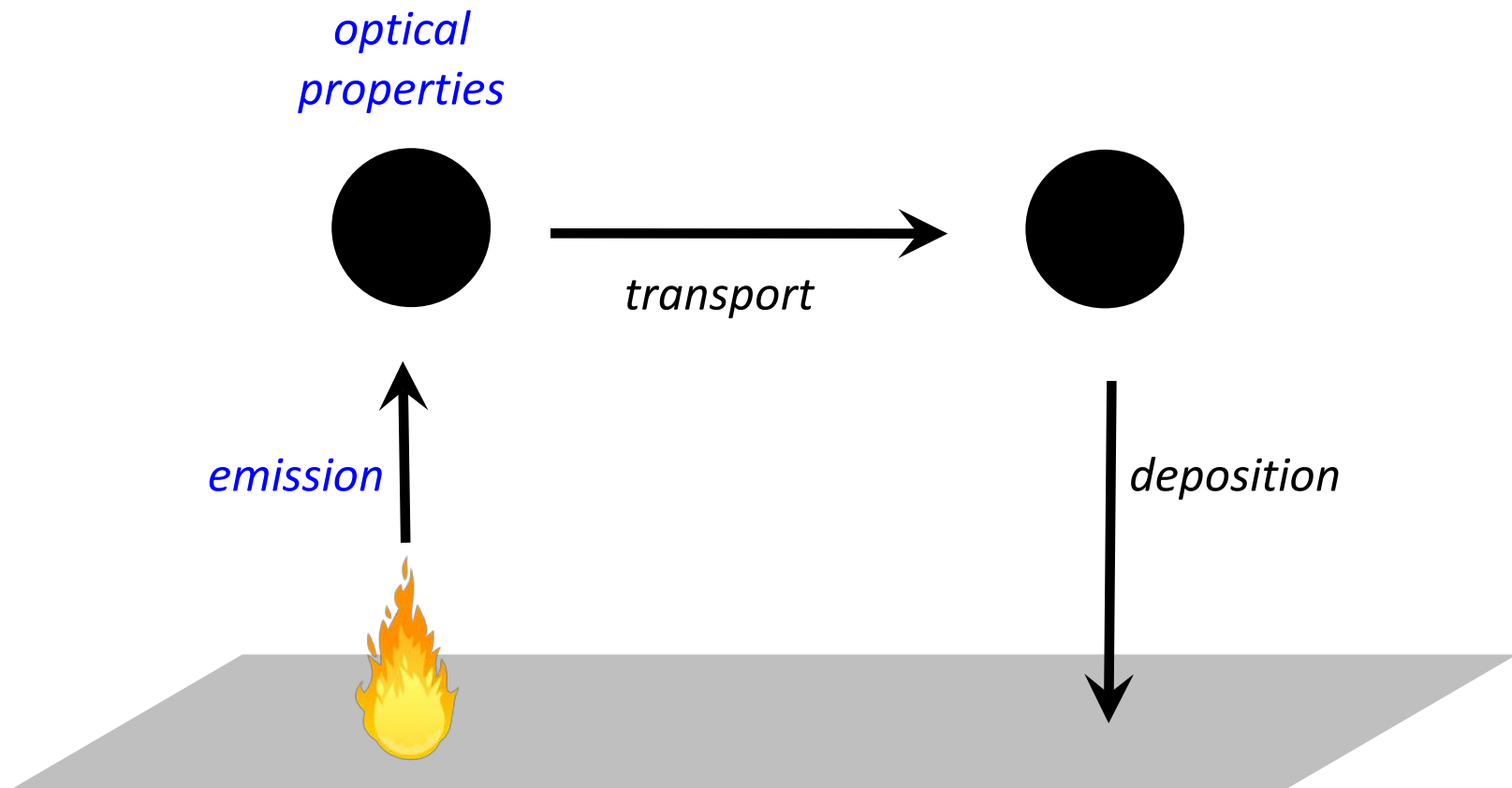
9 December 2016

BC climate forcing: Large, complex, uncertain

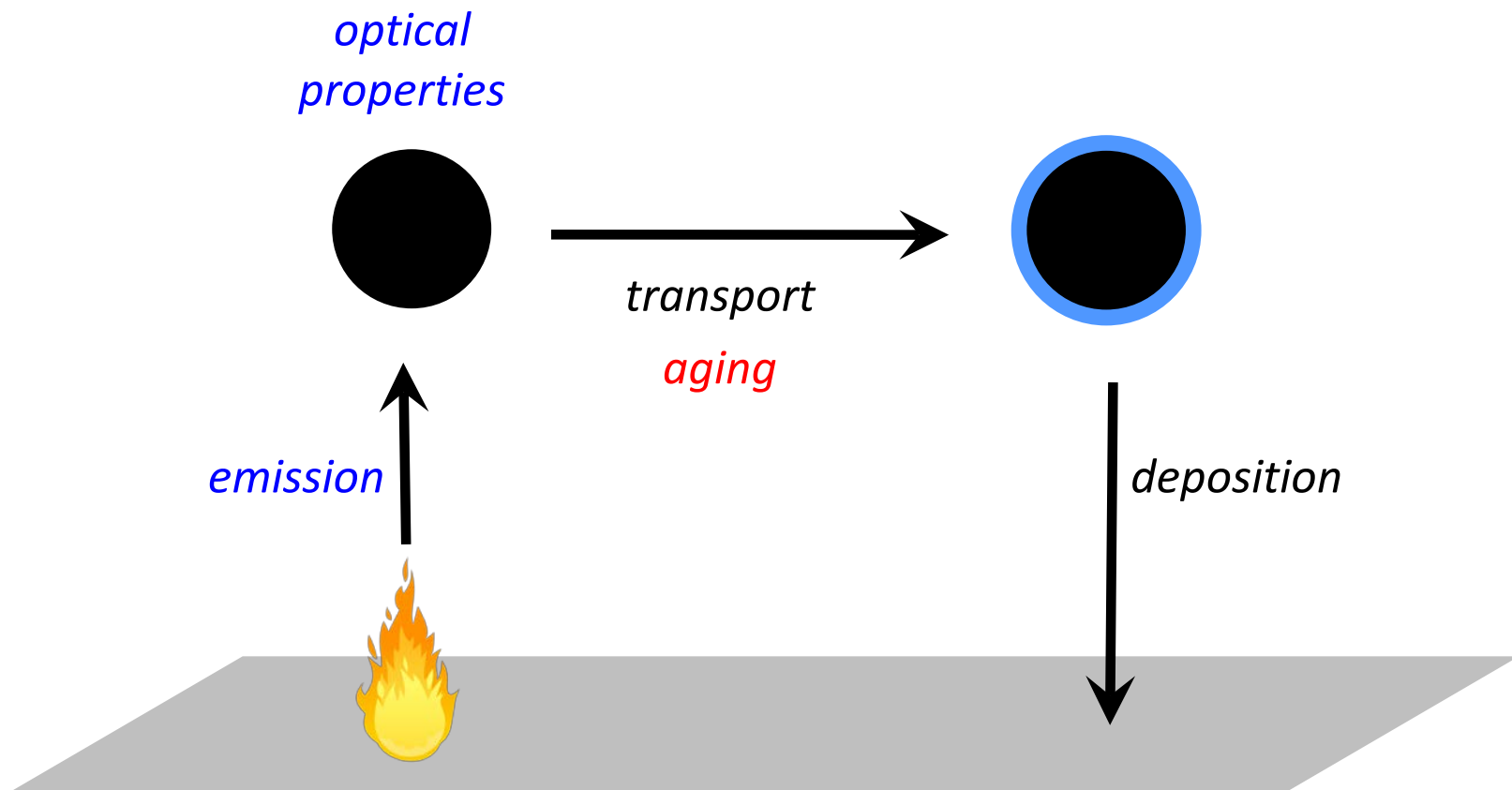
Global climate forcing of black carbon and co-emitted species in the industrial era (1750 - 2005)



