



United States
Environmental Protection Agency

Office of Chemical Safety and
Pollution Prevention

**Proposed Designation of
1, 2-Dichloroethane
(CASRN 107-06-2)
as a High-Priority Substance
for Risk Evaluation**

August 22, 2019

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Acronyms and Abbreviations

Term	Description
ACGIH	American Conference of Governmental Industrial Hygienists
ATSDR	Agency for Toxic Substances and Disease Registry
Biomon.	Biomonitoring
BOD	Biochemical oxygen demand
BP	Boiling point
CAA	Clean Air Act
CASRN	Chemical Abstracts Service Registry Number
CBI	Confidential business information
CDR	Chemical Data Reporting
CERCLA	Comprehensive Environmental Response, Compensation, and Liability Act
CFR	Code of Federal Regulations
Concen.	Concentration
CWA	Clean Water Act
CPDat	Chemical and Products Database
ECHA	European Chemicals Agency
FDA	U.S. Food and Drug Administration
FR	Federal Register
EPCRA	Emergency Planning and Community Right-to-Know Act
GC	Gas chromatography
HPLC	High performance liquid chromatography
IUR	Inventory Update Rule
IRIS	Integrated Risk Information System
K	Thousand
K _{OC}	Organic carbon-water partition coefficient
K _{OW}	Octanol-water partition coefficient
M	Million
MCL	Maximum contaminant level
MCLG	Maximum contaminant level goal
MITI	Ministry of International Trade and Industry
MP	Melting point
NAICS	North American Industry Classification System
NIH	National Institute of Health
NKRA	Not known or reasonably ascertainable
NPDES	National Pollutant Discharge Elimination System
NR	Not reported
OECD	Organisation for Economic Co-operation and Development
·OH	Hydroxyl radical
PEL	Permissible Exposure Limit

Term	Description
ACGIH	American Conference of Governmental Industrial Hygienists
ATSDR	Agency for Toxic Substances and Disease Registry
Biomon.	Biomonitoring
POTW	Publicly owned treatment works
RCRA	Resource Conservation and Recovery Act
REL	Recommended Exposure Limit
SARA	Superfund Amendments and Reauthorization Act of 1986
SMILES	Simplified molecular-input line-entry system
TG	Test guidance
TLV	Threshold Limit Value
TRI	Toxics Release Inventory
TSCA	Toxic Substances Control Act
VP	Vapor pressure
WS	Water solubility

1. Introduction

In section 6(b)(1)(B) of the Toxic Substances Control Act (TSCA), as amended, and in the U.S. Environmental Protection Agency's (EPA's) implementing regulations (40 CFR 702.3)¹, a high-priority substance for risk evaluation is defined as a chemical substance that EPA determines, without consideration of costs or other non-risk factors, may present an unreasonable risk of injury to health or the environment because of a potential hazard and a potential route of exposure under the conditions of use, including an unreasonable risk to potentially exposed or susceptible subpopulations identified as relevant by EPA.

Before designating prioritization status, under EPA's regulations at 40 CFR 702.9 and pursuant to TSCA section 6(b)(1)(A), EPA will generally use reasonably available information to screen the candidate chemical substance under its conditions of use against the following criteria and considerations:

- the hazard and exposure potential of the chemical substance;
- persistence and bioaccumulation;
- potentially exposed or susceptible subpopulations;
- storage near significant sources of drinking water;
- conditions of use or significant changes in the conditions of use of the chemical substance;
- the chemical substance's production volume or significant changes in production volume; and
- other risk-based criteria that EPA determines to be relevant to the designation of the chemical substance's priority.

This document presents the review of the candidate chemical substance against the criteria and considerations set forth in 40 CFR 702.9 for a may present risk finding. The information sources used are relevant to the criteria and considerations and consistent with the scientific standards of TSCA section 26(h), including, as appropriate, sources for hazard and exposure data listed in Appendices A and B of the *TSCA Work Plan Chemicals: Methods Document* (February 2012) (40 CFR 702.9(b)). EPA uses scientific information that is consistent with the best available science. Final designation of the chemical substance as a high-priority substance would immediately initiate the risk evaluation process as described in the EPA's final rule, *Procedures for Chemical Risk Evaluation Under the Amended Toxic Substances Control Act* (40 CFR 702)¹.

1,2-Dichloroethane is one of the 40 chemical substances initiated for prioritization as referenced in the March 21, 2019 notice (84 FR 10491)². EPA has determined that 1,2-dichloroethane is a suitable candidate for the proposed designation as a high-priority chemical substance. The proposed designation is based on the results of the review against the aforementioned criteria and considerations as well as review of the reasonably available information on 1,2-dichloroethane, including relevant information received from the public and other information as appropriate.

¹ NOTE: For all 40 CFR 702 citations, please refer to: <https://www.govinfo.gov/content/pkg/CFR-2018-title40-vol33/xml/CFR-2018-title40-vol33-part702.xml> and <https://www.regulations.gov/document?D=EPA-HQ-OPPT-2016-0654-0108>

² <https://www.federalregister.gov/documents/2019/03/21/2019-05404/initiation-of-prioritization-under-the-toxic-substances-control-act-tsca>

EPA will take comment on this proposed designation for 90 days before finalizing its designation of 1,2-dichloroethane. The docket number for providing comments on 1,2-dichloroethane is EPA-HQ-OPPT-2018-0427 and is available at www.regulations.gov.

The information, analysis and basis used for the review of the chemical is organized as follows:

- *Section 1 (Introduction):* This section explains the requirements of the amended TSCA and implementing regulations – including the criteria and considerations – pertinent to the prioritization and designation of high-priority chemical substances.
- *Section 2 (Production volume or significant changes in production volume):* This section presents information and analysis on national aggregate production volume of the chemical substance.
- *Section 3 (Conditions of use or significant changes in conditions of use):* This section presents information and analysis regarding the chemical substance’s conditions of use under TSCA.
- *Section 4 (Potentially exposed or susceptible subpopulations):* This section presents information and analysis regarding potentially exposed or susceptible subpopulations, including children, women of reproductive age, and workers, with respect to the chemical substance.
- *Section 5 (Persistence and bioaccumulation):* This section presents information and analysis regarding the physical and chemical properties of the chemical substance and the chemical’s fate characteristics.
- *Section 6 (Storage near significant sources of drinking water):* This section presents information and analysis considered regarding the risk from storage of the chemical substance near significant sources of drinking water.
- *Section 7 (Hazard Potential):* This section presents the hazard information relevant to the chemical substance.
- *Section 8 (Exposure Potential):* This section presents information and analysis regarding the exposures to the chemical substance.
- *Section 9 (Other risk-based criteria):* This section presents the extent to which EPA identified other risk-based criteria that are relevant to the designation of the chemical substance’s priority.
- *Section 10 (Proposed designation):* Based on the results of the review performed and the information and analysis presented, this section describes the basis used by EPA to support the proposed designation.

2. Production volume or significant changes in production volume

Approach

EPA considered current volume or significant changes in volume of the chemical substance using information reported by manufacturers (including importers). EPA assembled reported information for years 1986 through 2015 on the production volume for 1,2-dichloroethane reported under the Inventory Update Reporting (IUR) rule and Chemical Data Reporting (CDR) rule.³

Results and Discussion

The national aggregate production volume, which is presented as a range to protect individual site production volumes that are confidential business information (CBI), is presented in Table 1.

Table 1. 1986–2015 National Aggregate Production Volume Data (Production Volume in Pounds)

Chemical ID	1986	1990	1994	1998	2002	2006	2011	2012	2013	2014	2015
1,2-Dichloroethane (107-06-2)	>10M to 50M	>50M to 100M	>50M to 100M	>50M to 100M	>50M to 100M	≥1B	28,109,768, 185	20B to 30B	20B to 30B	20B to 30B	20B to 30B

M = million, B = billion

Reference: [U.S. EPA \(2013\)](#) and [U.S. EPA \(2017\)](#)

Production volume of 1,2-dichloroethane in 2015, as reported to EPA during the 2016 CDR reporting period, was between 20 and 30 billion lbs. National aggregate production volume for 1,2-dichloroethane has generally increased over time. In reporting year (RY) 1986, aggregate production volume was between 10 million and 50 million lbs, and from RY 1990 to RY 2002, it was between 50 million and 100 million lbs. In RY 2006, aggregate production volume was at least 1 billion lbs, and in 2011, 28.1 billion lbs of 1,2-dichloroethane were manufactured or imported. From 2011 to 2015, national aggregate production volume for this chemical was between 20 billion and 30 billion lbs (Table 1).

