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1.0 Scope

at usedate or the extraction, clean-up. derivatization, and final determination of dicofol and dicofol

metabolite residues in a California sandy loam soil. The limit of detection is 0.01 ppm or 0.03 ppm, depending of detection is 0.01 ppm or 0.03 ppm, depending upon the component. Chemical structures for o.p'- and p.p'-dicarol, o.p'and p,p'-FW152, o,p'- and p,p'-DCBP, 3-OH-p,p'-DCBP, o,p'-DCBH, p-CBA, and o-CBA are shown in Figure 1.

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#### 2.0 Principle

Dicofol and its metabolites are extracted from soil with acidic methanol. Sample cleanup is performed via liquid-liquidpartition using deionized water and methylene chloride followed by gel-permeation chromatography. Three compounds (p-CBA, o-CBA, and 3-OH-p,p'-DCBP) are derivatized (methylated) and analyzed via GC. The remaining seven compounds are analyzed via GC directly.

#### 3.0 Materials

3.1 Equipment (where brand names are listed, equivalent items may be used) HDPE bottles and caps, Nalgene, 8-oz. Shaker, Eberbach Funnels, Buchner, 12.5 cm Filter paper, Whatman GF7A, 12.5 cm Separatory funnels, 1000-ml Flasks, flat-bottom, 500-ml Funnels, glass powder Cotton balls Rotary evaporator GPC Autoprep Model 1001, ABC Laboratories, Columbia, Mo. 65202 Pipets, Class A, various volumes Syringes Slass luer-lock, 10 cc Pipets, disposable 9-inch Pasteur Centrifuge, IEC Clinical Vortex - Thermolyne® Thermometer, VWk (-10 to 260°C) Syringes, Hamilton (various volumes) pH Meter Corning, Model 140 Diazald® Kit, Aldrich Diazomethane Generator, Catalog Number 210,025-0 Tipets, disposable: 1-, 5-, and 10-ml Culture tubes and caps, 16 X 125 mm, 15-mm caps Culture tubes and caps, 13 X 100 mm, 13-mm caps Muffle furnace, Lab-Heat, Blue M

Dual Magnetic Stirrer/Heater, VWR Dyla-Dual Cylinder graduated, 500-ml Tippetor, 50-m1 Tippetor, 100-ml Flask, Erlenmeyer, 2000-ml Flask, Erlenmeyer, 4000 ml Separatory funnel, 2000-ml Mixing jug, glass, 5 gallon "This is an exact sopy of Buffer, pH 4, VWR Buffer, pH 7, VWR The original document" Balance, Mettler PM200 Balance, Sartorius 1702 date 10 1/5 89 Spatulas, measuring Cylinder, graduated, 1000-ml Cylinder, graduated, 50-ml Volumetrics and caps, Class A, 25, 50, 100 and 500-m1 Ultrasonic Cleaner, Branson, B-220

#### 3.2 Reagents

2,2,4 - Trimethylpentane, reagent grade
Acetone, reagent grade
Ethyl Ether, Burdick and Jackson
Methylene Chloride, Burdick and Jackson
Anhydrous Alcohol, Phototrex, reagent grade
Cyclohexane, Burdick and Jackson
Methanol, reagent grade
Diazald®, Aldrich

#### 3.3 Inorganics

Potassium Hydroxide, pellets
Sodium Sulfate, granular, 12-60 mesh
(Place into a muffle furnace and ignite for 6 to 8 hours at 600°C)
5N HC1
Hydrochloric acid, concentrated reagent, is diluted 1:1
(\*/\*\*) with deionized water.

#### 3.4 Solutions

#### 3.4.1 Diazomethane

The method used for diazomethane generation is derived from the Official Methods of Analysis of the AOAC, Section 41.015C, page 779, 14th Ed., 1984. Place 35 ml anhydrous allcohol, 10 ml ethyl other, 6 g KOH and 10 ml deionized Hol

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Into a point round-bottom flask obtained from the Diazarda kit. Place a magnetic stirring bar into the flask to ensure proper mixing. In the 125-ml separatory funnel from the kit, fitted on one arm of a Claisen adapter, dissolve 21.5 g of Diazalda with a total of 140 ml ethyl ether. Using a thermometer to keep the water bath temperature between 50-60°C, slowly add the Diazalda mixture to the reaction flask (submerged in the water bath) while the generated diazomethane and ethyl ether distill through a distillation apparatus fitted on the other neck of the Claisen adapter. To achieve a closed system, place a flask containing ethyl ether at the gas outlet of the vacuum distilling adapter. Store the generated diazomethane in a non-ground glass container in a freezer.