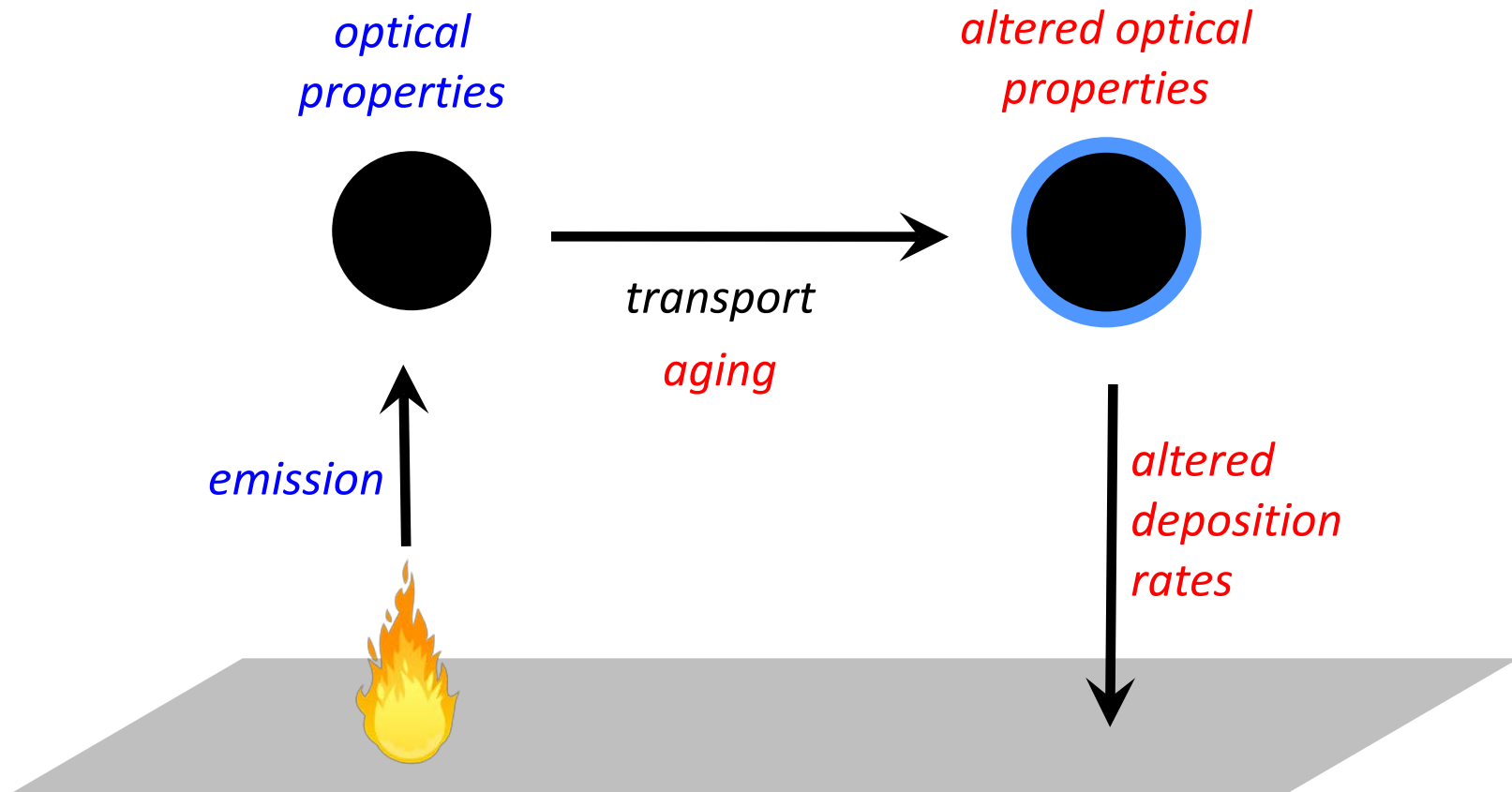
Simple lifecycle of atmospheric BC



Simple lifecycle of atmospheric BC

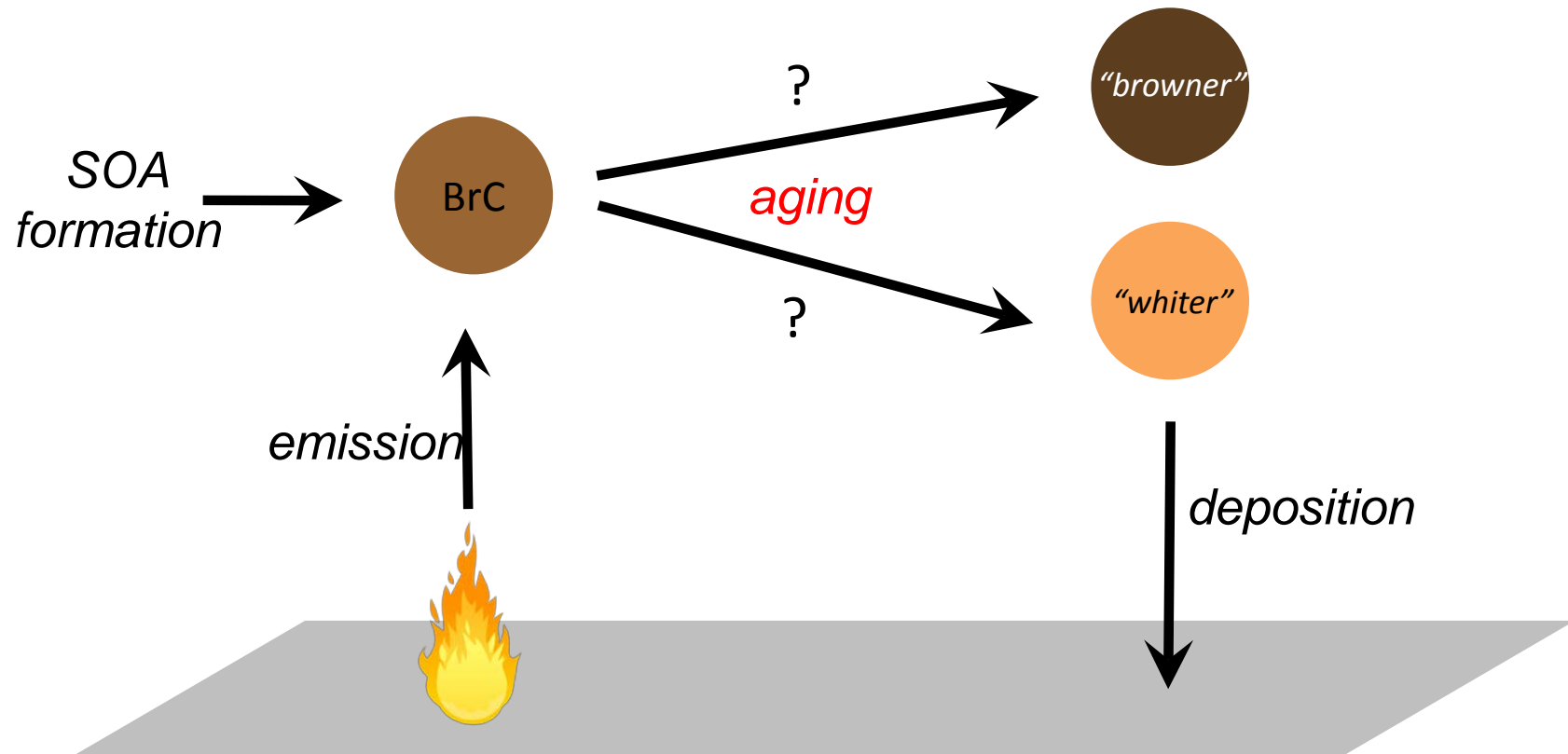


Simple lifecycle of atmospheric BC



Simple lifecycle of atmospheric BrC

“**Brown** carbon”: OC that absorbs in the UV/visible (distinct molecules)



This project

Black/brown carbon aerosol is chemically dynamic, subject to atmospheric aging reactions; these can lead to dramatic changes in physicochemical properties, and therefore climate forcing effects.

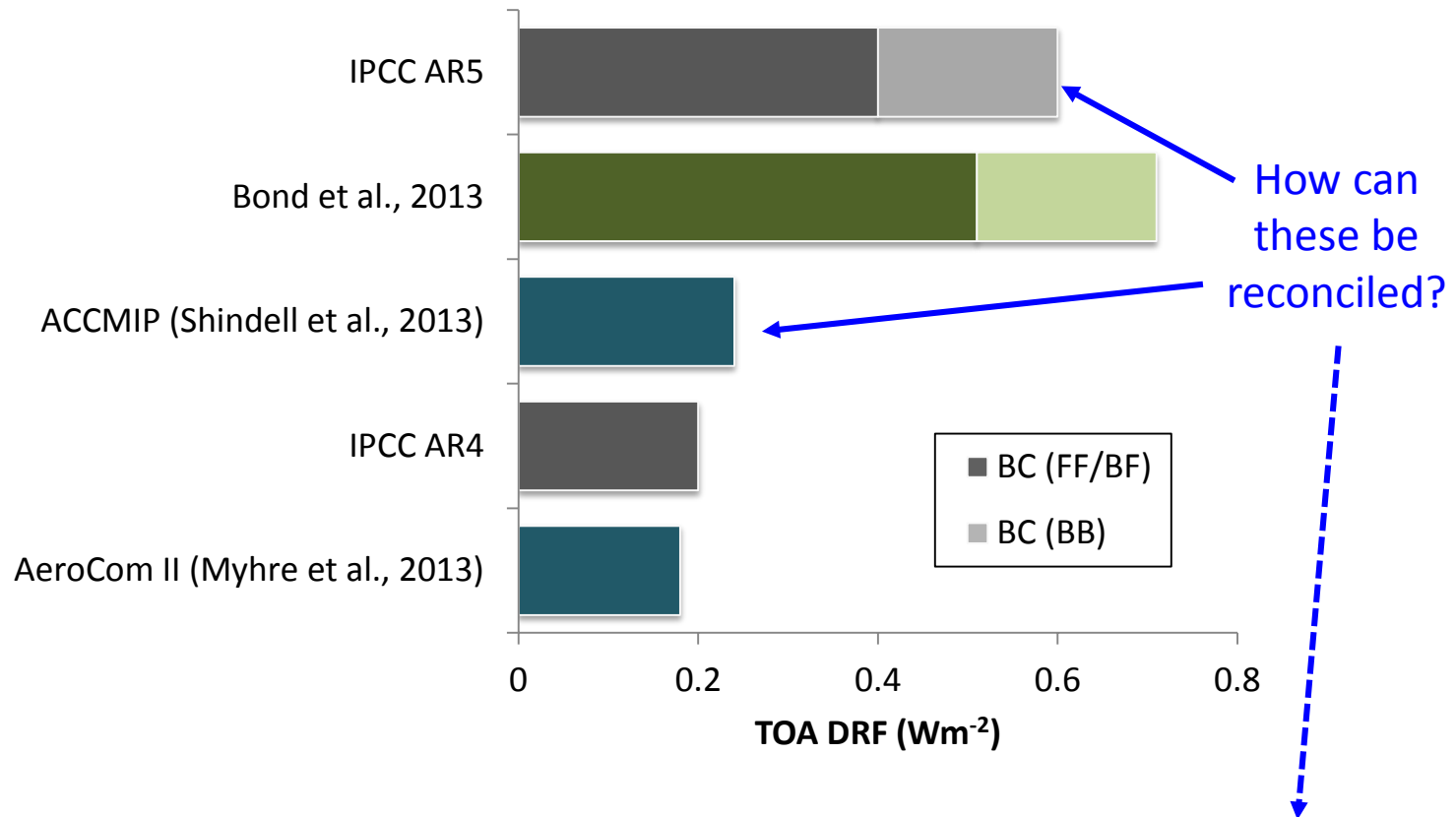
A complete understanding of this aging, and representation of this aging within models, is necessary for the accurate simulation of global direct radiative forcing.

Major questions:

- what are the most important atmospheric aging transformations of BC/BrC?
- what effects does aging have on climate-relevant properties of BC/BrC?
- how do these aging reactions impact BC/BrC direct radiative forcing?

IPCC AR5 Estimates that Black Carbon is the 2nd Largest Warming Agent in the Atmosphere.

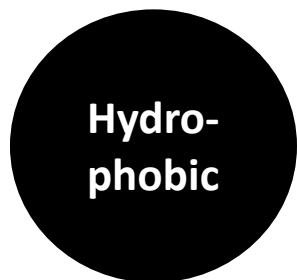
(but that's not what models say)



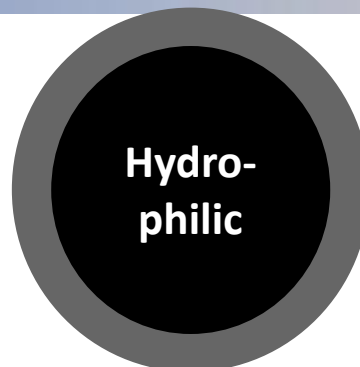
At the same time, AeroCom models overestimate BC over Americas by factor ~8 [Koch et al., 2009], overestimate remote HIPPO BC by factor ~5 [Schwarz et al., 2010].

New Model Aging Processes for BC

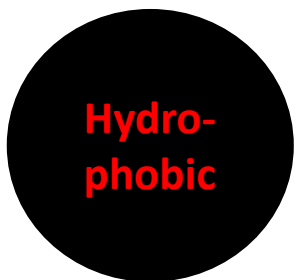
Old Assumptions



1.15 days



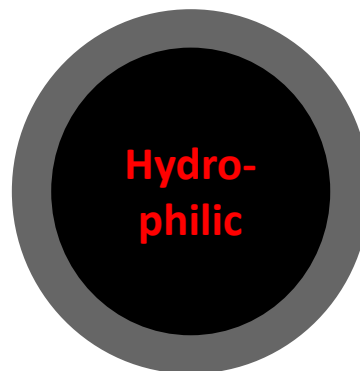
Anthropogenic



Sulfate, etc.



$$k = 1/\tau = a [\text{SO}_2] [\text{OH}] + b$$



New Assumptions



Organic components



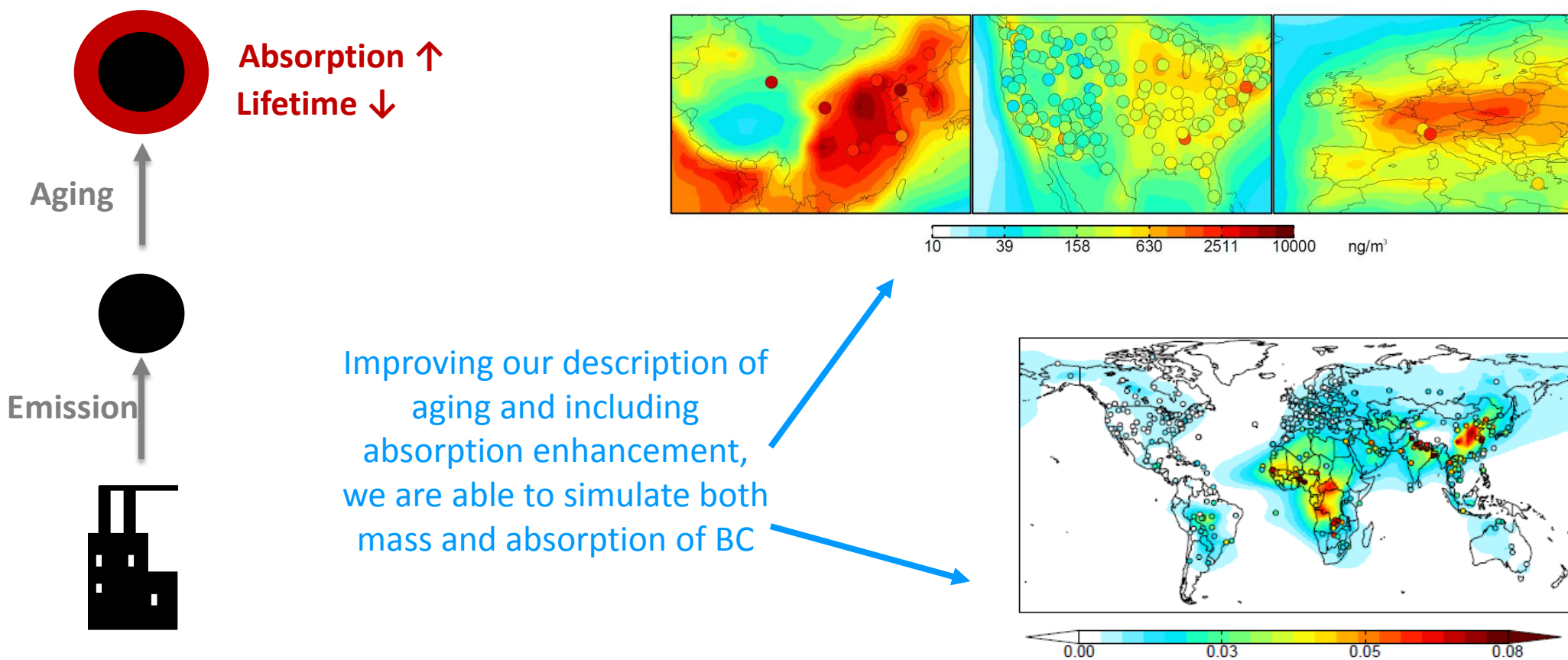
4 hours

(also increase fraction emitted as hydrophilic to 70%)