³ Over time, the requirements for reporting frequency, production volume thresholds, and chemical substances under the Chemical Data Reporting (CDR) rule have changed. CDR was formerly known as the Inventory Update Rule (IUR). The first IUR collection occurred in 1986 and continued every four years through 2006. As part of two rulemakings in 2003 and 2005, EPA made a variety of changes to the IUR, including to change the reporting frequency to every five years to address burden associated with new reporting requirements. Additional changes to reporting requirements were made in 2011, including to suspend and replace the 2011 submission period with a 2012 submission period, return to reporting every four years, and require the reporting of all years beginning with 2011 production volumes. The reporting of production volumes for all years was added because of the mounting evidence that many chemical substances, even larger production volume chemical substances, often experience wide fluctuations in production volume from year to year. In addition, also as part of the 2011 IUR Modifications final rule (76 FR 50816, Aug 16, 2011), EPA changed the name of the regulation from IUR to CDR to better reflect the distinction between this data collection (which includes exposure-related data) and the TSCA Inventory itself (which only involves chemical identification information).

3. Conditions of use or significant changes in conditions of use

Approach

EPA assembled information to determine conditions of use or significant changes in conditions of use of the chemical substance. TSCA section 3(4) defines the term “conditions of use” to mean the circumstances, as determined by the Administrator, under which a chemical substance is intended, known, or reasonably foreseen to be manufactured, processed, distributed in commerce, used, or disposed of.

A key source of reasonably available information that EPA considered for determining the conditions of use for 1,2-dichloroethane was submitted by manufacturers (including importers) under the 2012 and 2016 CDR reporting cycles. CDR requires manufacturers (including importers) to report information on the chemical substances they produce domestically or import into the United States greater than 25,000 lbs per site, except if certain TSCA actions apply (in which case the reporting requirement is greater than 2,500 lbs per site). CDR includes information on the manufacturing, processing, and use of chemical substances. Based on the known manufacturing, processing and uses of this chemical substance, EPA assumes distribution in commerce. CDR may not provide information on other life-cycle phases such as distribution or chemical end-of-life after use in products (i.e., disposal). While EPA may be aware of additional uses, CDR submitters are not required to provide information on chemical uses that are not regulated under TSCA.

For chemical substances under review that are included on the Toxics Release Inventory (TRI) chemical list, information disclosed by reporting facilities in Part II Section 3 (“Activities and Uses of the Toxic Chemical at the Facility”) of their TRI Form R reports was used to supplement the CDR information on conditions of use (Tables 4, 5 and 6). There is not a one-to-one correlation between conditions of use reported under CDR and information reported in Part II Section 3 of the TRI Form R because facilities are not required to disclose in their Form R submissions the specific uses of TRI chemical substances they manufactured on-site or imported. In addition to the information disclosed in Part II Section 3 of the TRI Form R, the information pertaining to waste management activities (i.e., disposal/releases, energy recovery, recycling, and treatment) disclosed in other sections of the Form R was also used to supplement the CDR information on conditions of use as shown in Tables 4, 5, and 6. For purposes of this proposed prioritization designation, EPA assumed end-of-life pathways that include releases to air, wastewater, and solid and liquid waste based on the conditions of use.

CDR and TRI Tables

Based on the publicly available⁴ manufacturing information, industrial processing and use information, and consumer and commercial use information reported under CDR, EPA developed a list of conditions of use for the 2016 and 2012 reporting cycles (Tables 2 and 3, respectively).

⁴ Some specific chemical uses may be claimed by CDR submitters as confidential business information (CBI) under section 14 of TSCA. In these cases, EPA has indicated that the information is CBI.

Table 2. 1,2-Dichloroethane (107-06-2) Categories and Subcategories of Conditions of Use⁵ (2016 CDR Reporting Cycle)

Life Cycle Stage	Category	Subcategory	Reference
Manufacture	Domestic manufacture/Import	CBI ⁶	U.S. EPA (2019a)
Manufacture	Domestic Manufacture	Domestic Manufacture	U.S. EPA (2019a)
Processing	Processing – incorporating into formulation, mixture or reaction product	Fuels and fuel additives in: – All other petroleum and coal products manufacturing	U.S. EPA (2019a)
Processing	Processing – incorporating into formulation, mixture or reaction product	Functional fluids: – Pharmaceutical and medicine manufacturing	U.S. EPA (2019a)
Processing	Processing as a reactant	Intermediate in: – Petrochemical manufacturing – Plastic material and resin manufacturing – All other basic organic chemical manufacturing	U.S. EPA (2019a)
Processing	Recycling	Recycling	U.S. EPA (2019a)
Distribution in Commerce ^{a,b}	Distribution	Distribution	U.S. EPA (2019a)
Industrial Uses	Laboratory chemicals	– Services – Wholesale and retail trade	U.S. EPA (2019a)
Commercial Uses	Fuels and related products	Fuels and related products	U.S. EPA (2019a)
Commercial Uses	Plastic and rubber products not covered elsewhere	Plastic and rubber products not covered elsewhere	U.S. EPA (2019a)
Consumer Uses	Plastic and rubber products not covered elsewhere	Plastic and rubber products not covered elsewhere	U.S. EPA (2019a)
Disposal ^a	Disposal		
<p>^a CDR includes information on the manufacturing, processing, and use of chemical substances. CDR may not provide information on other life-cycle phases such as distribution or chemical end-of-life after use in products (i.e., disposal). The table row is highlighted in gray to indicate that no information is provided for this life-cycle stage.</p> <p>^b EPA is particularly interested in information from the public on distribution in commerce. CBI = confidential business information</p>			

⁵ Certain other uses that are excluded from TSCA are not captured in this table.

⁶ At this time, “CBI” indicates that a data element has been claimed CBI by the information submitter; it does not reflect the result of an EPA substantiation review.

Table 3. 1,2-Dichloroethane (107-06-2) Categories and Subcategories of Conditions of Use⁷ (2012 CDR Reporting Cycle)

Life Cycle Stage	Category	Subcategory	Reference
Manufacture	Domestic manufacture	Domestic manufacture	U.S. EPA (2019a)
Manufacture	Import	Import	U.S. EPA (2019a)
Processing	Processing as a reactant	Intermediates in: All other basic organic chemical manufacturing Plastics material and resin manufacturing Petrochemical manufacturing	U.S. EPA (2019a)
Processing	Processing as a reactant	Other: Used to manufacture vinyl chloride in petrochemical manufacturing	U.S. EPA (2019a)
Processing	Processing - incorporating into formulation, mixture or reaction product	Processing aids, specific to petroleum production: Oil and gas drilling, extraction, and support activities	U.S. EPA (2019a)
Processing	Recycling	Recycling	U.S. EPA (2019a)
Distribution in Commerce ^{a,b}	Distribution	Distribution	U.S. EPA (2019a)
Commercial Uses	Plastic and rubber products not covered elsewhere		U.S. EPA (2019a)
Consumer Uses	Plastic and rubber products not covered elsewhere		U.S. EPA (2019a)
Disposal ^a	Disposal		

^a CDR includes information on the manufacturing, processing, and use of chemical substances. CDR may not provide information on other life-cycle phases such as distribution or chemical end-of-life after use in products (i.e., disposal). The table row is highlighted in gray to indicate that no information is provided for this life-cycle stage.

^b **EPA is particularly interested in information from the public on distribution in commerce.**

⁷ Certain other uses that are excluded from TSCA are not captured in this table.

EPA used TRI data to identify additional conditions of use and to supplement CDR information about conditions of use. In addition, TRI information from 2017 is useful for demonstrating that a condition of use reported to CDR in 2015 is still ongoing.

Table 4. Activities and Uses Reported to TRI for 1,2-Dichloroethane, Reporting Year 2011

Activity Type	Activity	Industry Group	NAICS Code
Manufacture	Produce	Basic Chemical Manufacturing	3251
		Resin, Synthetic Rubber, and Artificial and Synthetic Fibers and Filaments Manufacturing	3252
		Pesticide, Fertilizer, and Other Agricultural Chemical Manufacturing	3253
		Other Chemical Product and Preparation Manufacturing	3259
		Plastics Product Manufacturing	3261
		Other Pipeline Transportation	4869
	Import	Basic Chemical Manufacturing	3251
		Resin, Synthetic Rubber, and Artificial and Synthetic Fibers and Filaments Manufacturing	3252
		Other Pipeline Transportation	4869
	Produce or import for on-site use/processing	Basic Chemical Manufacturing	3251
		Resin, Synthetic Rubber, and Artificial and Synthetic Fibers and Filaments Manufacturing	3252
		Other Chemical Product and Preparation Manufacturing	3259
		Plastics Product Manufacturing	3261
		Other Pipeline Transportation	4869
	Produce or import for sale/distribution	Basic Chemical Manufacturing	3251
		Resin, Synthetic Rubber, and Artificial and Synthetic Fibers and Filaments Manufacturing	3252
		Other Pipeline Transportation	4869
	Produce or import as a byproduct	Basic Chemical Manufacturing	3251
		Other Chemical Product and Preparation Manufacturing	3259
		Other Pipeline Transportation	4869
	Produce or import as an impurity	Basic Chemical Manufacturing	3251
Pesticide, Fertilizer, and Other Agricultural Chemical Manufacturing		3253	
Process	Process as a reactant	Basic Chemical Manufacturing	3251
		Resin, Synthetic Rubber, and Artificial and Synthetic Fibers and Filaments Manufacturing	3252
		Other Chemical Product and Preparation Manufacturing	3259
		Plastics Product Manufacturing	3261
		Other Pipeline Transportation	4869