Note: Diazomethane is very reactive; read all cautionary material before use.

#### 3.4.2 Washed Water, pH=2

A 5-gallon mixing jug is villed with deionized H<sub>2</sub>O. A Corning pH probe is calibrated using VWR buffer solutions and is placed into the mixing jug. Acid solution (6N ECI) is added until the pH meter reads 2.0.

Approximately 1600 ml of the pH-adjusted  $\rm H_2O$  is placed into a 2000-ml separatory funnel along with 250 ml of methylene chloride. The mixture is shaken and the phases allowed to separate. The lower methylene chloride layer is discarded and the  $\rm H_2O$  phase is placed into storage containers.

#### 3.4.3 50% (v/v) Methylene Chloride/Cyclohexane

Mix exactly 2000 ml of methylene chloride and 2000 ml of cyclohexane.

#### 3.4.4 2,2,4-Trimethylpentane with 0.01% 0V-101

Add 50 µl of CV-101 (neat) to 500 ml of 2,2,4-trimethylpentane in a 500-ml volumetric flask. This yields a 0.01% CV-101/2,2,4-trimethylpentane solution. The solution is shaken and then sonicated (See 6.0, Note 5).

#### -. 3 Procedure

#### 4.1 Preparation of Sample

Soil is homogenized by processing through a Straub 42 grist mill (or equivalent) in the presence of dry ice. The

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-is placed in an upright freezer (at -20°C) working the dry ice has sublimed.

> The sample is then placed in a walk-in treezer for storage at -20°C.

#### 4.2 Extraction

Weigh a 50.0-g subsample into an 300c. HDPE Walgene bottle. All fortifications are made at this time. Add 2 ml of 6N HCl and 100 ml of methanol (Sec-6.0, Note 1). Adjust the 100 ml of methanol to account for the volume of solvent used for fortification. Place the sample in a shaker box and shake for 30 minutes on high speed, then 30 minutes on low speed.

Vacuum filter the sample through GF/A filter paper into a 1000-ml separatory funnel. Rinse the sample and filter paper with two 50-ml methanul washes and filter into the separatory funnel.

#### 4.3 Liquid-Liquid Partition

Add 500 ml of methylene chloride-washed deionized water (pH=2) to the sample. Partition the sample with three 100-ml portions of methylene chloride. Pass the methylene chloride layer through a powder funnel containing ignited sodium sulfate into a 500-ml flat-bottom flask. Rinse the funnel and sodium sulfate with 50 ml of methylene coloride. Evaporate the sample and bring to 15 ml with cyclchexane: methylene chloride, 50:50 (v/v) GPC solvent.

#### 4.4 Gel-Permention Chromatography (GPC)

Load s 5-ml aliquot of the sample onto the GPC with the following parameters:

Flow: 5 ml/minute

Mobile Phase: cyclohexane: methylene chloride, 50:50

(v/v)

Dump Time: 25 minutes Collect Time: 29 minutes

Wash Time: O minutes

Indiect the sample in a 250-ml flat-bottom flask. Evaporate the sample to dryness and bring to exactly 5 ml with ethyl ether.

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# 4.5 Underivatized Sample Analys Say R.F. Kennedy date 10 H

Place a 1-ml aliquot of the 5-ml ethyl ether extract into a test tube. Take to dryness under a gentle stream of nitrogen and bring to an appropriate volume with 2,2,4-trimethylpentane/0.01% OV-101; add \alpha-chloredane to yield 20 ng/ml. Submit an aliquot for GC analysis (See 6.0, Note 7).