Biomass burning

Aging Processes Reconcile Both Mass and Absorption Constraints

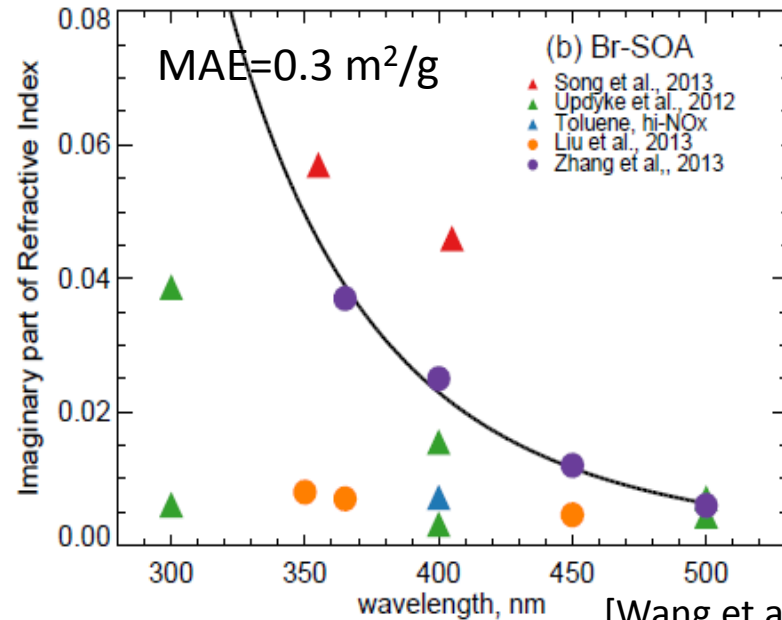
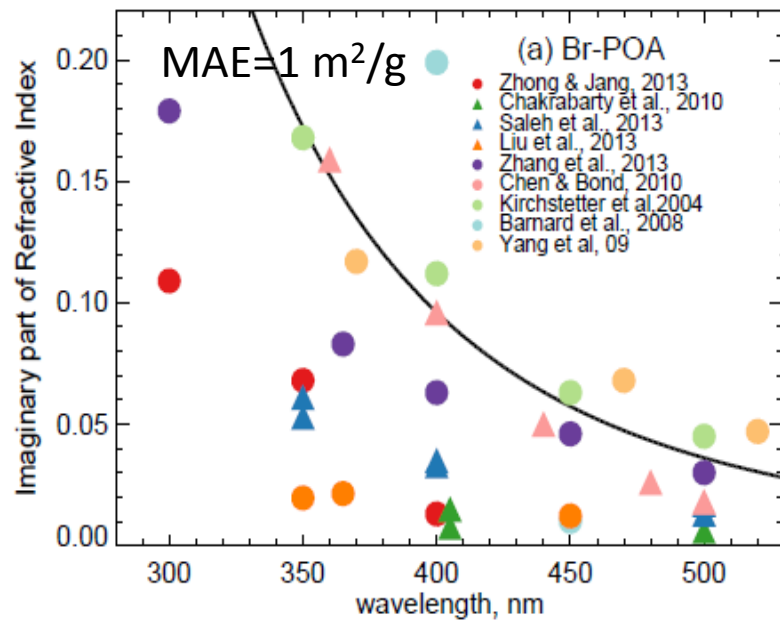


We estimate that the BC DRF (0.21 W/m^2) is much less than AR5 estimate (0.6 W/m^2). Suggests that controlling BC is less effective for climate mitigation.

How Important is Brown Carbon?



Modelling studies estimate that BrC contributes 20% to 40% of total carbonaceous aerosol absorption and that its direct radiative effect (DRE) ranges from +0.1 to + 0.6 Wm^{-2} (Feng et al., 2013; Lin et al., 2014; Wang et al., 2014; Saleh et al., 2015; Jo et al., 2016)

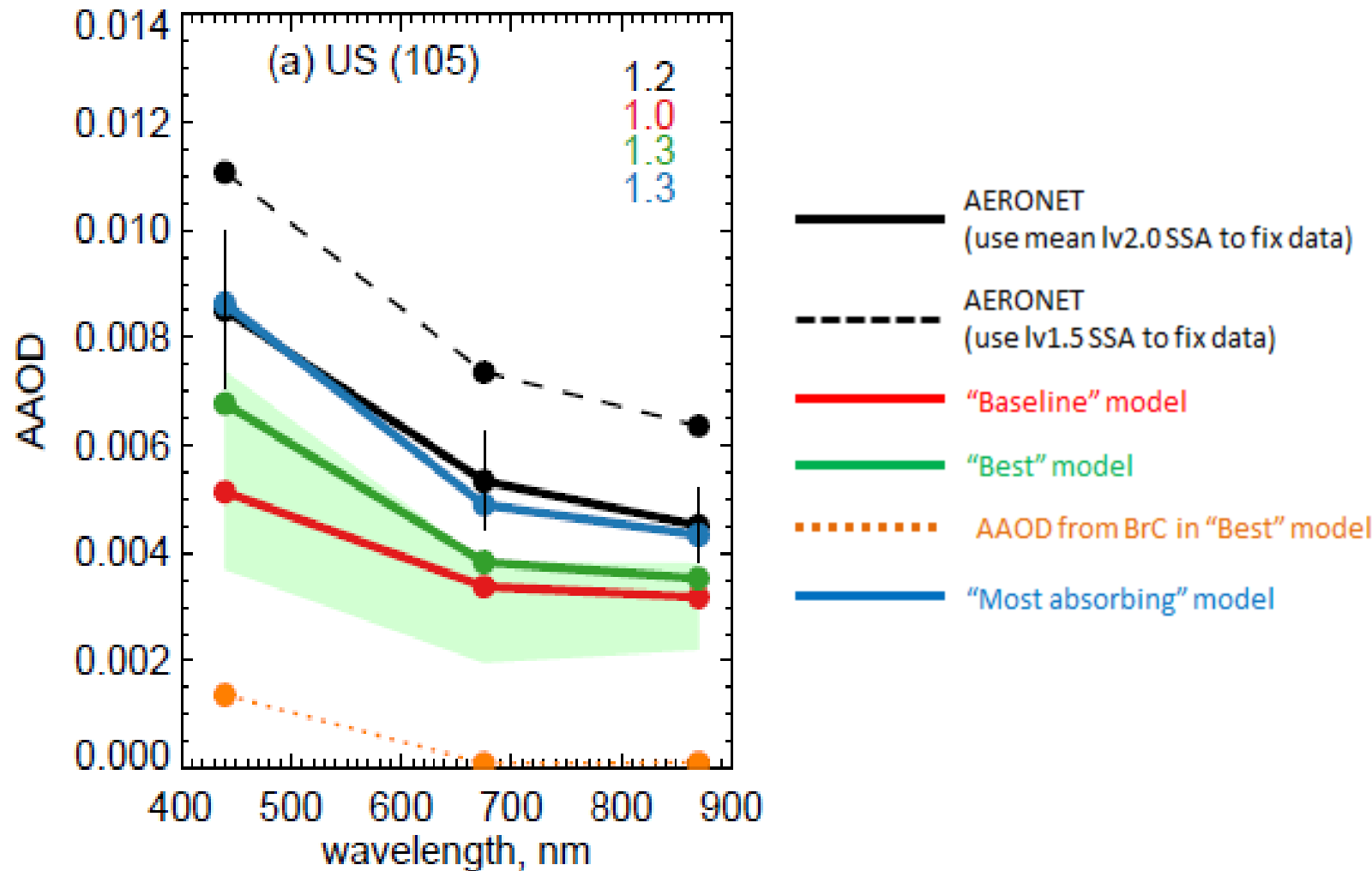
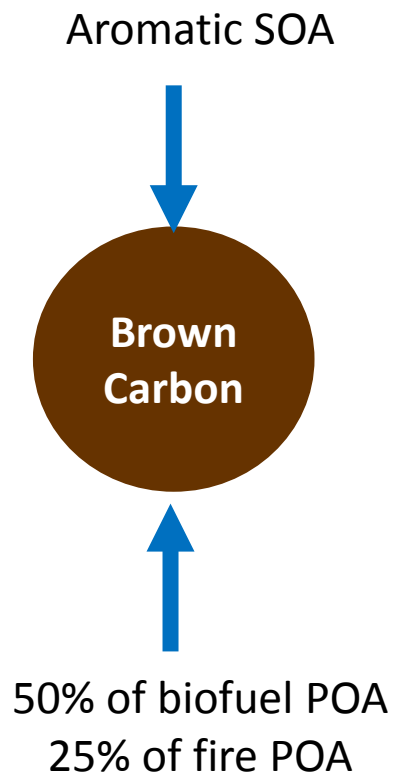


[Wang et al., ACP, 2014]

But sources and properties are so uncertain, we need observational constraints.

Including Brown Carbon is Critical to Capturing the Spectral Dependence of AERONET AOD

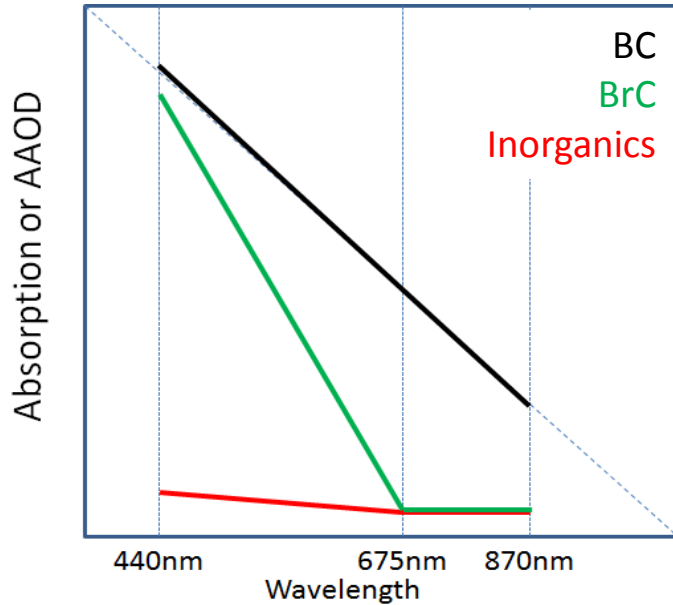
*AAOD product here using lev2 SSA with lev1.5 AOD



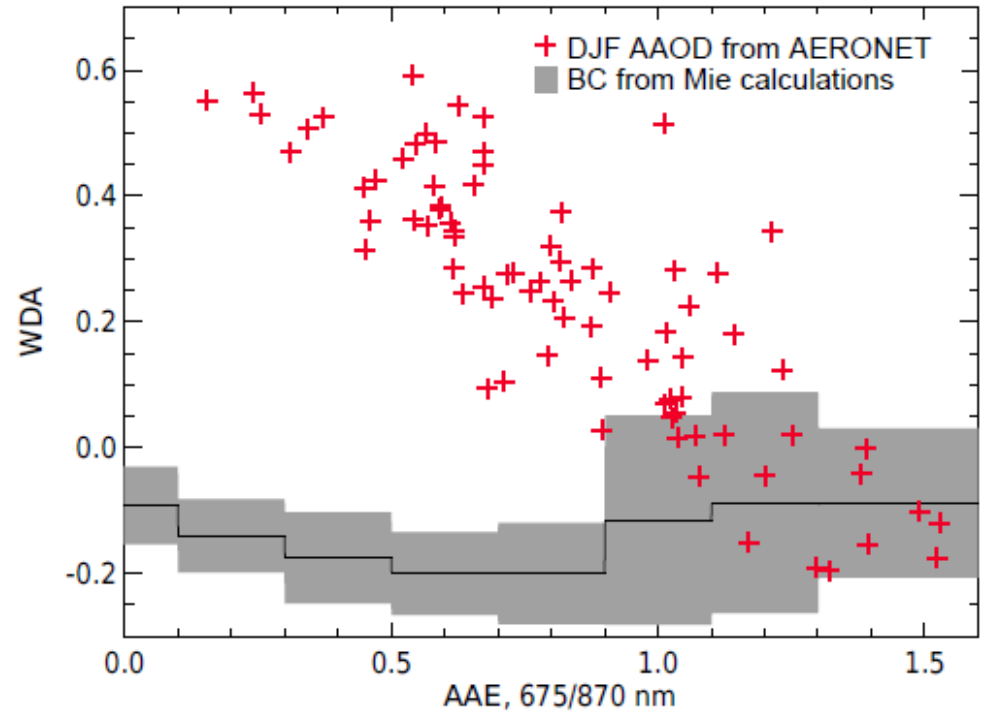
But given uncertainties in BC absorption, this constitutes a qualitative constraint at best

How to Separate BrC from Absorption Measurements?