Activity Type	Activity	Industry Group	NAICS Code	
	Process as an article component	Resin, Synthetic Rubber, and Artificial and Synthetic Fibers and Filaments Manufacturing	3252	
	Process as an impurity	Nonmetallic Mineral Mining and Quarrying	2123	
		Petroleum and Coal Products Manufacturing	3241	
		Basic Chemical Manufacturing	3251	
		Resin, Synthetic Rubber, and Artificial and Synthetic Fibers and Filaments Manufacturing	3252	
		Waste Treatment and Disposal	5622	
	Process as a formulation component	Basic Chemical Manufacturing	3251	
		Pesticide, Fertilizer, and Other Agricultural Chemical Manufacturing	3253	
		Other Chemical Product and Preparation Manufacturing	3259	
		Other Pipeline Transportation	4869	
	Process – repackaging	Basic Chemical Manufacturing	3251	
		Resin, Synthetic Rubber, and Artificial and Synthetic Fibers and Filaments Manufacturing	3252	
		Ship and Boat Building	3366	
		Other Pipeline Transportation	4869	
		Waste Treatment and Disposal	5622	
	Otherwise Use	Otherwise use – as a chemical processing aid	Petroleum and Coal Products Manufacturing	3241
			Basic Chemical Manufacturing	3251
			Resin, Synthetic Rubber, and Artificial and Synthetic Fibers and Filaments Manufacturing	3252
			Pesticide, Fertilizer, and Other Agricultural Chemical Manufacturing	3253
		Otherwise use – ancillary or other use	Petroleum and Coal Products Manufacturing	3241
Basic Chemical Manufacturing			3251	
Resin, Synthetic Rubber, and Artificial and Synthetic Fibers and Filaments Manufacturing			3252	
Other Chemical Product and Preparation Manufacturing			3259	
Cement and Concrete Product Manufacturing			3273	
Other Nonmetallic Mineral Product Manufacturing			3279	
Other Pipeline Transportation			4869	
Waste Treatment and Disposal			5622	
Waste Management		Disposal/releases	Nonmetallic Mineral Mining and Quarrying	2123
			Petroleum and Coal Products Manufacturing	3241
	Basic Chemical Manufacturing		3251	
	Resin, Synthetic Rubber, and Artificial and Synthetic Fibers and Filaments Manufacturing		3252	
	Pesticide, Fertilizer, and Other Agricultural Chemical Manufacturing		3253	

Activity Type	Activity	Industry Group	NAICS Code
		Other Chemical Product and Preparation Manufacturing	3259
		Plastics Product Manufacturing	3261
		Cement and Concrete Product Manufacturing	3273
		Other Nonmetallic Mineral Product Manufacturing	3279
		Ship and Boat Building	3366
		Other Pipeline Transportation	4869
		Waste Treatment and Disposal	5622
	Energy recovery	Nonmetallic Mineral Mining and Quarrying	2123
		Petroleum and Coal Products Manufacturing	3241
		Basic Chemical Manufacturing	3251
		Resin, Synthetic Rubber, and Artificial and Synthetic Fibers and Filaments Manufacturing	3252
		Other Chemical Product and Preparation Manufacturing	3259
		Cement and Concrete Product Manufacturing	3273
		Other Nonmetallic Mineral Product Manufacturing	3279
	Recycling	Waste Treatment and Disposal	5622
		Petroleum and Coal Products Manufacturing	3241
		Basic Chemical Manufacturing	3251
		Resin, Synthetic Rubber, and Artificial and Synthetic Fibers and Filaments Manufacturing	3252
		Other Nonmetallic Mineral Product Manufacturing	3279
	Treatment	Waste Treatment and Disposal	5622
		Nonmetallic Mineral Mining and Quarrying	2123
		Petroleum and Coal Products Manufacturing	3241
		Basic Chemical Manufacturing	3251
		Resin, Synthetic Rubber, and Artificial and Synthetic Fibers and Filaments Manufacturing	3252
		Pesticide, Fertilizer, and Other Agricultural Chemical Manufacturing	3253
		Other Chemical Product and Preparation Manufacturing	3259
		Plastics Product Manufacturing	3261
		Other Nonmetallic Mineral Product Manufacturing	3279
		Ship and Boat Building	3366
		Other Pipeline Transportation	4869
Waste Treatment and Disposal		5622	

Reference: [U.S. EPA, 2019b](#)

Table 5. Activities and Uses Reported to TRI for 1,2-Dichloroethane, Reporting Year 2015

Activity Type	Activity	Industry Group	NAICS Code	
Manufacture	Produce	Basic Chemical Manufacturing	3251	
		Resin, Synthetic Rubber, and Artificial and Synthetic Fibers and Filaments Manufacturing	3252	
		Pesticide, Fertilizer, and Other Agricultural Chemical Manufacturing	3253	
		Other Chemical Product and Preparation Manufacturing	3259	
		Other Pipeline Transportation	4869	
	Import	Basic Chemical Manufacturing	3251	
		Other Pipeline Transportation	4869	
	Produce or import for on-site use/processing	Basic Chemical Manufacturing	3251	
		Resin, Synthetic Rubber, and Artificial and Synthetic Fibers and Filaments Manufacturing	3252	
		Other Pipeline Transportation	4869	
	Produce or import for sale/distribution	Basic Chemical Manufacturing	3251	
		Resin, Synthetic Rubber, and Artificial and Synthetic Fibers and Filaments Manufacturing	3252	
		Other Pipeline Transportation	4869	
	Produce or import as a byproduct	Basic Chemical Manufacturing	3251	
		Other Pipeline Transportation	4869	
	Produce or import as an impurity	Basic Chemical Manufacturing	3251	
		Pesticide, Fertilizer, and Other Agricultural Chemical Manufacturing	3253	
		Other Chemical Product and Preparation Manufacturing	3259	
	Process	Process as a reactant	Basic Chemical Manufacturing	3251
			Resin, Synthetic Rubber, and Artificial and Synthetic Fibers and Filaments Manufacturing	3252
Other Chemical Product and Preparation Manufacturing			3259	
Other Pipeline Transportation			4869	
Process as an article component		Resin, Synthetic Rubber, and Artificial and Synthetic Fibers and Filaments Manufacturing	3252	
		Waste Treatment and Disposal	5622	
Process as an impurity		Nonmetallic Mineral Mining and Quarrying	2123	
		Basic Chemical Manufacturing	3251	
		Waste Treatment and Disposal	5622	
Process as a formulation component		Basic Chemical Manufacturing	3251	
		Pesticide, Fertilizer, and Other Agricultural Chemical Manufacturing	3253	
		Other Chemical Product and Preparation Manufacturing	3259	
		Other Pipeline Transportation	4869	
			Basic Chemical Manufacturing	3251

Activity Type	Activity	Industry Group	NAICS Code
	Process – repackaging	Resin, Synthetic Rubber, and Artificial and Synthetic Fibers and Filaments Manufacturing	3252
		Ship and Boat Building	3366
		Other Pipeline Transportation	4869
		Waste Treatment and Disposal	5622
Otherwise Use	Otherwise use – as a chemical processing aid	Petroleum and Coal Products Manufacturing	3241
		Basic Chemical Manufacturing	3251
		Resin, Synthetic Rubber, and Artificial and Synthetic Fibers and Filaments Manufacturing	3252
		Pesticide, Fertilizer, and Other Agricultural Chemical Manufacturing	3253
		Soap, Cleaning Compound, and Toilet Preparation Manufacturing	3256
	Otherwise use – ancillary or other use	Basic Chemical Manufacturing	3251
		Resin, Synthetic Rubber, and Artificial and Synthetic Fibers and Filaments Manufacturing	3252
		Other Chemical Product and Preparation Manufacturing	3259
		Cement and Concrete Product Manufacturing	3273
		Other Pipeline Transportation	4869
		Waste Treatment and Disposal	5622
Waste Management	Disposal/releases	Nonmetallic Mineral Mining and Quarrying	2123
		Petroleum and Coal Products Manufacturing	3241
		Basic Chemical Manufacturing	3251
		Resin, Synthetic Rubber, and Artificial and Synthetic Fibers and Filaments Manufacturing	3252
		Pesticide, Fertilizer, and Other Agricultural Chemical Manufacturing	3253
		Soap, Cleaning Compound, and Toilet Preparation Manufacturing	3256
		Other Chemical Product and Preparation Manufacturing	3259
		Cement and Concrete Product Manufacturing	3273
		Ship and Boat Building	3366
		Other Pipeline Transportation	4869
		Waste Treatment and Disposal	5622
		Energy recovery	Petroleum and Coal Products Manufacturing
	Basic Chemical Manufacturing		3251
	Resin, Synthetic Rubber, and Artificial and Synthetic Fibers and Filaments Manufacturing		3252
	Pesticide, Fertilizer, and Other Agricultural Chemical Manufacturing		3253
	Soap, Cleaning Compound, and Toilet Preparation Manufacturing		3256