#### 4.6 Derivatized Sample Analysis

Place a 1-ml aliquot of the 5-ml athyl ether extract into a test tube. Add 10 µl of 6N HC1 and 0.5 ml of diazomethane solution. Cap the sample, vortex, and derivatize undisturbed for 30 minutes.

Add 2,2,4-trimethylpentane (approximately 1 ml) to the sample as a keeper. Evaporate under a gentle stream of nitrogen to remove the diazomethane and the ethyl ether.

Note: Do not take the sample to dryness. The methylated standards are voratile (See 6.0, Note 4).

Bring the sample to an appropriate volume with 2.2,4-trimethylpentane/0.01% OV-101 and add  $\alpha$ -chlordane to yield 20 ng/ml. Submit an aliquot for GC analysis.

Note: All ethyl ether must be removed before injection into the GC.

## 4.7 Capillary GC Determination of Dicofol and Dicofol Metabolites

# 4.7.1 General

Concentrations of o,p'-and p,p'-dicofol, o,p'- and p.p'-FWIS2, o,p'- and p,p'-DCBP, o,p'-DCBH and methyl-derivatives of o- and p-CBA and 3-OH-p,p'-DC3P are determined by Capillary Gas Chromatography using EC detection and a DB-5 column (28.5m X 0.25mm, 0.25µ film thickness). GC conditions for the underivatized and derivatized compound analyses are given in Tables I and II, respectively. The programs may be used interchangeably if interferences permit.

#### 4.7.2 Standardization

Prepare separate stock solutions containing 25 mg of compound in 25 ml of methanol. Prepare a mixed stock

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By R.F. Kennedy date 10-16-89 solution (4 mix) by combining 1 ml (of the o.p'-/ and

p,p'-DCBP and o,p'- and p,p'-FW152 1 mg/ml stock solutions and diluting to a final volume of 25 ml with methanol-Prepare a second mixed stock solution (6 mix) by combining 1 ml each of o,p'- and p,p'-dicofol, methylated o-CfA, and methylated 3-OH-p,p'-DCBP and 3 ml each of o.p'-DCBH and methylated p-CBA; dilute to a final volume of ml with methanol. Low EC detector response for o.p-DCBH and methylated p-CBA dictates the three-fold concentration increase of these compounds over the concentrations of the other compounds in the 6 mix standard solution.

Isomers of FW152 and DCBP cannot mixed with the dicofol isomers since the latter can degrade to metabolites upon injection into the

Dilute the mixed stock solutions 1 to 40 with 2,2,4-trimethylpentane to yield pg/ml GC mixed stock solutions. Make serial dilucions of the GC mixed stock solutions using 2,2,4-trimethylpentane containing 0.01% OV-101 to obtain working GC sofutions in the range of 5 to 100 ng/ml (o,p-DCBH and methylated p-CBA are 15 to 300 During each Einal standard dilution, add ng/ml). a-chlordane (internal standard) at the rate of 20 ng/ml (See 6.0, Note 7).

Standardize the gas chromatograph under the conditions stated in Tables for II by making 1-2 µl injections of the mixed standard solutions. Determine the peak heights or areas of the injected standards using an integrator or computer acquisition system and calculate the ratio of the compound response to the internal standard response. Enter the standardization data into an appropriate electronic calculator or computer (e.g. RP-1000) with a customized program to calculate a standard curve of standard concentration (ng/ml) versus height or area ratio. The curve may be linear or quadratic, depending on the detector responsa.

### 4.7. □ Detection of Sample Residues

Inject a 1-2 µl aliquot (equivalent to the standard volumes for that run) of the sample extract into the GC using the same conditions as for the standards. Compare the peak height or area ratio for the unknown sample to the standard curve to determine ng/ml of detected compound in the final GC solution.