Schematic of wavelength dependence of absorption



$$AAE = -\frac{\ln\left(\frac{abs(\lambda_1)}{abs(\lambda_2)}\right)}{\ln\left(\frac{\lambda_1}{\lambda_2}\right)}$$

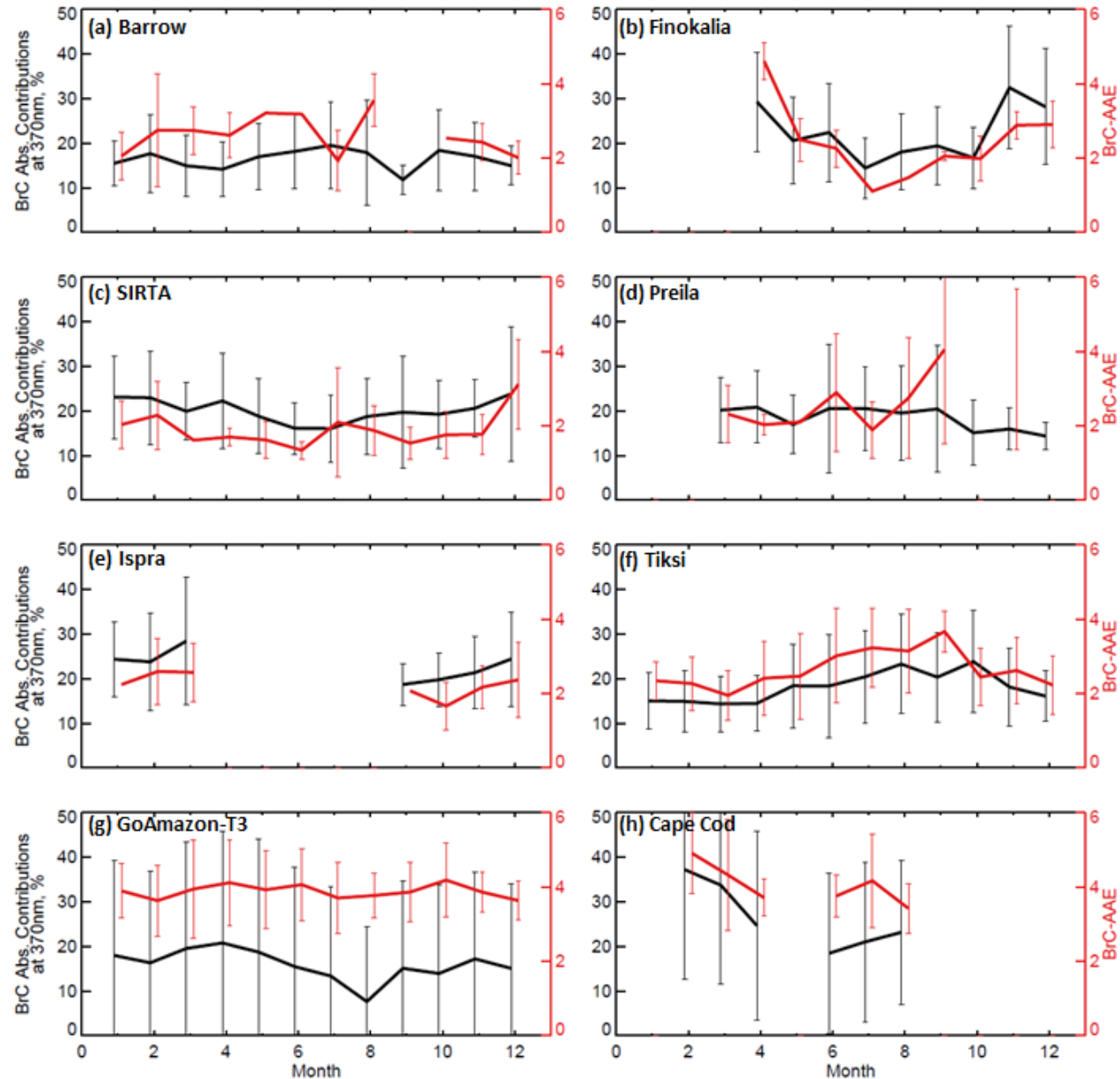
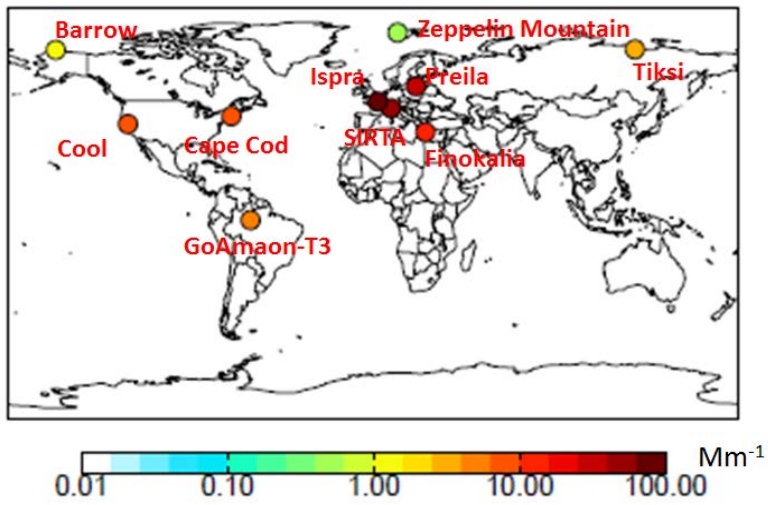


$$WDA = AAE_{440/870} - AAE_{675/870}$$

Many studies assume AAE of BC=1 and estimate BrC, but this is a poor assumption. Instead we estimate the wavelength dependence of BC absorption, and use this to estimate BrC absorption.

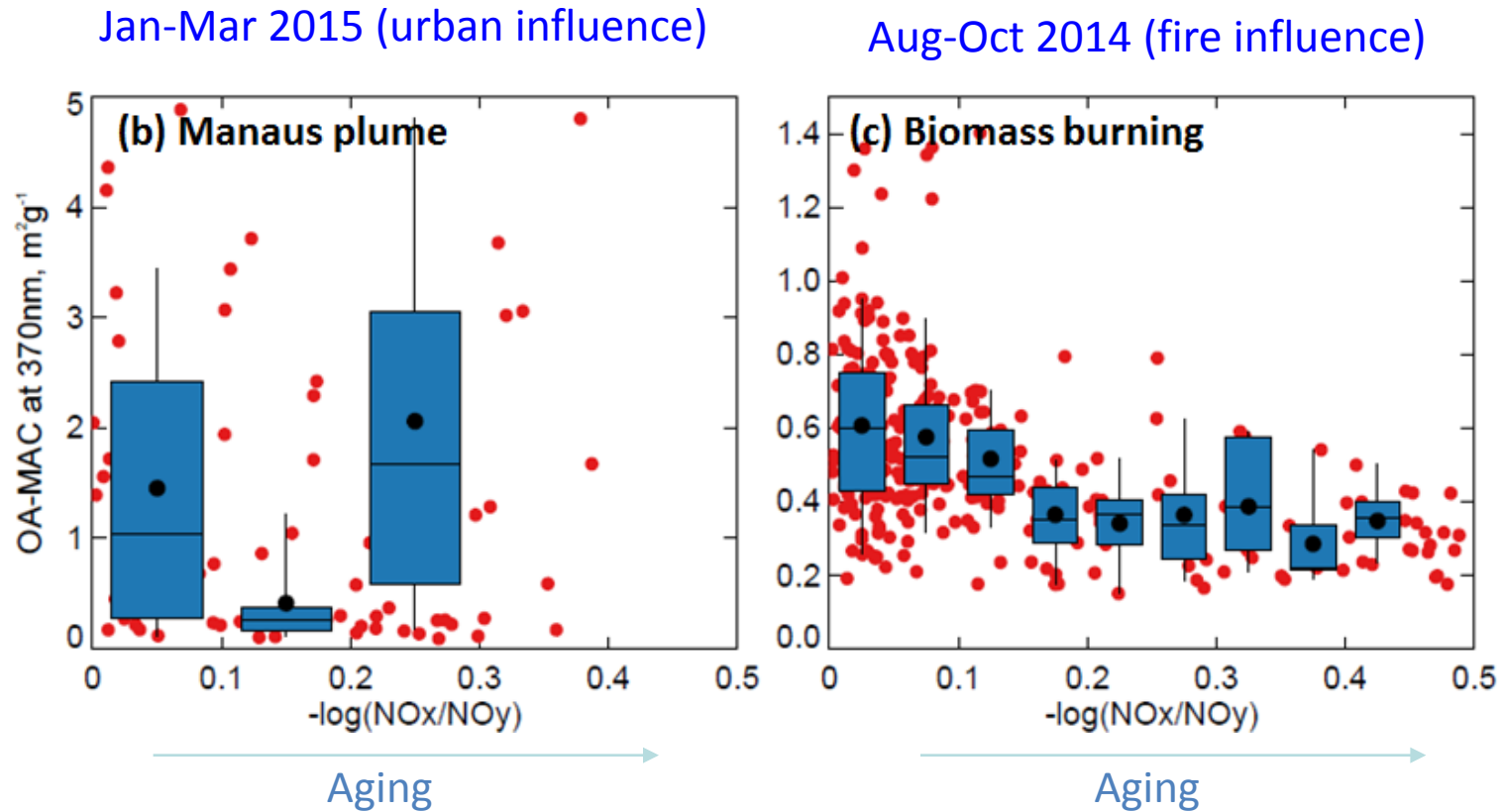
Globally BrC makes up 10-35% of Absorption

(370 nm)



Does BrC Photochemically “Whiten” in the Atmosphere?

GoAmazon Site in Manacapuru



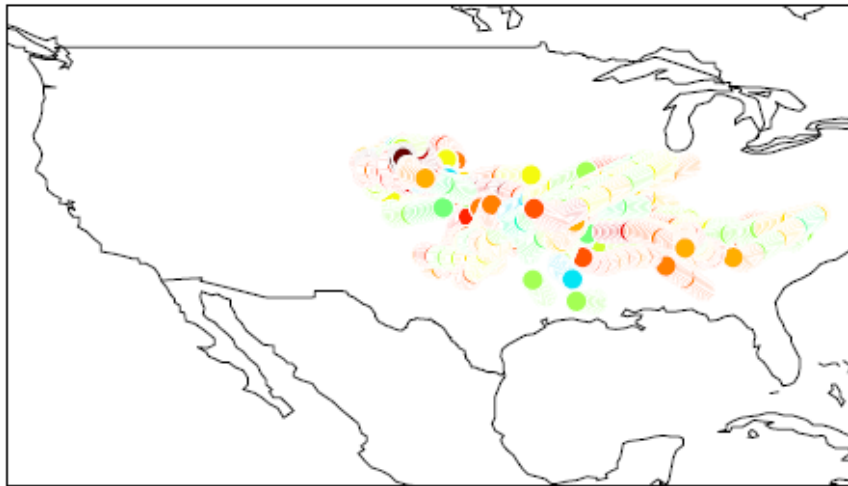
Suggests that urban/fossil BrC absorption is not photochemically altered (we conclude the same for Cape Cod), but that BrC from fires whitens with a lifetime of ~ 1 day (consistent with Forrister et al., 2015). Possible means of bringing models into better agreement with observational constraints?

On-Going Work: Exploring the Radiative Impacts of BrC

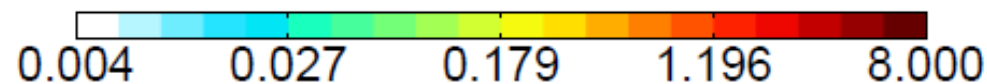
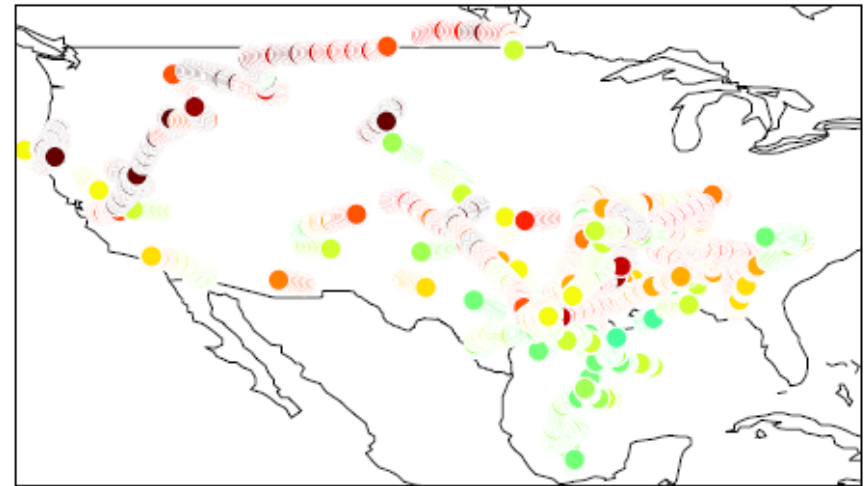
Building a BrC simulation for GEOS-Chem

- Optical properties as a function of BC:OA of source
- Impact of including whitening
- Testing against observations of OA mass and OA absorption
- Estimating the DRE with range of assumptions