Activity Type	Activity	Industry Group	NAICS Code
		Other Chemical Product and Preparation Manufacturing	3259
		Cement and Concrete Product Manufacturing	3273
		Waste Treatment and Disposal	5622
	Recycling	Basic Chemical Manufacturing	3251
		Resin, Synthetic Rubber, and Artificial and Synthetic Fibers and Filaments Manufacturing	3252
		Soap, Cleaning Compound, and Toilet Preparation Manufacturing	3256
	Treatment	Nonmetallic Mineral Mining and Quarrying	2123
		Basic Chemical Manufacturing	3251
		Resin, Synthetic Rubber, and Artificial and Synthetic Fibers and Filaments Manufacturing	3252
		Pesticide, Fertilizer, and Other Agricultural Chemical Manufacturing	3253
		Soap, Cleaning Compound, and Toilet Preparation Manufacturing	3256
		Other Chemical Product and Preparation Manufacturing	3259
		Ship and Boat Building	3366
		Other Pipeline Transportation	4869
		Waste Treatment and Disposal	5622

Reference: [U.S. EPA, 2019b](#)

Table 6. Activities and Uses Reported to TRI for 1,2-Dichloroethane, Reporting Year 2017

Activity Type	Activity	Industry Group	NAICS Code
Manufacture	Produce	Basic Chemical Manufacturing	3251
		Resin, Synthetic Rubber, and Artificial and Synthetic Fibers and Filaments Manufacturing	3252
		Pesticide, Fertilizer, and Other Agricultural Chemical Manufacturing	3253
		Other Pipeline Transportation	4869
	Import	Basic Chemical Manufacturing	3251
		Other Pipeline Transportation	4869
	Produce or import for on-site use/processing	Basic Chemical Manufacturing	3251
		Resin, Synthetic Rubber, and Artificial and Synthetic Fibers and Filaments Manufacturing	3252
		Other Pipeline Transportation	4869
	Produce or import for sale/distribution	Basic Chemical Manufacturing	3251
		Resin, Synthetic Rubber, and Artificial and Synthetic Fibers and Filaments Manufacturing	3252
		Other Pipeline Transportation	4869
			Basic Chemical Manufacturing

Activity Type	Activity	Industry Group	NAICS Code
	Produce or import as a byproduct	Other Pipeline Transportation	4869
	Produce or import as an impurity	Basic Chemical Manufacturing	3251
		Pesticide, Fertilizer, and Other Agricultural Chemical Manufacturing	3253
Process	Process as a reactant	Basic Chemical Manufacturing	3251
		Resin, Synthetic Rubber, and Artificial and Synthetic Fibers and Filaments Manufacturing	3252
		Other Chemical Product and Preparation Manufacturing	3259
		Other Pipeline Transportation	4869
	Process as an article component	Resin, Synthetic Rubber, and Artificial and Synthetic Fibers and Filaments Manufacturing	3252
	Process as an impurity	Nonmetallic Mineral Mining and Quarrying	2123
		Basic Chemical Manufacturing	3251
		Waste Treatment and Disposal	5622
	Process as a formulation component	Basic Chemical Manufacturing	3251
		Pesticide, Fertilizer, and Other Agricultural Chemical Manufacturing	3253
		Other Pipeline Transportation	4869
	Process – repackaging	Basic Chemical Manufacturing	3251
		Resin, Synthetic Rubber, and Artificial and Synthetic Fibers and Filaments Manufacturing	3252
		Other Pipeline Transportation	4869
		Waste Treatment and Disposal	5622
	Otherwise Use	Otherwise use – as a chemical processing aid	Petroleum and Coal Products Manufacturing
Basic Chemical Manufacturing			3251
Resin, Synthetic Rubber, and Artificial and Synthetic Fibers and Filaments Manufacturing			3252
Pesticide, Fertilizer, and Other Agricultural Chemical Manufacturing			3253
Soap, Cleaning Compound, and Toilet Preparation Manufacturing			3256
Otherwise use – as a manufacturing aid		Pesticide, Fertilizer, and Other Agricultural Chemical Manufacturing	3253
Otherwise use – ancillary or other use		Basic Chemical Manufacturing	3251
		Resin, Synthetic Rubber, and Artificial and Synthetic Fibers and Filaments Manufacturing	3252
		Pesticide, Fertilizer, and Other Agricultural Chemical Manufacturing	3253
		Cement and Concrete Product Manufacturing	3273

Activity Type	Activity	Industry Group	NAICS Code	
		Other Nonmetallic Mineral Product Manufacturing	3279	
		Other Pipeline Transportation	4869	
		Waste Treatment and Disposal	5622	
Waste Management	Disposal/releases	Nonmetallic Mineral Mining and Quarrying	2123	
		Petroleum and Coal Products Manufacturing	3241	
		Basic Chemical Manufacturing	3251	
		Resin, Synthetic Rubber, and Artificial and Synthetic Fibers and Filaments Manufacturing	3252	
		Pesticide, Fertilizer, and Other Agricultural Chemical Manufacturing	3253	
		Soap, Cleaning Compound, and Toilet Preparation Manufacturing	3256	
		Other Chemical Product and Preparation Manufacturing	3259	
		Cement and Concrete Product Manufacturing	3273	
		Other Nonmetallic Mineral Product Manufacturing	3279	
		Other Pipeline Transportation	4869	
		Waste Treatment and Disposal	5622	
		Energy recovery		Petroleum and Coal Products Manufacturing
	Basic Chemical Manufacturing			3251
	Resin, Synthetic Rubber, and Artificial and Synthetic Fibers and Filaments Manufacturing			3252
	Pesticide, Fertilizer, and Other Agricultural Chemical Manufacturing			3253
	Soap, Cleaning Compound, and Toilet Preparation Manufacturing			3256
	Other Chemical Product and Preparation Manufacturing			3259
	Cement and Concrete Product Manufacturing			3273
	Other Nonmetallic Mineral Product Manufacturing			3279
	Waste Treatment and Disposal			5622
	Recycling		Basic Chemical Manufacturing	3251
			Resin, Synthetic Rubber, and Artificial and Synthetic Fibers and Filaments Manufacturing	3252
			Soap, Cleaning Compound, and Toilet Preparation Manufacturing	3256
			Waste Treatment and Disposal	5622
	Treatment		Nonmetallic Mineral Mining and Quarrying	2123
			Petroleum and Coal Products Manufacturing	3241
			Basic Chemical Manufacturing	3251
			Resin, Synthetic Rubber, and Artificial and Synthetic Fibers and Filaments Manufacturing	3252
			Pesticide, Fertilizer, and Other Agricultural Chemical Manufacturing	3253

Activity Type	Activity	Industry Group	NAICS Code
		Soap, Cleaning Compound, and Toilet Preparation Manufacturing	3256
		Other Chemical Product and Preparation Manufacturing	3259
		Other Nonmetallic Mineral Product Manufacturing	3279
		Other Pipeline Transportation	4869
		Waste Treatment and Disposal	5622

Reference: [U.S. EPA, 2019b](#)

CDR and TRI Summary and Additional Information on Conditions of Use

For the 2016 CDR, five sites reported use of 1,2-dichloroethane as an intermediate reactant in plastic material and resin manufacturing, seven sites reported as an intermediate reactant in all other basic organic chemical manufacturing, one site reported as an intermediate reactant in petrochemical manufacturing, and three sites reported their industrial sector as CBI for use as an intermediate reactant. One site reported use of 1,2-dichloroethane as a fuel additive in all other petroleum and coal products manufacturing. One site reported use of 1,2-dichloroethane as a closed system functional fluid in pharmaceutical and medicine manufacturing. One site reported use of 1,2-dichloroethane as a laboratory chemical in the services and wholesale and retail trade sector.

In the 2016 CDR reports, one site reported use of 1,2-dichloroethane in plastic and rubber products for both commercial and consumer uses. One site reported use of 1,2-dichloroethane in fuels and related products for commercial use. Consumer uses were also identified in additional databases, which are included in the Exposure Potential section (Section 8).

For 1,2-dichloroethane there has not been a significant change in the conditions of use based on 2012 and 2016 CDR reports.

TRI data reported in Part II Section 3 of the TRI Form R (“Activities and Uses of the Toxic Chemical at the Facility”) were compiled for RY 2011, RY 2015, and RY 2017. RY 2011, RY 2015, and RY 2017 reflect the chemical activities at reporting facilities in calendar years 2011, 2015, and 2017, respectively. Each facility filing a TRI Form R discloses activities that apply to the TRI chemical at the facility. The TRI data presented above are from the TRI dataset updated in April 2019. Tables 4, 5, and 6 present the activities and uses reported to TRI by industry group for 2011, 2015, and 2017. Waste management activity type include all industry groups that reported to TRI using each waste management activity for 1,2-dichloroethane.