Calculate the residue values in ppm using the following equation:

ppm = axbxcxd exfxg

where:

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a = concentration in the final solution (ng/mb)

b = final GC volume (ml)

c = sample volume after GPC cleanup (ml)

d = sample volume before GPC cleanup (ml)

e = aliquot taken for derivatization or dilution (ml)

f = aliquot taken for GPC cleanup (ml)

g = sample weight (g)

4.7.4 Correction of Residue Values for On-Column Degradation

Should the breakdown of the accofol to DCBP or FW152 in the GC be significant (response greater than the lowest injection standard), then sample residues are corrected for this breakdown. Using the calculated standard concentration from the standard curves of dicofol and the metabolite produceu, corresponding produced, calculate a ratio between the metabolite and its dicofol //parent o,p'-DCBP (e.g. So,p'-dicofol). Generally, standards from the upper portion of the curve (60 ag/ml and higher) should be used to determine the ratio. Less concentrated standards tend to give variable degradation ratio values. The decision as to which standards will give reliable degradation gatio values is left to the judgement of the analyst. The average ratio of the metabolite to the parent is multiplied by the parent residue value in each sample, and the resulting residue value is then subtracted from the metabolite residue value. An example calculation follows:

Detected residue values: 0.115 ppm p,p'-DCBP and 1.25 ppm p,p'-dicofol

Average metabolite-to-parent standard ratio = 0.075

Corrected residue  $\pi$  value for p,p'-DCBP = 0.115 ppm - (1.25 % 0.075) = 0.021 ppm

On-column degradation is deemed unacceptable if the average metabolite-to-parent standard ratio is over 0.20. If this occurs, steps will be taken to reduce the breakdown, and the sample set will be reinjected.

#### 4.7.5 GC Glass Insert

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To minimize the breakdown of the dicofols to their metabolites upon injection into the GC, the glass intert in the injector is treated with phosphoric acid and or the following manner:

Boil the insert (Varian #03-94943-7-00) in 4.41 prosphoric scid for I hour. Remove the insert from the acid and dry in an oven at -120°C. Do not rinse; blot dry only. Thoroughly coat the inside and outside of the insert with the liquid OV-101 using a pipe cleaner. Also, coat any metal parts of the injector that may come in contact with the sample. Place the insert into the injector and relie the temperature slowly (5°C/min) to 240°C. Purge the injector with helium (column disconnected) and condition the insert overnight. Reconnect the column and begin injections (See 6.0, Notes 5 and 6).

Table I

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GC Operating Conditions for the Determination of o,p'and p,p'-Dicofol, o,p'- and p,p'-FW152, o,p'and p,p'-DCBP and o,p'-DCBH

Instrument: Varian Vista 6000 Capillary Gas Christatograph, or equivalent

Detector: Electron Capture

Column: DB-5 28.5 m X 0.25 mm, 0.25 µ film thickness, or equivalent

Gas Flow Rates: Carrier: 1.7 ml/min Helium

Makeup: 28 ml/min Nitrogen

Split Flow: 50 ml/min

Temperatures: Injector: 240°C

Detector: 300°C

Oven: Initial: 100°C

Initial Hold: 2.1 min.

Ramp: 30°C/min Final: 220°C

Final Hold: 16.9 min

Injector split times: On = ab 2.00 min

Off= at 23.00 min

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Table II

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GC Operating Conditions for the Determination of Methylated o- and p-CBA and 3-OH-p,p'-DCBP.

Instrument: Varian Vista 6000 Capillary

Gas Chyomatograph,

equivalent

Detector: Electron Capture

Column: DB-5 28.5 m X 0.25 mm, 0.25 μ film thic

Gas Flow Rates: Carrier: 1.7 ml/min Helion/ Makeup: 28 ml/min Nitrogen

Split Flow: 50 ml/min

Temperatures:

Injector: 240°C

Detector: 300°C

Oven: Initial: 95°6

Initial Holden 3.1 min.

Ramp: 10 Comin lst Final 135°C Ramp: 30°C/min 2nd Final: 220°C Final Hold: 14.50 min

Injector split times: On = at 2.00 min Off= at 23.00 min

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6.0 (Cautionary Notes)

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There are several critical steps in this method of which the analyst should be aware. They are listed below:

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- (1) Dicofol is easily degraded to DCBP in neutral or alkaline environments. Sample extraction using an acidic extraction solvent is mandatory (step 4.2).
- (2) Substantial chromatographic interferences, especially at the low temperature encountered early in a chromatographic run, can be avoided by pre-extracting the acidified deionized water noted in step 3.4.2 with methylene coloride.
  - (3) Step 4.3, Liquid-Liquid Partition, user acidified (pH = 2) water to dilute the methanol extract. Soil (50 g) contains substantial buffer capacity which partially neutralizes the 6N HCl added to the methanol extraction solvent. Failure to use water at pH = 2 results to partial degradation of to DCBP and substantial losses dicofol will not partition chlorobenzoates (R-COOT methylene chloride).
  - Step 4.6 details a solvent exchange from ether to 2,2,4-trimethylpentane. The latter is added to the ether prior to evaporation under nitrogen gas. It acts as a keeper to minimize losses of methylated components, especially the CBA a. The sample must not go dry at any time after derivatization.
  - (5) Dicofol (o,p'-'and p,p'-isomers) degrades to the corresponding DCBP in unprotected GC injectors. At elevated temperatures (>200°C), the conversion is quantitative. lower temperatures, the conversion is minimal (ex. at 120°C), but the dicofols are not sufficiently volatile at these temperatures and do not load onto the GC column. Two procedures afford a solution, as follows: (1) Step 4.7.5 outlines a procedure for protecting injector and acid-washed injector glass inserts by coating with OV-101. The method is crude, but reproducible; an overnight bake-out at 240°C is, of course, necessary. Repeated dipping of the insert into a 5% OV-101 in CH2Cl2 solution affords a satisfactory insert, but the effect is transitory (ca 40 injections) at (2) Final sample dilutions are prepared in 2,2,4-trimethylpentane containing 0.01% OV-101. The OV-101 affords additional protection from the dicofol to DCBP conversion.

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- (6) Step 4.7.5 notes that glass inserts are boiled in acid before coating with OV-101. Inserts which are boiled in acid, but not coated with OV-101, completely inhibit the dicofol to DCBP conversion (<2%). However, these same uncoated inserts cause an unuse ' and unexpected coarrision of FW152 isomers to DDE isomers, especially at elevated temperatures (>200°C). Boiling in acid and coating with OV-101 inhibits both processes.
- (7) Chlordane (a-isomer) was used as an internal standard for this method; its choice was ideal in that it yielded a sensitive, well-shaped peak at an interference-free retention time. The internal standard was mandatory for reproducible results. The combination of 10 separate components eluting over a wide temperature range (95° 220°C) yielded irreproducible peak areas and heights between individual injections. However, the ratio between component areas/heights and the internal standard area/height remained remarkably consistent for long periods of time.

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#### FIGURE 1

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p,p'-dicofol

{1-(2-chlorophenyl)-1-(4-chlorophenyl)-2,2,2-trichloroethanol]

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ρ.ρ'-FW152

√l-bis(4-chlorophenyl)trichloroethanol]

[1-(2-chlorophenyl)-1-(4-chlorophenyl)-2,2-dichloroethanol)

P,P'-DCBP

(1,1-bis(4-chloropheny1)-2,2-dichloroethanol]

[2,4'-dichlorobenzophenone

o.p'-DCBH

[2,4'-dichlorobenzhydrol]

'-dichlorobenzophenone]

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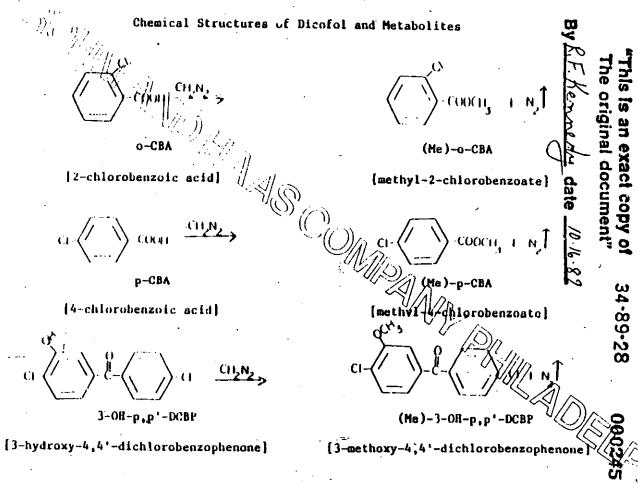
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### FIGURE 1 (Continued)



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