DC3



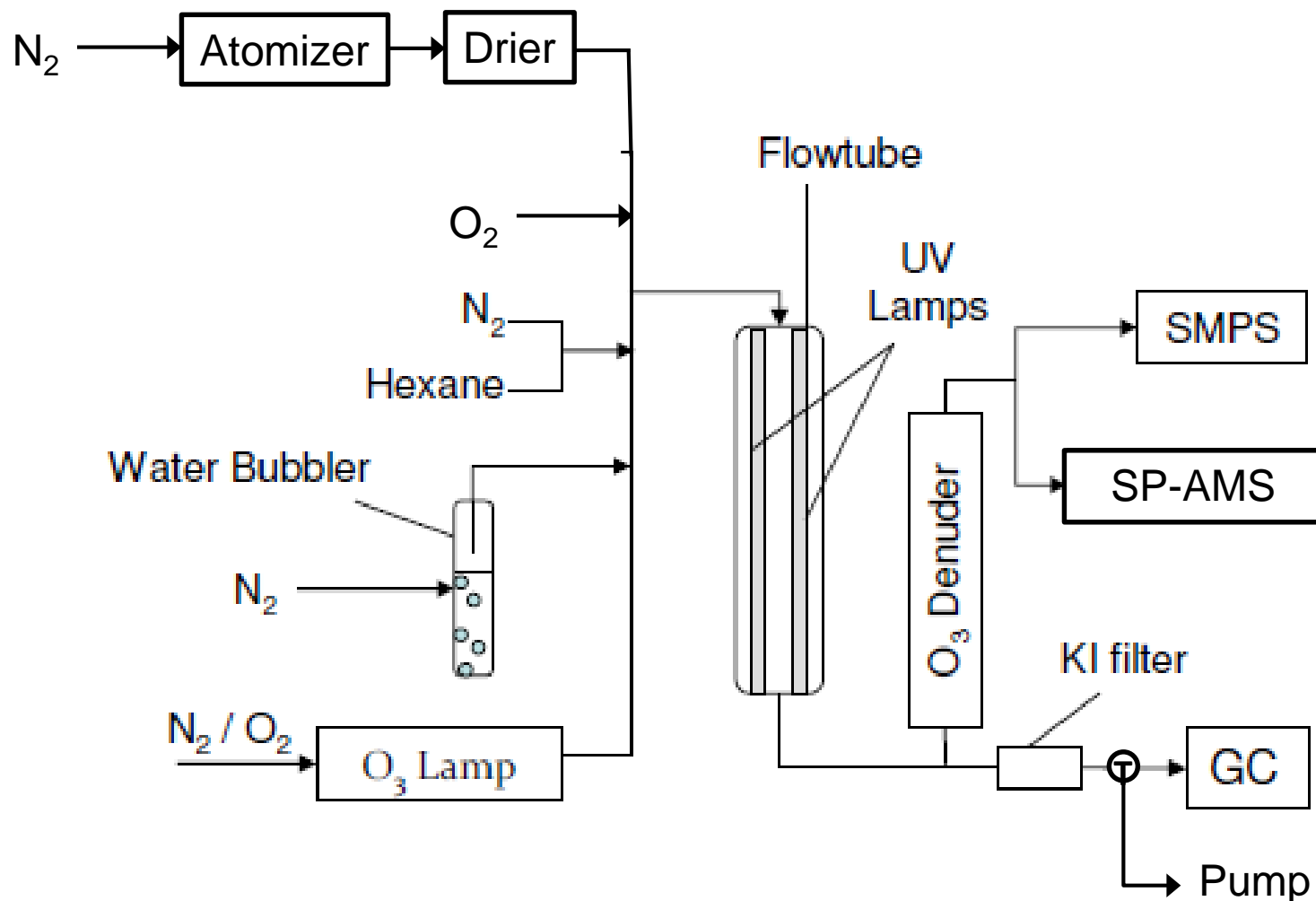
SEAC⁴RS



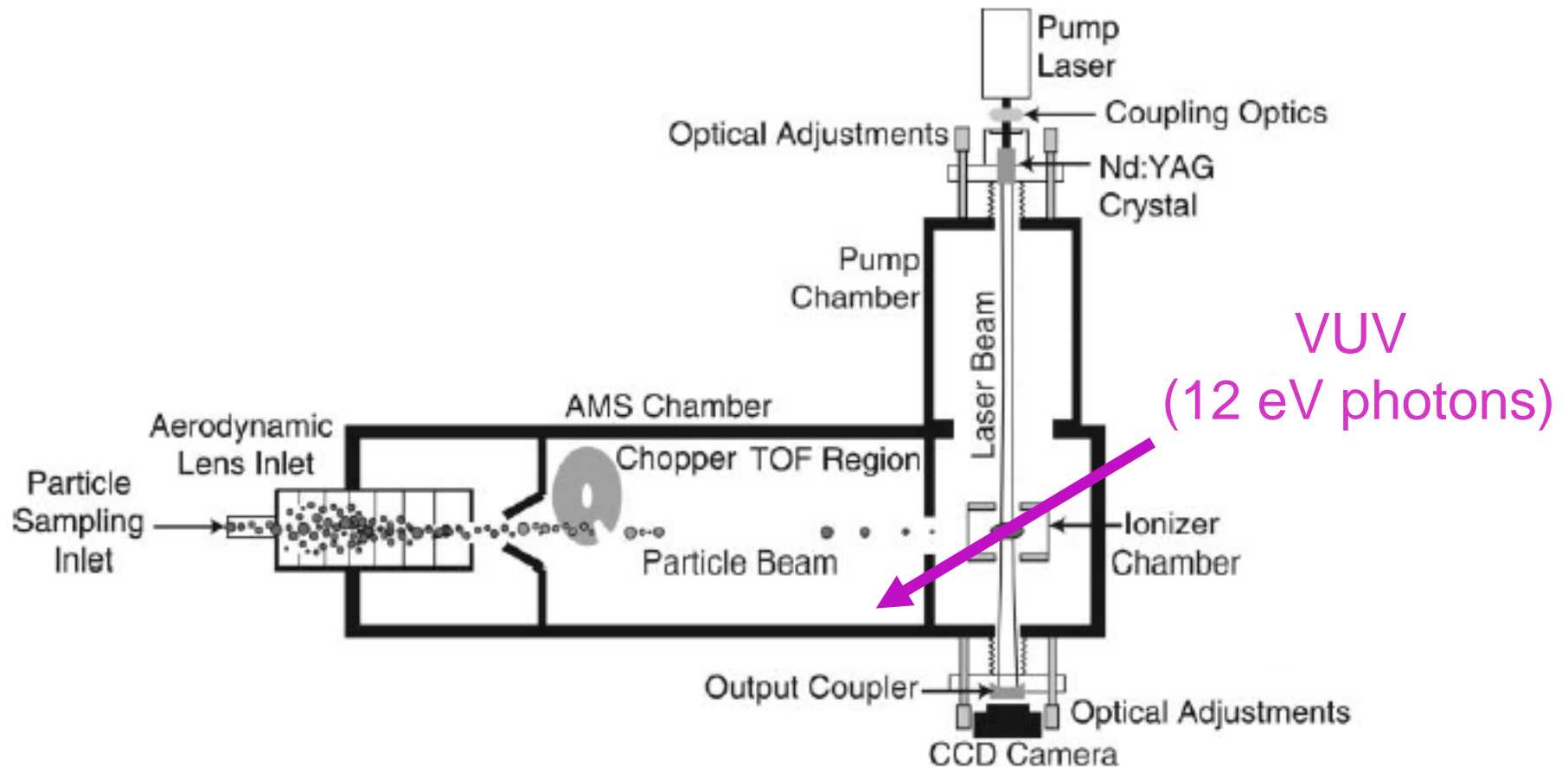
Absorption from Organic Aerosols at 365nm, Mm^{-1}

Heterogeneous oxidation of BC

with Kevin Wilson (LBNL), Manjula Canagartna and Paola Massoli (ARI)