During the first public comment period for the draft high-priority designation of 1,2-dichloroethane, one commenter stated that 1,2-dichloroethane is used for conversion to vinyl chloride, conversion to other chlorinated hydrocarbons, conversion to ethylene amines, as a pharmaceutical precursor, as a precursor for flame retardants, as a laboratory reagent, as a laboratory standard, and as an oxidation inhibitor in large scale controlled oxidative chemical reactions (EPA-HQ-OPPT-2018-0427-0006). Another public comment indicates that 1,2-dichloroethane is found as an impurity in some adhesives in amounts less than 0.1% (EPA-HQ-OPPT-2018-0421-0003).

One public comment states that specific aerospace industrial uses of that 1,2-dichloroethane include, but may not be limited to, heat resistant adhesives for primary and secondary structural and external metallic airframe parts, intermediates for the synthesis of organics, dispersants for plastics and elastomers, components of fumigants and insecticides, low friction and anti-knock coatings, bond primers, as a component in paint and varnish and paint removers, as a component of degreasing and cleaning solvents, and as a reagent (EPA-HQ-OPPT-2018-0421-0003). The commenter also stated that 1,2-dichloroethane can be used as a constituent in adhesives, intermediates, dispersants, fumigants and insecticides, coatings and paint, paint removers, solvents, and reagents (EPA-HQ-OPPT-2018-0427-0005). The use as a pesticide ingredient (fumigants and insecticides), however, was said by one commenter to have ceased (EPA-HQ-OPPT-2018-0427-0006).

Should the Agency decide to make a final decision to designate this chemical substance as a high-priority substance, further characterization of relevant TSCA conditions of use will be undertaken as part of the process of developing the scope of the risk evaluation.

4. Potentially exposed or susceptible subpopulations

Approach

In this review, EPA considered reasonably available information to identify potentially exposed or susceptible subpopulations, such as children, women of reproductive age, consumers, workers, or the elderly. EPA analyzed processing and use information included on the CDR Form U. These data provide an indication about whether children or other susceptible subpopulation may be potentially exposed. EPA also used human health hazard information to identify potentially exposed or susceptible subpopulations.

Results and Discussion

At this stage, EPA identified consumers and workers as subpopulations who may be potentially exposed or susceptible subpopulations for 1,2-dichloroethane assessment.

Children

EPA used data reported to the 2012 and 2016 CDR to identify uses in products and articles intended for children over time for 1,2-dichloroethane. The 2012 and 2016 CDR did not report any use in children's products.

Women of reproductive age (e.g., pregnant women per TSCA statute)

EPA identified developmental and reproductive toxicity studies following 1,2-dichloroethane exposure; however, no effects were observed in these studies. Therefore, women of reproductive age are not considered a susceptible subpopulation with respect to 1,2-dichloroethane at this time. During the scoping and risk evaluation process, the issue of potentially-exposed or susceptible subpopulations will be reconsidered following a systematic search of the relevant scientific literature.

Workers

Please refer to the Exposure Potential section (Section 8) for a summary of potential occupational exposures, which EPA indicates that workers are potentially exposed or susceptible subpopulations based on greater exposure.

Consumers

Please refer to the Exposure Potential section (Section 8) for a summary of potential consumer exposures, which EPA indicates that consumers are potentially exposed or susceptible subpopulations based on greater exposure.

5. Persistence and bioaccumulation

Approach

EPA reviewed reasonably available data, such as physical and chemical properties and environmental fate characteristics, to understand 1,2-dichloroethane's persistence and bioaccumulation.

Physical and Chemical Properties and Environmental Fate Tables

Tables 7 and 8 summarize the physical and chemical properties and environmental fate characteristics of 1,2-dichloroethane, respectively.

Table 7. Physical and Chemical Properties of 1,2-Dichloroethane

Property or Endpoint	Value ^a	Reference
Molecular Formula	C ₂ H ₄ Cl ₂	CRC Handbook (Rumble, 2018)
Molecular Weight	98.959 g/mole	CRC Handbook (Rumble, 2018)
Physical State	Liquid	CRC Handbook (Rumble, 2018)
Melting Point	-35.6 °C ^b	HSDB (2010) ; CRC Handbook (Rumble, 2018)
	-35.5 °C	Physprop (2012)
	-35.5–36 °C	OECD (2002)
Boiling Point	83.4 °C ^b	HSDB (2010) ; CRC Handbook (Rumble, 2018)
	83.5 °C	Physprop (2012)
	83.5–85.1 °C	OECD (2002)
Density	1.2454 at g/cm ³ at 25 °C ^b	HSDB (2010) ; CRC Handbook (Rumble, 2018)
	1.23 g/cm ³ at 20 °C	ATSDR (2001) citing CRC Handbook (Lide, 1998)
Vapor Pressure	78.9 mm Hg at 25 °C ^b	Physprop (2012) ; HSDB (2010) citing Daubert and Danner (1989)
	79.1 mm Hg at 25 °C	ATSDR (2001) citing Daubert et al. (1989)
	79.96 mm Hg at 25 °C	Mackay et al. (2006) citing McGovern (1943)
Vapor Density	3.42 (relative vapor density to air = 1)	NTP, 2016
Water Solubility	8,600 mg/L at 25 °C ^b	Physprop (2012) ; HSDB (2010) citing Horvath et al. (1999)

Property or Endpoint	Value ^a	Reference
	8,690 mg/L at 20 °C ^b	ATSDR (2001) citing Verschueren (1996)
	8,490–9,000 mg/L at 20 °C	OECD (2002)
Log K _{ow}	1.48 ^b	Physprop (2012) ; HSDB (2010) ; Mackay et al. (2006) citing Hansch et al. (1995)
	1.76	WHO IPCS (1995)
	1.45 at 23±1.5 °C	OECD (2002)
Henry's Law Constant	1.18×10^{-3} atm·m ³ /mol at 25 °C ^b	Physprop (2012) citing Leighton and Calo (1981)
	1.1×10^{-3} atm·m ³ /mol at 20 °C	ATSDR (2001) citing Staudinger and Roberts (1996)
Flash Point	13 °C (closed cup); 18 °C (open cup)	ATSDR (2001) citing Budavari et al. (1989); HSDB (2010)
Auto Flammability	413 °C	ATSDR (2001) citing Weiss (1980)
Viscosity	0.829 mPa·second at 20 °C; 0.775 mPa·second at 25 °C	ECHA 2019 citing Mumford and Phillips (1950)
Refractive Index	1.4449 at 20 °C	WHO IPCS (1995) ; HSDB (2010) ; CRC Handbook (Rumble, 2018)
Dielectric Constant	10.45 at 20 °C as liquid; 1.0048 at 120 °C as vapor	HSDB (2010) citing Kirk-Othmer (Snedecor, 2004)

^aMeasured unless otherwise noted

^bSelected value

Table 8. Environmental Fate Characteristics of 1,2-Dichloroethane

Property or Endpoint	Value ^a	Reference
Direct Photodegradation	Not expected to be susceptible to direct photolysis by sunlight because 1,2-dichloroethane does not contain chromophores that absorb at wavelengths >290 nm	HSDB (2010) citing Lyman (1990)
Indirect Photodegradation	$t_{1/2} = 65$ days (based on ·OH reaction rate constant of 2.48×10^{-13} cm ³ /molecule·second at 25 °C)	HSDB (2010) ; Physprop (2012) citing Kwok and Atkinson (1994)
	$t_{1/2} = 73$ days (based on ·OH reaction rate constant of 2.2×10^{-13} cm ³ /molecule·second and an ·OH concentration of 5×10^5 ·OH/cm ³ at 25 °C)	ATSDR (2001) citing Arnts (1989) and Atkinson (1989)
	Atmospheric degradation products: formyl chloride, chloroacetyl chloride, hydrogen chloride, and chloroethanol	ATSDR (2001) citing U.S. EPA (1993)
Hydrolysis	$t_{1/2} = 65$ and 72 years (based on first order rate constant = 2.1×10^{-8} second ⁻¹ and 1.8×10^{-8} second ⁻¹ , respectively, in neutral conditions at 25 °C)	ATSDR (2001) citing Barbash and Reinhard (1989) and Jeffers et al. (1989)

Property or Endpoint	Value ^a	Reference
Biodegradation (Aerobic)	Water: 0%/21 days (modified shake-flask test)	HSDB (2010) citing Mudder and Musterman (1982)
	Water: 20–63%/7 days with 5–27% from volatilization after an unspecified acclimation period (static-flask method)	WHO IPCS (1995) citing Tabak (1981)
	Water: 1.6%/14 days based on BOD 1.1% after 14 days based on GC (Japanese MITI test)	NITE (2010)
	Water: $t_{1/2} = 100$ days	ATSDR (2001) citing Capel and Larsen (1995)
Biodegradation (Anaerobic)	Groundwater: $t_{1/2} = 63–165$ days	NICNAS (2014) citing Lawrence (2006)
	Water: $t_{1/2} = 400$ days	ATSDR (2001) citing Capel and Larsen (1995)
	Sediment: $t_{1/2} = 52$ days based on an observed 0.013/day	ATSDR (2001) citing Peijnenburg et al. (1998)
	Sediment: 0%/35 days	WHO IPCS (1995) citing Jafvert and Wolfe (1987)
Wastewater Treatment	45% total removal (16% by biodegradation, 1% by sludge and 28% by volatilization to air; estimated) ^b	U.S. EPA (2012)
Bioconcentration Factor	2 (<i>Lepomis macrochirus</i>); $t_{1/2} = 2$ days for clearance from tissues	WHO IPCS (1995) citing Barrows (1980)
Bioaccumulation Factor	3.8 (estimated) ^b	U.S. EPA (2012)
Soil Organic Carbon:Water Partition Coefficient (Log K_{OC})	1.28–1.62 ($K_{OC} = 194–2$)	ATSDR (2001) citing Chiou (1980), Sabljic (1995), and Borisover (1997)

^aMeasured unless otherwise noted

^bEPI Suite™ (physical property inputs: Log $K_{OW} = 1.48$, BP = 83.4 °C, MP = -35.6 °C, VP = 78.9 mm Hg, WS = 8600 mg/L, HLC = 1.18×10^{-3} atm-m³/mole), BioP = 120, BioA = 30 and BioS = 30 SMILES: C1CC1

·OH = hydroxyl radical; BOD = biochemical oxygen demand; OECD = Organisation for Economic Co-operation and Development; TG = test guideline; GC = gas chromatography; MITI = Ministry of International Trade and Industry

Results and Discussion

1,2-Dichloroethane is a volatile liquid and a halogenated hydrocarbon that is highly water soluble (8,600 mg/L). The measured Henry's Law constant (1.18×10^{-3} atm-m³/mol) and vapor pressure (78.9 mm Hg) suggest that this chemical will not be persistent in surface water and soil as it will likely volatilize upon release.

In air, 1,2-dichloroethane is not expected to be susceptible to direct photolysis by sunlight due to the absence of chromophores that absorb at wavelengths >290 nm. Additionally, hydrolysis is not expected to be an important fate process for 1,2-dichloroethane with the hydrolysis half-life being between 65 and 72 years. In the vapor phase, 1,2-dichloroethane is expected to react with photochemically produced hydroxyl radicals with a half-life between 65 and 73 days. The

degradation products that can form include formyl chloride, chloroacetyl chloride, hydrogen chloride, and chloroethanol.

In aerobic water using the modified shake-flask test, 0 percent of 1,2-dichloroethane degraded over 21 days. In a static-flask test, 20–63 percent degraded over 7 days, with 5–27 percent lost to volatilization. In anaerobic groundwater, 1,2-dichloroethane degraded with a half-life of 63–165 days. Therefore, this chemical may persist in subsurface environments, groundwater, or enclosed pipes when volatilization is not an option. The measured bioconcentration factor in *Lepomis macrochirus* was 2 with a half-life of 2 days for clearance from tissues and the estimated bioaccumulation factor is 3.8 indicating that 1,2-dichloroethane has low bioaccumulation potential.

6. Storage near significant sources of drinking water

Approach

To support the proposed designation, EPA screened each chemical substance under its conditions of use with respect to the seven criteria in TSCA section 6(b)(1)(A) and 40 CFR 702.9. The statute specifically requires the Agency to consider the chemical substance’s storage near significant sources of drinking water, which EPA interprets as direction to focus on the chemical substance’s potential human health hazard and exposure.

EPA reviewed reasonably available information, specifically looking to identify certain types of existing regulations or protections for the proposed chemical substances. EPA considered the chemical substance’s potential human health hazards, including to potentially exposed or susceptible subpopulations, by identifying existing National Primary Drinking Water Regulations under the Safe Drinking Water Act (40 CFR Part 141) and regulations under the Clean Water Act (CWA) (40 CFR 401.15). In addition, EPA considered the consolidated list of chemical substances subject to reporting requirements under the Emergency Planning and Community Right-to-Know Act (EPCRA; Section 302 Extremely Hazardous Substances and Section 313 Toxic Chemicals), the Comprehensive Environmental Response, Compensation, and Liability Act (CERCLA; Hazardous Substances), and the Clean Air Act (CAA) Section 112(r) (Regulated Chemicals for Accidental Release Prevention). Regulation by one of these authorities is an indication that the substance is a potential health or environmental hazard which, if released near a significant source of drinking water, could present an unreasonable risk of injury to human health or the environment.

Results and Discussion

EPA has established a Maximum Contaminant Level Goal (MCLG) and Maximum Contaminant Level (MCL) for 1,2-dichloroethane due to potential health effects from long-term exposure above the MCL. 1,2-dichloroethane is also a Priority Pollutant under the CWA.

1,2-Dichloroethane is subject to reporting requirements under EPCRA (40 CFR 372.65). It is considered a CERCLA hazardous substance and releases in quantities equal to or greater than 100 pounds are subject to reporting to the National Response Center under CERCLA (40 CFR 302.4).

1,2-Dichloroethane is also subject to the Resource Conservation and Recovery Act [RCRA; hazardous waste numbers D028 (40 CFR 261.24), U077 (40 CFR 261.33), F024, F025 (40 CFR 261.31), K018, K029, K096 (40 CFR 261.32)]. RCRA directs EPA to develop and promulgate criteria for identifying the characteristics of hazardous waste, and for listing hazardous waste, taking into account toxicity, persistence, and degradability in nature, potential for accumulation in tissue and other related factors such as flammability, corrosiveness, and other hazardous characteristics.

One public comment looked at 15 sites that manufacture 1,2-dichloroethane and identified 90 1,2-dichloroethane storage tanks that hold pure or crude 1,2-dichloroethane. The comment states that fifty-seven percent of the tanks were located closest to freshwater rivers or tributaries, while the remaining 43% were located closest to brackish bays or tributaries. The comment also states that it is unclear whether any of these bodies of water (fresh and brackish) are used as drinking water sources. The comment further states that, overall, the identified 1,2-dichloroethane storage tanks were an average of 3,300 and 2,500 feet from freshwater and brackish water, respectively, and ranged from 100 to 8,000 feet between both types of water bodies (EPA-HQ-OPPT-2018-0427-0006).

7. Hazard potential

Approach

EPA considered reasonably available information from peer-reviewed assessments and databases to identify potential human health and environmental hazards for 1,2-dichloroethane (Tables 9 and 10, respectively).

Because, there are very few publicly available assessments for 1,2-dichloroethane with cited environmental hazard data, EPA used the infrastructure of ECOTOXicology knowledgebase (ECOTOX) to identify single chemical toxicity data for aquatic and terrestrial life (U.S. EPA, 2018). It uses a comprehensive chemical-specific literature search of the open literature that is conducted according to the Standard Operating Procedures (SOPs)⁸. The environmental hazard information was populated in ECOTOX and is available to the public. In comparison to the approach used to survey human health hazard data, EPA also used a read-across approach to identify additional environmental hazard data for isomers of 1,2-dichloroethane, if available, to fill in potential data gaps when there were no reported observed effects for specific taxa exposed to the 1,2-dichloroethane (Table 10).

Potential Human Health and Environmental Hazard Tables

EPA identified potential human health and environmental hazards based on a review of the reasonably available information for 1,2-dichloroethane (Tables 9 and 10, respectively).

⁸ The ECOTOX Standard Operating Procedures (SOPs) can be found at: <https://cfpub.epa.gov/ecotox/help.cfm?helptabs=tab4>

Table 9. Potential Human Health Hazards Identified for 1,2-Dichloroethane

Human Health Hazards	Tested for a Specific Effect?	Specific Effect Observed	Data Source
Acute Toxicity	X	X	NICNAS (2013) , U.S. EPA (2010) , OECD (2002) , ATSDR (2001) , RIVM (2001) , EnvCanada (1994)
Repeated Dose Toxicity	X	X	NICNAS (2013) , U.S. EPA (2010) , OECD (2002) , ATSDR (2001) , RIVM (2001) , IARC (1999) , EnvCanada (1994) , NTP (1991) , U.S. EPA (1987) , NCI (1978)
Genetic Toxicity	X	X	NICNAS (2013) , OECD (2002) , RIVM (2001) , ATSDR (2001) , IARC (1999) , NTP (1991)
Reproductive Toxicity	X		NICNAS (2013) , U.S. EPA (2010) , OECD (2002) , ATSDR (2001)
Developmental Toxicity	X		NICNAS (2013) , U.S. EPA (2010) , OECD (2002) , ATSDR (2001)
Toxicokinetic	X	X	NICNAS (2013) , U.S. EPA (2010) , OECD (2002) , ATSDR (2001) , RIVM (2001) , CalEPA (2005) , EnvCanada (1994)
Irritation/Corrosion	X	X	NICNAS (2013) , U.S. EPA (2010) , OECD (2002)
Dermal Sensitization	X		NICNAS (2013)
Respiratory Sensitization			
Carcinogenicity	X	X	NICNAS (2013) , U.S. EPA (2010) , OECD (2002) , ATSDR (2001) , RIVM (2001) , IARC (1999) , U.S. EPA (1987) , NCI (1978)
Immunotoxicology	X	X	OECD (2002) , ATSDR (2001) , EnvCanada (1994) , NTP (1991)
Neurotoxicity	X	X	NICNAS (2013) , U.S. EPA (2010) , OECD (2002) , ATSDR (2001) , CalEPA (2005) , EnvCanada (1994)
Epidemiological Studies or Biomonitoring Studies	X	X	U.S. EPA (2010) , CalEPA (2005) , ATSDR (2001) , IARC (1999)

Note: The “X” in the “Effect Observed” column indicates when a hazard effect was reported by one or more of the referenced studies. Blank rows indicate when information was not identified during EPA’s review of reasonably available information to support the proposed designation.