SP-AMS with VUV ionization



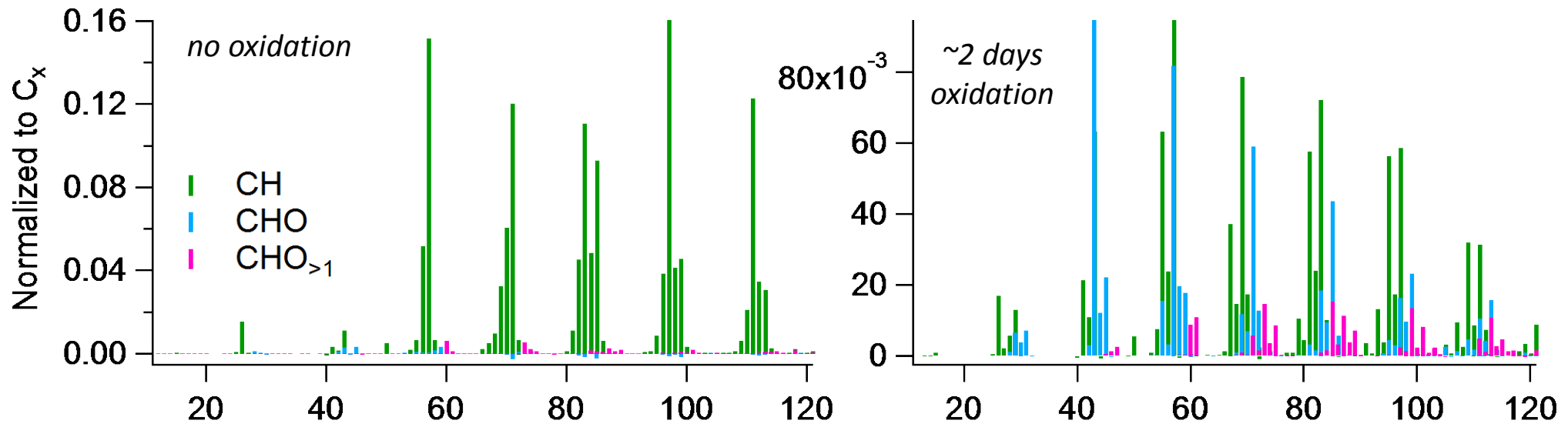
Ionization efficiencies

PAHs: 7-8 eV

Most organics: 9-10 eV

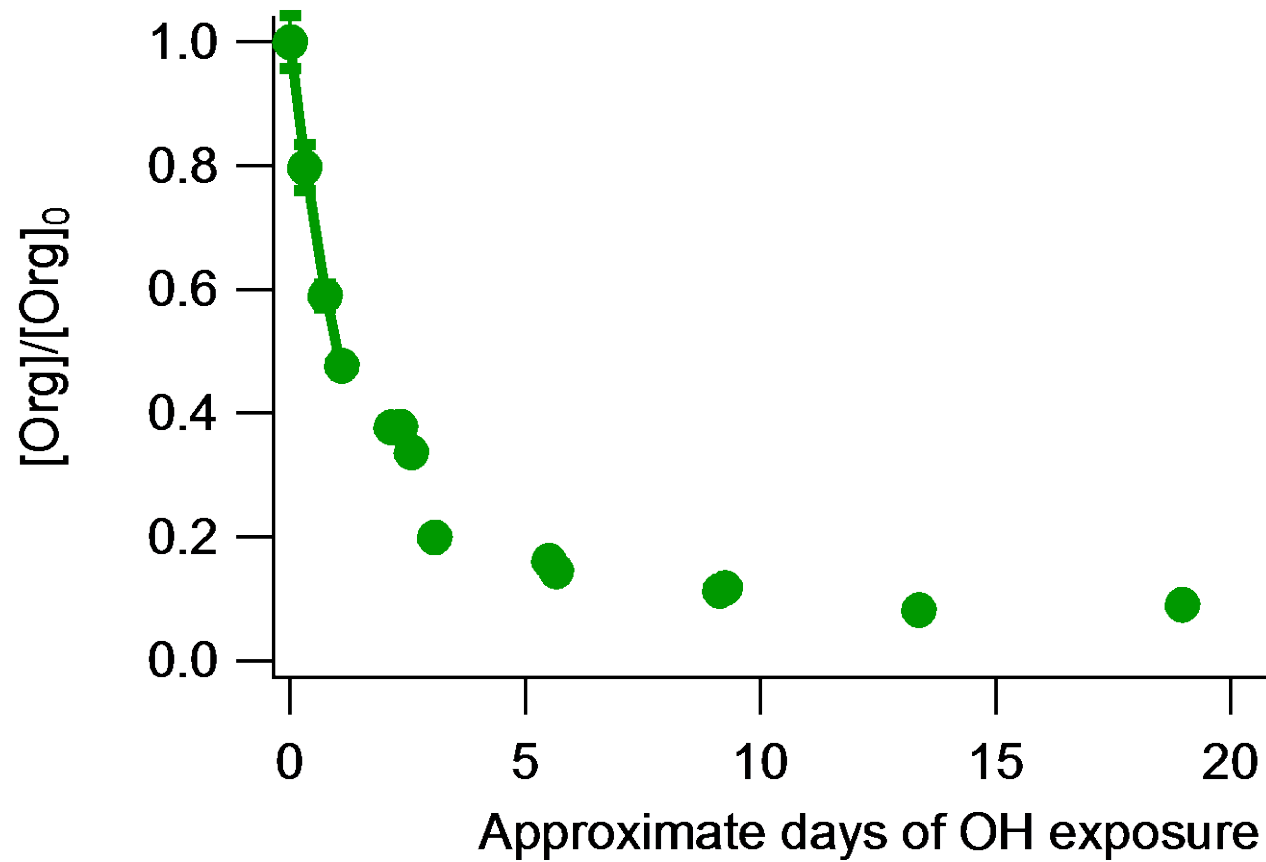
BC fragments (e.g. C_3): ~ 11.5 eV

Changes to adsorbed organic species



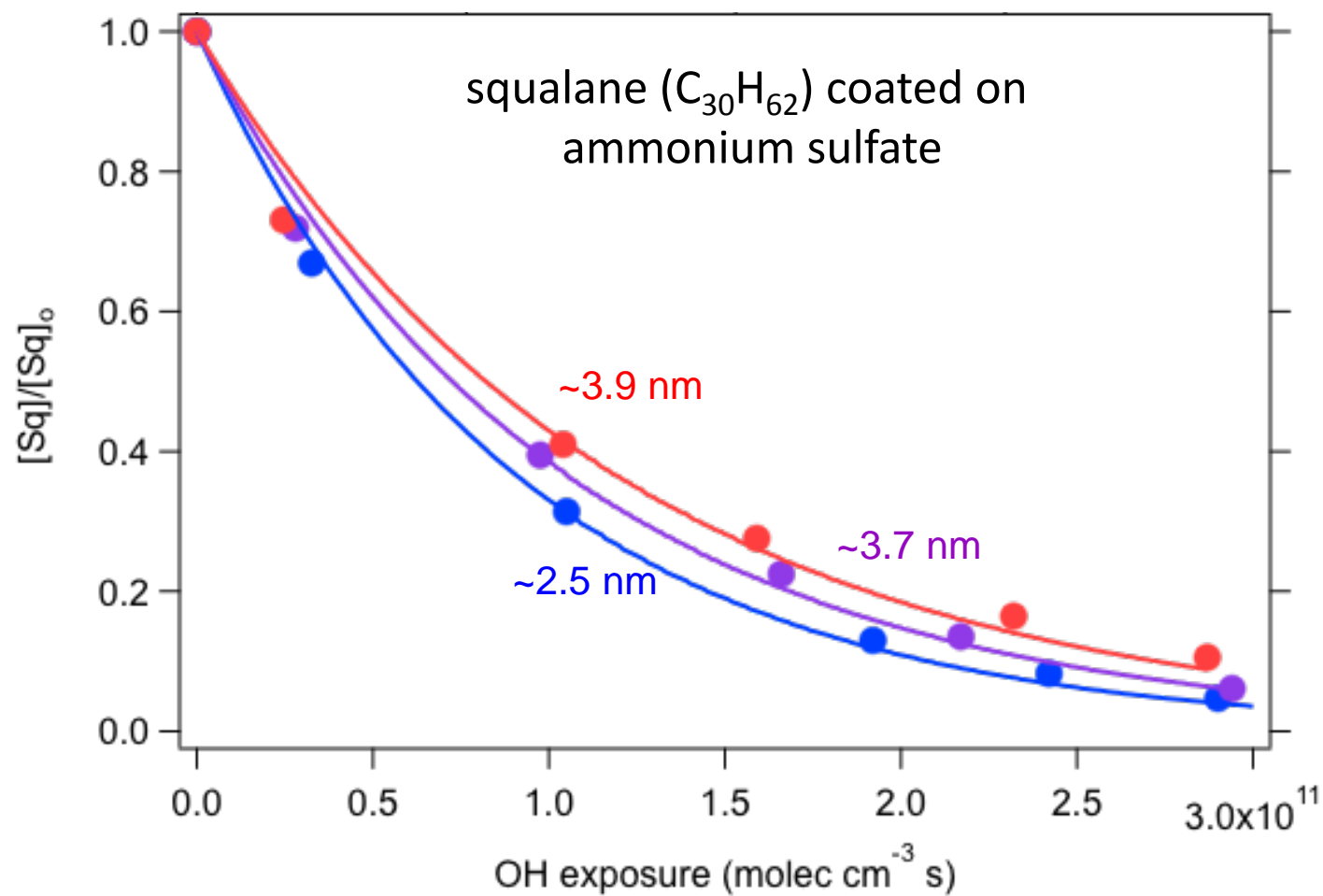
Organics associated with BC (representing small fraction of total particles) undergo dramatic chemical changes

Rapid loss of organic species



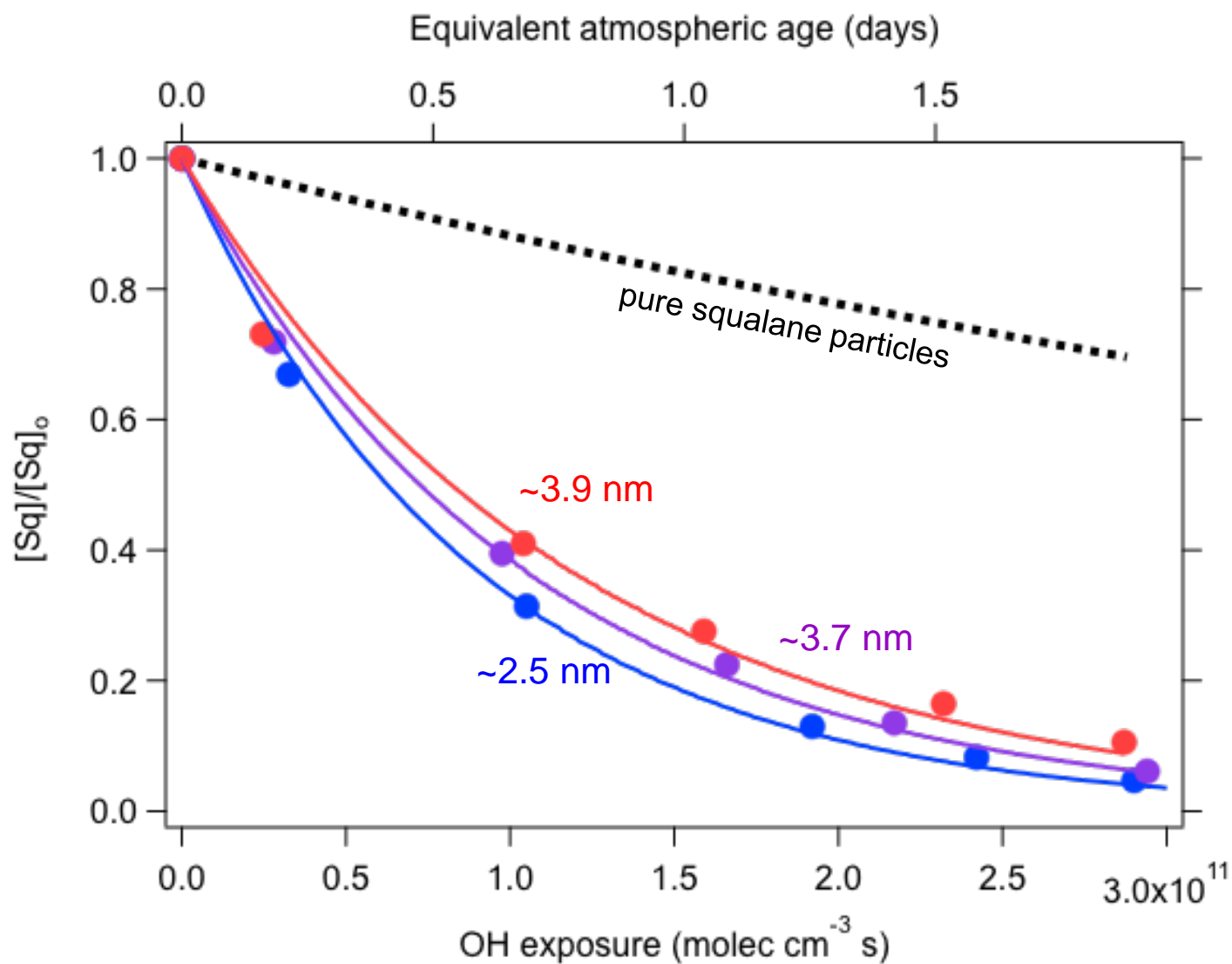
Lifetime of adsorbed organics is short

Heterogeneous oxidation of thin coatings



Thinner coating \rightarrow faster reaction

Heterogeneous oxidation of thin coatings

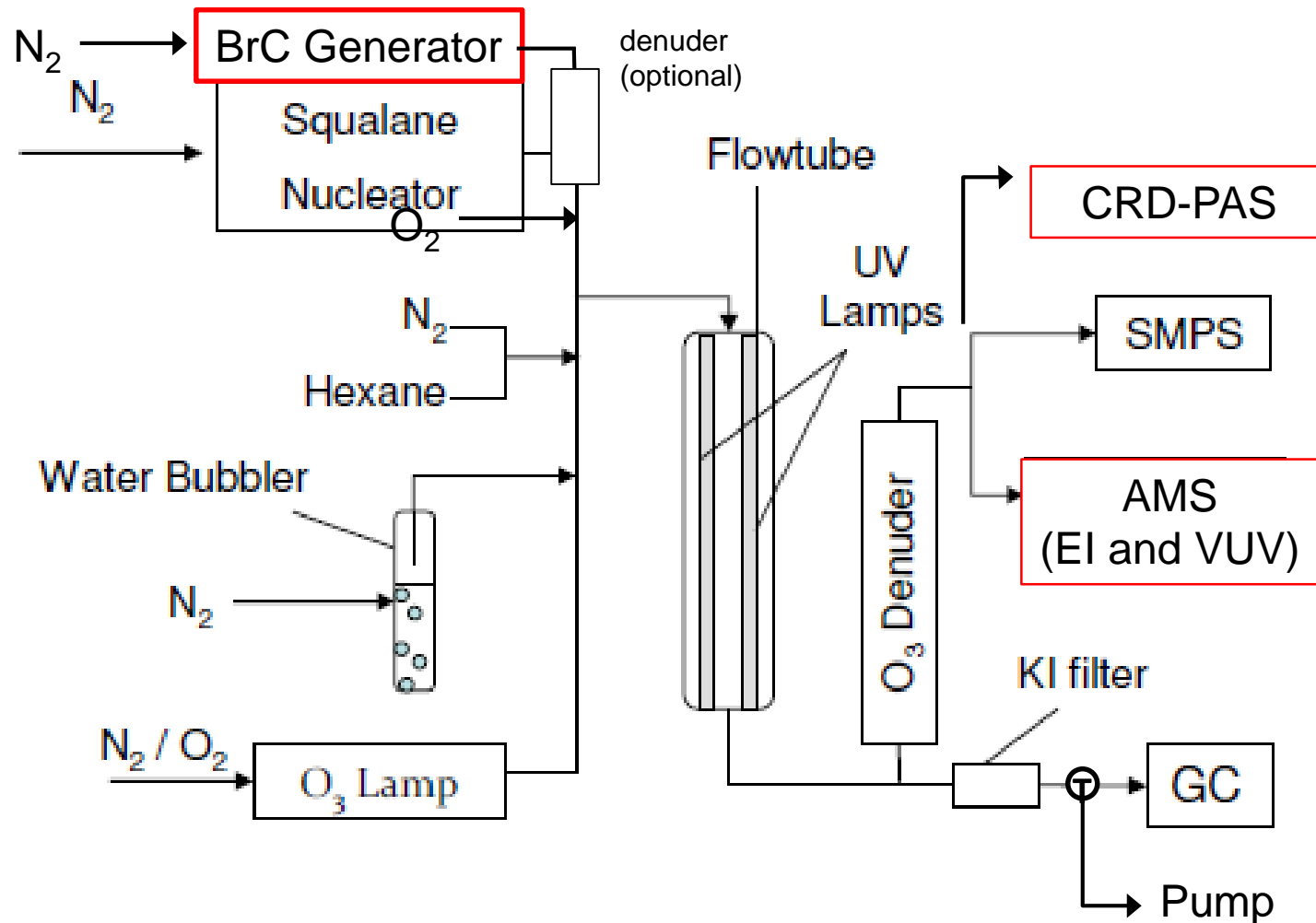


Thinner coating → faster reaction

Coated particles react far faster (x10) than pure particles

Heterogeneous oxidation of Brown Carbon

with Xiaolu Zhang, Chris Cappa (UCD); Kevin Wilson, Tom Kirchstetter (LBNL)



“Brown carbon generator”

Heated biomass: Smoldering only, no BC



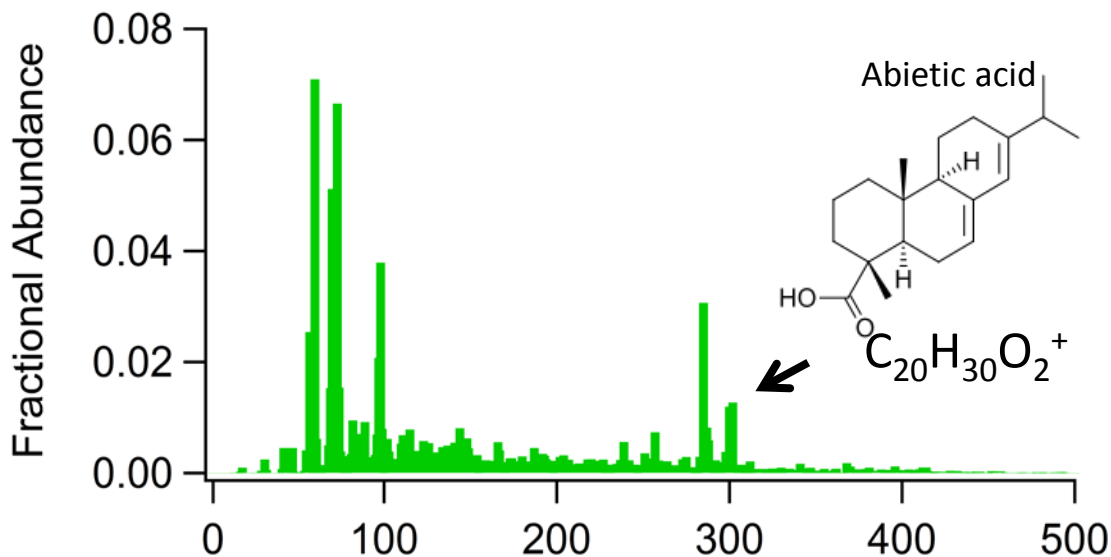
Single-scattering albedo (SSA)

@405 nm: ~ 0.99

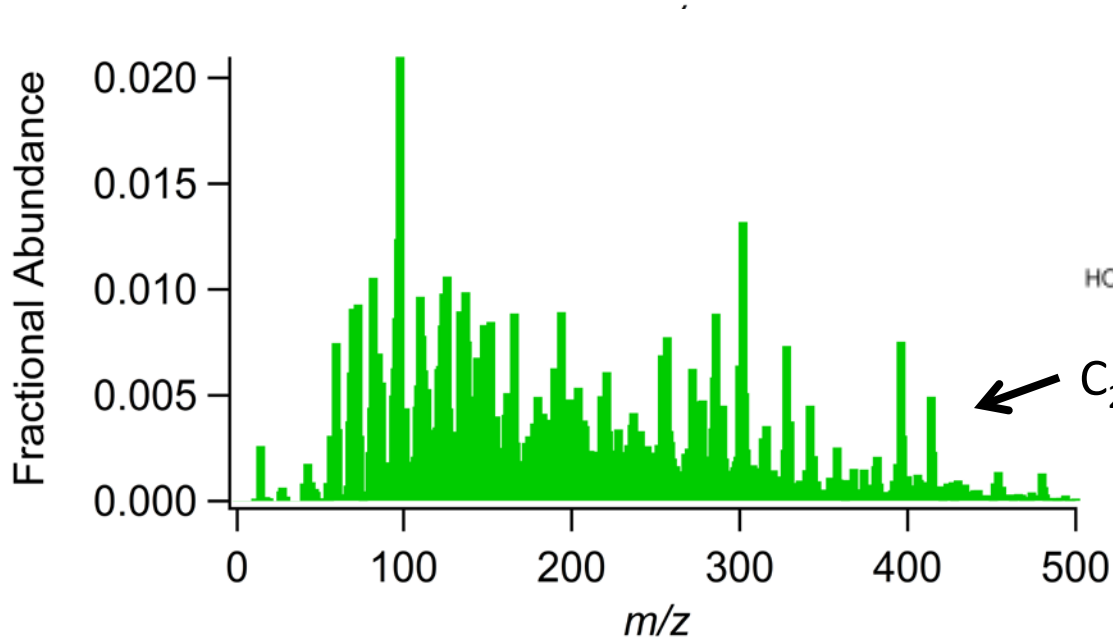
Volume-absorption coefficient (VAC)