Table 10. Potential Environmental Hazards Identified for 1,2-Dichloroethane

Media	Study Duration	Taxa Groups	High-Priority Chemical Candidate 1,2-Dichloroethane (CASRN 107-06-2)		Isomers of 1,2-Dichloroethane (CASRN 107-06-2) 1,1-Dichloroethane (CASRN 75-34-3) Dichloroethane (CASRN 1300-21-6)		Data Sources
			Number of Studies	Observed Effects	Number of Studies	Observed Effects	
Aquatic	Acute exposure	Vegetation	1	X	2	X	Tsai and Chen (2007); Wu et al. (2014)
		Invertebrate	12	X	3	X	Call et al. (1980); Foster and Tullis (1984); Foster and Tullis (1985); Freitag et al. (1984); Great Lakes Environment Center (2005); Kramer et al. (1983); LeBlanc (1980); Mayer and Ellersieck (1986); Price et al. (1974); Qureshi et al. (1982); Sanchez-Fortun et al. (1997); Sauvant et al. (1995a); Sauvant et al. (1995b); Sauvant et al. (1995c)
		Fish	5	X	1	X	Buccafusco et al. (1980); Geiger et al. (1985); Great Lakes Environment Center (2005); Heitmuller et al. (1981); Mayer and Ellersieck (1986); Walbridge et al. (1983)
		Non-Fish Vertebrates (i.e., amphibians, reptiles, mammals)	-		-		
	Chronic exposure	Vegetation	-		-		
		Invertebrate	1	X	-		Call et al. (1980)
		Fish	4	X	-		Ahmad et al. (1984); Barrows et al. (1978); Benoit et al. (1982); Black et al. (1982)
		Non-Fish Vertebrates (i.e., amphibians, reptiles, mammals)	2	X	-		Black et al. (1982)
Terrestrial		Vegetation	1	X	1	X	Crebelli (1988)

Media	Study Duration	Taxa Groups	High-Priority Chemical Candidate 1,2-Dichloroethane (CASRN 107-06-2)		Isomers of 1,2-Dichloroethane (CASRN 107-06-2) 1,1-Dichloroethane (CASRN 75-34-3) Dichloroethane (CASRN 1300-21-6)		Data Sources
			Number of Studies	Observed Effects	Number of Studies	Observed Effects	
	Acute exposure	Invertebrate	4	X	-		Bang and Telford (1966); Bhatia and Bansode (1971); Leesch (1984); Neuhauser et al. (1985)
		Vertebrates	4	X	-		Kitchin et al. (1993); Sasaki et al. (1998); Crebelli et al. (1999); Crebelli et al. (1995)
	Chronic exposure	Vegetation	-		3	X	Dietz and Schnoor (2001); Lewis et al. (1979)
		Invertebrate	15	X	-		Bang and Telford (1966); Jefferson (1942) Lindgren (1954); Punj (1970); Shivanandappa and Rajendran (1987)
		Vertebrates	1	X	-		Witt et al. (2000)

The dash indicates that no studies relevant for environmental hazard were identified during the initial review and thus the “Observed Effects” column is left blank. The X in the Observed Effects column indicates when a hazard effect was reported by one or more of the referenced studies. The N/A in the Observed Effects column indicates when a hazard effect was not reported by one of the referenced studies’ abstract (full reference review has not been conducted).

8. Exposure potential

Approach

EPA considered reasonably available information to identify potential environmental, worker/occupational, consumer, and general population exposures to 1,2-dichloroethane.

Release potential for environmental and human health exposure

In addition to other required information, a submission of a TRI Form R report must include the quantities of a TRI chemical the facility released on-site to air, water, or land, and the quantities it transferred off-site to another facility for further waste management. On-site release quantities are reported in Part II Section 5 of the TRI Form R, and off-site transfers are reported in Part II Section 6. Waste management activities include: transfers of a TRI chemical in wastewater to a publicly owned treatment works (POTW) facility or to a non-POTW wastewater treatment facility for the purpose of treatment for destruction or removal; combustion for energy recovery; treatment (treatment includes treatment via incineration for destruction and waste stabilization); recycling; and release, including disposal. During treatment, combustion for energy recovery, or recycling activities, it is possible that some of the quantities of the TRI chemical will be released to the environment.

Worker/Occupational and consumer exposure

EPA's approach for assessing exposure potential was to review the physical and chemical properties, conditions of use reported in CDR, and information from the National Institutes of Health Consumer Product Database and the Chemical and Products Database (CPDat) for 1,2-dichloroethane to inform occupational and consumer exposure potential. The results of this review are detailed in the following tables.

General population exposure

EPA identified environmental concentration, human and environmental biomonitoring data to inform 1,2-dichloroethane's exposure potential to the general population (Table 13).

Results and Discussion

Release potential for environmental and human health exposure

Aggregated quantities of 1,2-dichloroethane released on-site to air, water, and land, and aggregated quantities of 1,2-dichloroethane transferred off-site to POTW and other wastewater treatment facilities (non-POTW) are presented in Table 12 for RY 2011, 2015, and 2017. The table does not include any of the reported quantities pertaining to other waste management activities (e.g., recycling, combustion for destruction) that occurred on-site or off-site during RY 2011, 2015, and 2017. The "Number of Facilities" is the count of unique facilities that filed a TRI Form R report for 1,2-dichloroethane for RY 2011, 2015, and 2017. The TRI data presented were obtained from the TRI dataset following its update in April 2019.

Table 11. The TRI Data on 1,2-Dichloroethane from Reporting Years 2011, 2015, and 2017 Used in this Document to Assess Exposure Potential

Year	Number of Facilities That Reported	Total Quantities Released On-Site to Air (lbs.)	Total Quantities Released On-Site to Water (lbs.)	Total Quantities Released (Disposed of) On-Site to Land (lbs.)	Total Quantities Transferred to POTW (lbs.)	Total Quantities Transferred to Other (non-POTW) Wastewater Treatment Facilities (lbs.)
2011	60	392,609	3,284	44,728	136	1,042
2015	57	415,471	2,554	34,028	263	22,280
2017	55	434,807	2,309	20,996	261	2,464

POTW = publicly owned treatment works

Reference: [U.S. EPA, 2019b](#)

For RY 2017, 55 facilities submitted TRI reports for 1,2-dichloroethane. The total quantities of 1,2-dichloroethane these facilities released on-site to air (as fugitive and stack emissions), surface water and land are: 434,807 pounds; 2,309 pounds; and 20,996 pounds, respectively. These facilities reported 261 pounds of the chemical transferred to POTW and 2,464 pounds transferred off-site to other non-POTW wastewater treatment facilities for the purpose of wastewater treatment. These transfer categories represent two types of off-site transfers for wastewater treatment that may lead to releases from the receiving facilities. They do not include quantities sent off-site for other types of waste management activities that include, or may lead to, releases of the chemical.

Quantities transferred off-site represent the amount of a toxic chemical a facility sent off-site prior to any waste management (e.g., treatment) at a receiving facility. Some of the quantities of 1,2-dichloroethane received by the non-POTW wastewater treatment facilities may have been released to surface waters or to air during treatment processes at the facilities.

1,2-Dichloroethane has a vapor pressure of around 79 mm Hg at 25 °C/77°F. This chemical's vapor pressure indicates potential for air releases from volatilization during manufacturing, processing and use.

When chemical substances are used as reactants and as intermediates, the industrial releases may be a relatively low percentage of the production volume. Lower percentage releases occur when a high percentage of the chemical reacts without excess loss during its use as a reactant or an intermediate. It is unknown the actual percentages, quantities, and media of releases of the reported chemical associated with this processing or use.

Worker/Occupational exposure

Worker exposures to this chemical may be affected by many factors, including but not limited to volume produced, processed, distributed, used and disposed of; physical form and concentration; processes of manufacture, processing, and use; chemical properties such as vapor pressure, solubility, and water partition coefficient; local temperature and humidity; and exposure controls such as engineering controls, administrative controls, and the existence of a personal protective equipment (PPE) program.