@405 nm: $\sim 0.20 \text{ m}^3 \text{ cm}^{-3}$

Primary BrC is complex and highly variable

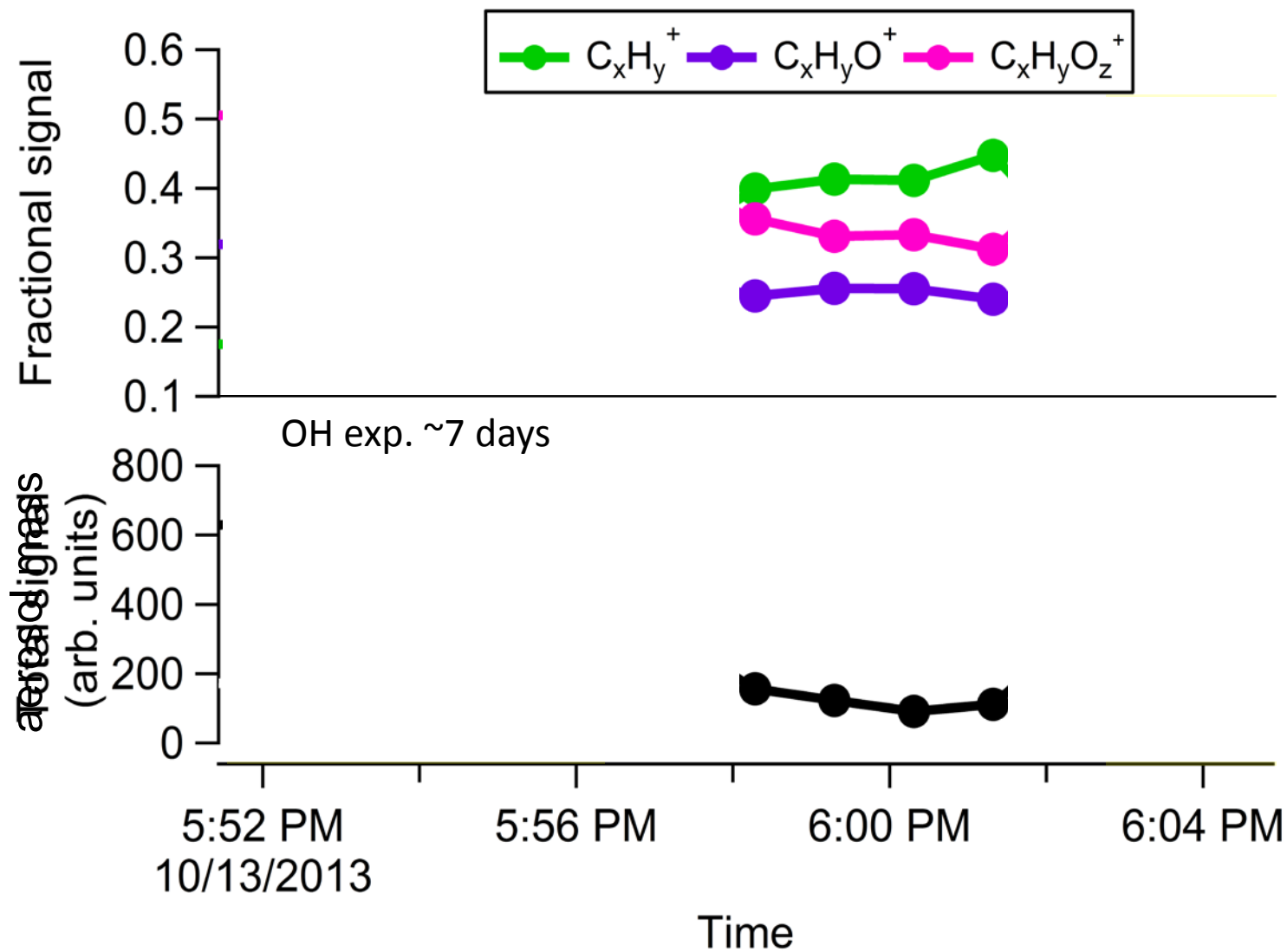


VAC = 0.05 m^2/cm^3

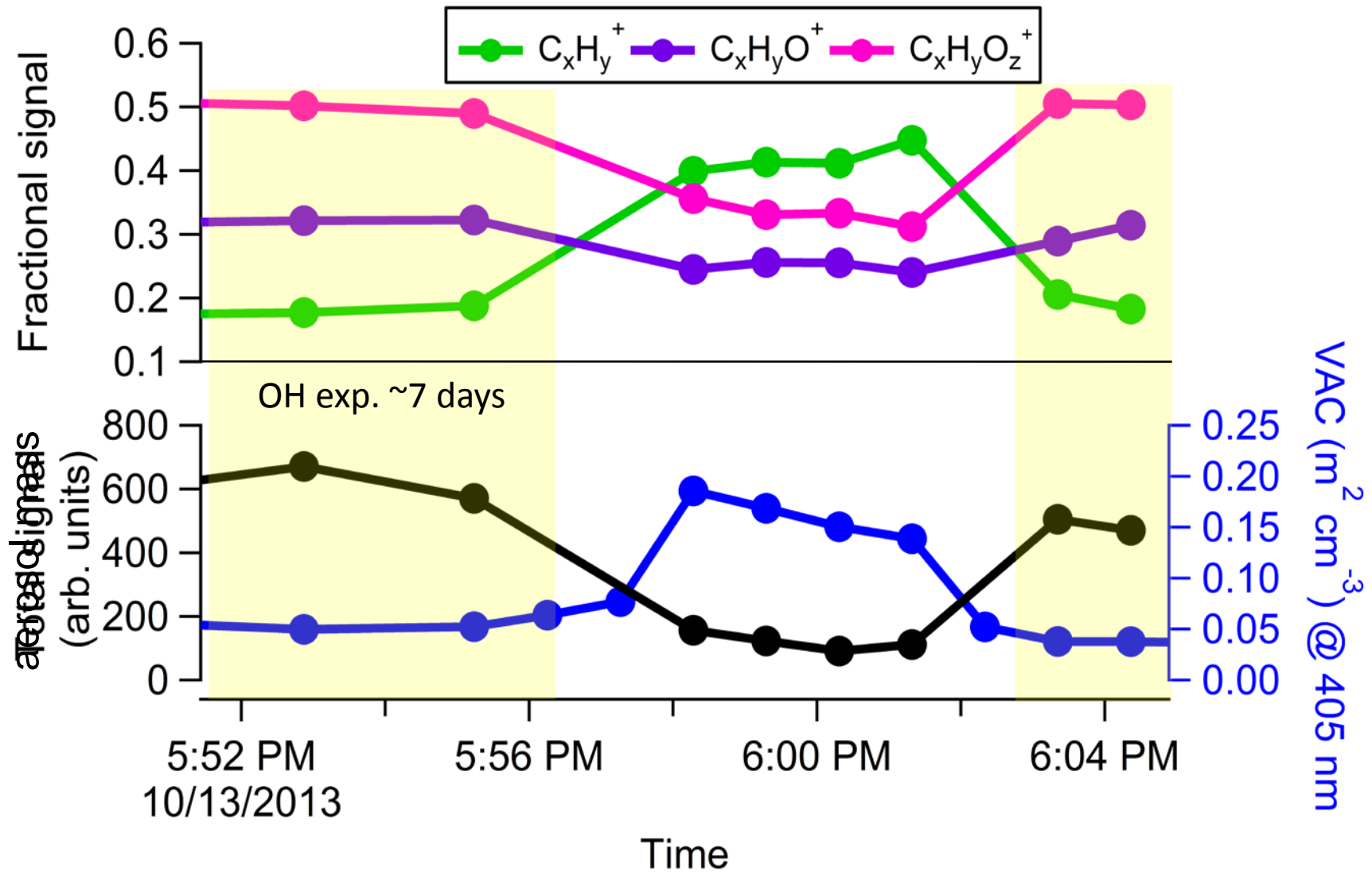


VAC = 0.30 m^2/cm^3

SOA formation from POA + vapors + OH

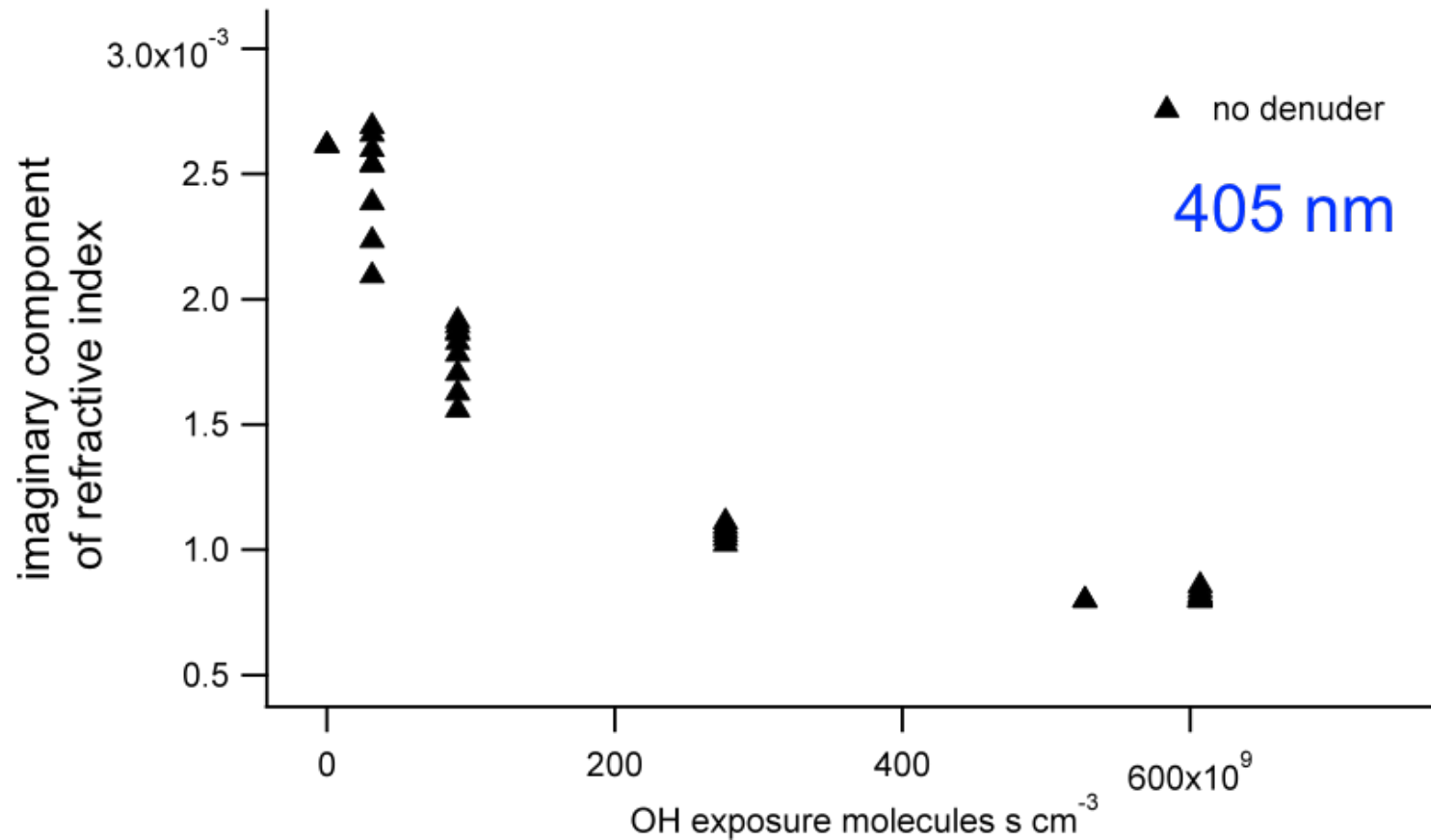


SOA formation from POA + vapors + OH



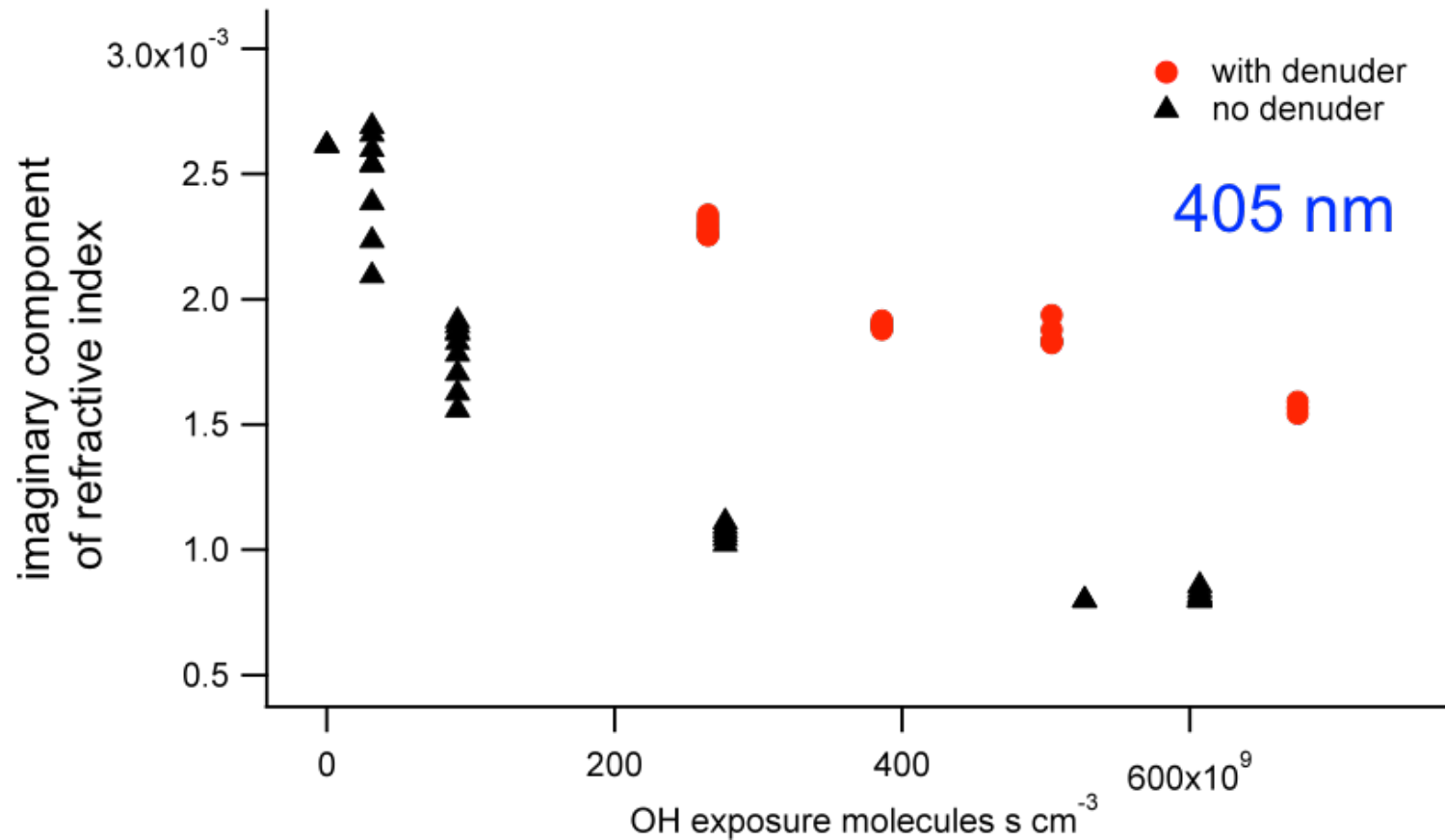
SOA is much less "brown" than POA
(also more hygroscopic)

Optical properties vs. degree of aging



SOA formation less absorption (SOA is less brown than POA)

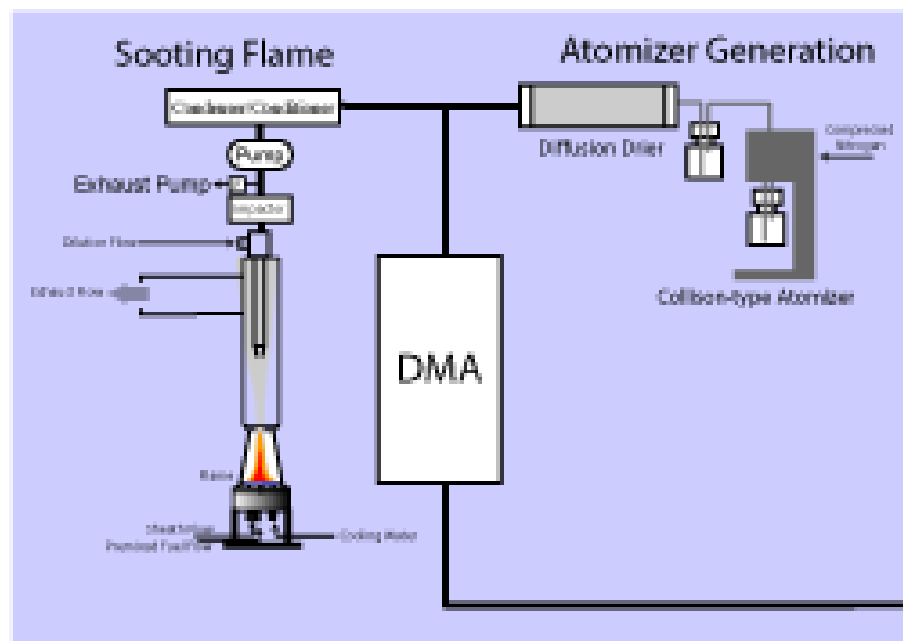
Optical properties vs. degree of aging



SOA formation less absorption (SOA is less brown than POA)

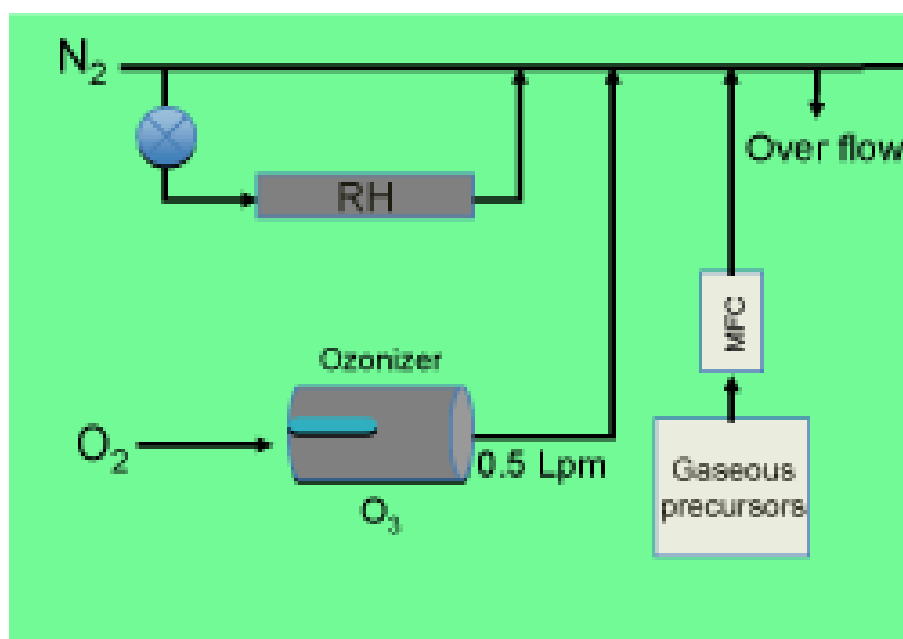
POA aging less absorption (bleaching)

BC+coating experimental intensive

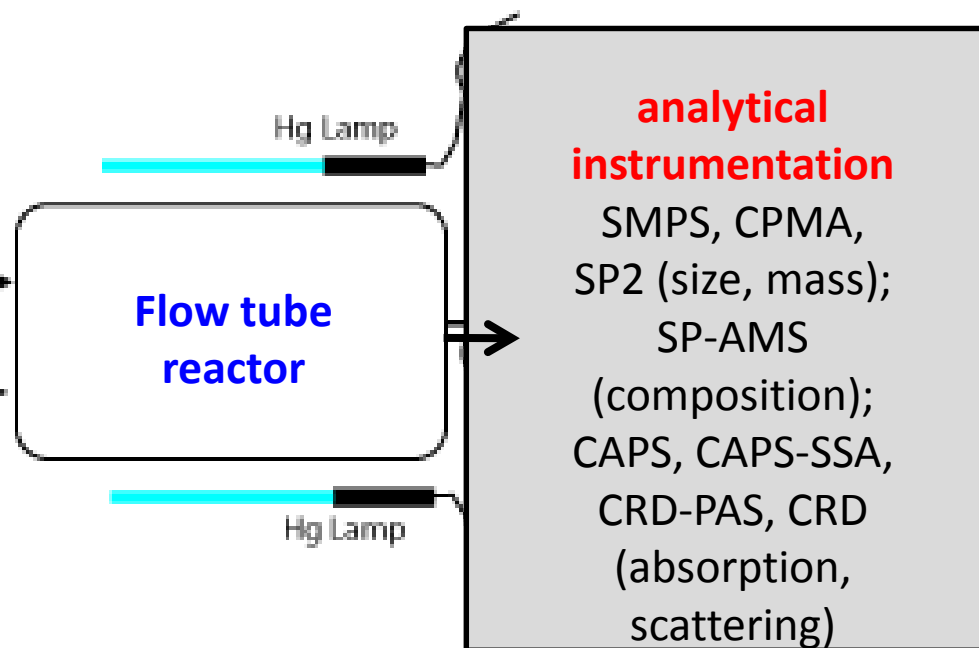


soot particle generation

*March 2015,
Boston College*



reagent/oxidant preparation



BC-coatings: Experimental matrix

BC source

- soot from inverted diffusion flame (denuded at 300°C)
 - CH₄, ethylene
- also atomized black carbon spheres

Particle size

- mostly polydisperse, 30-300 nm

Aging type

- OH/O₃ only (heterogeneous ox.)
- OH + SO₂ (sulfuric acid coating)
- OH/O₃ + VOCs (SOA coating: isoprene, α -pinene, or naphthalene)
- mixed (SOA+sulfuric acid, multiple SOA types)

Degree of oxidation

- 2-4 [OH] levels per system

Ongoing work: parameterization

Changes to composition/hygroscopicity + optics
as a function of atmospheric exposure

Key optical parameters determined, included in a “lookup table”
(or interpolated function) based on experimental results



Summary/conclusions

- Modeling results: Better descriptions of BC aging can reconcile models with observational constraints on both mass loadings of BC and aerosol absorption; resulting estimate of the BC DRF is lower than AR5
- Observational analysis: We develop a new indirect method for estimating BrC absorption from multi-wavelength measurements. Key results: BrC contributes 10-35% of UV absorption, BrC from biomass burning appears to “whiten” with atmospheric aging.
- Laboratory results: Heterogeneous oxidation an efficient way to change organic components of soot; oxidation can dramatically decrease “brown-ness” of BrC
- Next steps:
 - Explore how uncertainty in BrC sources, evolution impacts DRF
 - Laboratory results → implementation in models

Acknowledgements/collaborators

MIT: Eleanor Browne, Xuan Wang, Christopher Lim, David Ridley

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U. Hawaii: Anthony Clarke

U. Georgia: Geoffrey Smith

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U. Sao Paulo: Paulo Artaxo

PNNL: Lizabeth Alexander

Brookhaven: Arthur Sedlacek, Thomas Watson, Stephen Springston

Los Alamos: Allison Aiken, Manvendra Dubey

Publications from this Project:

Heald et al., ACP, 2014

Wang et al., ACP, 2014

Wang et al., ACP, 2016

Browne et al., JPCA, 2015

3 more in preparation

Lambe et al., ES&T, 2013

Sedlacek et al. AS&T, 2015

Lambe et al., J. Aerosol Sci., 2015

Onasch et al., AS&T, 2015