1,2-Dichloroethane has an Occupational Safety and Health Administration (OSHA) Permissible Exposure Limit (PEL) ([OSHA, 2009](#)). The PEL is 50 parts per million (ppm) over an 8-hour work day, time weighted average (TWA), with 100 ppm acceptable Ceiling limit and 200 ppm acceptable maximum peak above the acceptable ceiling limit for 5 min in any 3 hours period. This chemical also has a National Institute for Occupational Safety and Health (NIOSH) Recommended Exposure Limit (REL) of 1 ppm TWA, with short term exposure limit (STEL) of 2 ppm ([NIOSH, 2005](#)). The American Conference of Governmental Industrial Hygienists (ACGIH) set the Threshold Limit Value (TLV) at 10 ppm TWA.

1,2-Dichloroethane has a vapor pressure of approximately 79 mm Hg at 25 °C/77 °F. 1,2-Dichloroethane’s vapor pressure indicates the potential for inhalation exposure to vapors generated by the liquid at ambient room temperature conditions.

Consumer exposure

CDR reporting and information from the National Institutes of Health Consumer Product Database do not report on the use of 1,2-dichloroethane in consumer products. However, according to the Chemical and Products Database ([CPDat](#)), 1,2-dichloroethane may be present in consumer products such as fragrances, electronics, plastics, rubbers and solvents (Table 12).

Table 12. Exposure Information for Consumers

Chemical Identity	Consumer Product Database
	Consumer Uses (List)
1,2-Dichloroethane (107-06-2)	Electronics, fragrance, plastic, rubber, solvent

Reference: [CPDat](#)

General population exposure

Releases of 1,2-dichloroethane from certain conditions of use, such as manufacturing, disposal, or waste treatment activities, may result in general population exposures via drinking water ingestion, dermal contact, and inhalation from air releases. 1,2-Dichloroethane has been used in leaded gasoline ([ATSDR 2011](#)).

Table 13. Exposure Information for the Environment and General Population

Database Name	Env. Concen. Data Present?	Human Biomon. Data Present?	Ecological Biomon. Data Present?	Reference
California Air Resources Board	No	no	no	CARB (2005)
Comparative Toxicogenomics Database	Yes	no	no	MDI (2002)
EPA Ambient Monitoring Technology Information Center – Air Toxics Data	Yes	no	no	U.S. EPA (1990)
EPA Discharge Monitoring Report Data	Yes	no	no	U.S. EPA (2007)
EPA Unregulated Contaminant Monitoring Rule	Yes	no	no	U.S. EPA (1996)

Database Name	Env. Concen. Data Present?	Human Biomon. Data Present?	Ecological Biomon. Data Present?	Reference
FDA Total Diet Study	Yes	no	no	FDA (1991)
Great Lakes Environmental Database	Yes	no	no	U.S. EPA (2018)
Information Platform for Chemical Monitoring Data	Yes	no	yes	EC (2018)
International Council for the Exploration of the Sea	No	no	no	ICES (2018)
OECD Monitoring Database	No	yes	no	OECD (2018)
Targeted National Sewage Sludge Survey	No	no	no	U.S. EPA (2006)
The National Health and Nutrition Examination Survey	No	yes	no	CDC (2013)
USGS Monitoring Data –National Water Quality Monitoring Council	Yes	no	no	USGS (1991a)
USGS Monitoring Data –National Water Quality Monitoring Council, Air	Yes	no	no	USGS (1991b)
USGS Monitoring Data –National Water Quality Monitoring Council, Ground Water	Yes	no	no	USGS (1991c)
USGS Monitoring Data –National Water Quality Monitoring Council, Sediment	Yes	no	no	USGS (1991d)
USGS Monitoring Data –National Water Quality Monitoring Council, Soil	Yes	no	no	USGS (1991e)
USGS Monitoring Data –National Water Quality Monitoring Council, Surface Water	Yes	no	no	USGS (1991f)
USGS Monitoring Data –National Water Quality Monitoring Council, Tissue	No	no	yes	USGS (1991g)

^a Concen.= concentration

^b Biomon.= biomonitoring

EPA anticipates releases of 1,2-dichloroethane into the environment because of the conditions of use for 1,2-dichloroethane, particularly activities associated with the chemical's manufacturing. A review of monitoring data collected under EPA rules and statutes (e.g., CAA, CWA, SDWA, National Pollutant Discharge Elimination System), indicate that 1,2-dichloroethane is detected in air, water, soil, vegetation, and other matrices. EPA also found data indicating that 1,2-dichloroethane is used in commercial and industrial products ([U.S. EPA 2014](#)).

1,2-dichloroethane is known to be present in biomonitoring, drinking water, indoor environments, surface water, ambient air, groundwater, and soil ([U.S. EPA 2014](#)). Moreover, it has been reported to be released into the environment at high levels ([U.S. EPA 2014](#)).

1,2-dichloroethane is found in more than 570 hazardous waste sites on the National Priorities List ([ATSDR 2001](#)). An EPA survey found an average of 0.31 ppm 1,2-dichloroethane in

groundwater near hazardous waste sites ([NTP 1993](#)). EPA has found 1,2-dichloroethane to have moderate environmental persistence and a low bioaccumulation potential ([U.S. EPA 2014](#)).

9. Other risk-based criteria that EPA determines to be relevant to the designation of the chemical substance's priority

EPA did not identify other risk-based criteria relevant to the designation of the chemical substance's priority.

10. Proposed Designation and Rationale

Proposed Designation: High-priority substance

Rationale: EPA identified and analyzed reasonably available information for exposure and hazard and is proposing to find that 1,2-dichloroethane may present an unreasonable risk of injury to health and/or the environment, including potentially exposed or susceptible subpopulations (e.g., worker, consumers). This is based on the potential hazard and potential exposure of 1,2-dichloroethane under the conditions of use described in this document to support the prioritization designation. Specifically, EPA expects that the manufacturing, processing, distribution, use and disposal of 1,2-dichloroethane may result in presence of the chemical in surface water and groundwater, ingestion of the chemical in drinking water, inhalation of the chemical from air releases and exposure to workers, exposure to consumers and exposure to the general population, including exposure to children. In addition, EPA identified potential environmental (e.g. aquatic toxicity, terrestrial toxicity) and human health hazards (e.g., acute toxicity, repeated dose toxicity, genetic toxicity, irritation/corrosion, carcinogenicity, immunotoxicity, neurotoxicity, and observations in epidemiological studies).

11. References

Note: All hyperlinked in-text citations are also listed below

Ahmad, N; Benoit, D; Brooke, L; Call, D; Carlson, A; Defoe, D; Huot, J; Moriarity, A; Richter, J; Shubat, P; Veith, G; Wallbridge, C. (1984). Aquatic toxicity tests to characterize the hazard of volatile organic chemicals in water: A toxicity data summary--Parts I and II (pp. 103). (EPA 600/3-84-009). Duluth, MN: U.S. Environmental Protection Agency.

Arnts, RR; Seila, RL; Bufalini, JJ. (1989). Determination of room temperature OH rate constants for acetylene, ethylene dichloride, ethylene dibromide, p-dichlorobenzene and carbon disulfide. *Journal of Air Pollution Control Association* 39: 453-460.
<http://dx.doi.org/10.1080/08940630.1989.10466544>

Atkinson, R. (1989). Kinetics and mechanisms of the gas-phase reactions of the hydroxyl radical with organic compounds. *Journal of Physical and Chemical Reference Data* 1: 1-24

ATSDR (Agency for Toxic Substances and Disease Registry). (2001). Toxicological profile for 1,2-dichloroethane. Atlanta, GA: U.S. Department of Health and Human Services, Public Health Service, Agency for Toxic Substances and Disease Registry.
<https://www.atsdr.cdc.gov/ToxProfiles/tp38.pdf>

ATSDR (Agency for Toxic Substances and Disease Registry). (2011). Toxic substances portal: 1,2-dichloroethane. Available online at
<https://www.atsdr.cdc.gov/substances/toxsubstance.asp?toxid=110>

Bang, YH; Telford, HS. (1966). Effect of sublethal doses of fumigants on stored-grain insects (pp. 22 p.). Pullman, WA: Washington University.

Barbash, JE; Reinhard, M. (1989). Abiotic dehalogenation of 1,2-dichloroethane and 1,2-dibromoethane in aqueous solution containing hydrogen sulfide. *Environmental Science & Technology* 23: 1349-1358. <https://doi.org/10.1021/es00069a004>

Barrows, ME; Petrocelli, SR; Macek, KJ; Carroll, JJ. (1980). Bioconcentration and elimination of selected water pollutants by bluegill sunfish (*Lepomis macrochirus*). In R Haque (Ed.), *Dynamics, exposure and hazard assessment of toxic chemicals* (pp. 379-392). Ann Arbor, MI: Ann Arbor Science.

Benoit, DA; Puglisi, FA; Olson, DL. (1982). A fathead minnow *Pimephales promelas* early life stage toxicity test method evaluation and exposure to four organic chemicals. *Environmental Pollution - Series A: Ecological and Biological* 28: 189-197. [http://dx.doi.org/10.1016/0143-1471\(82\)90075-7](http://dx.doi.org/10.1016/0143-1471(82)90075-7)

Bhatia, SK; Bansode, PC. (1971). Studies on resistance to insecticides in *Tribolium castaneum* (Herbst). IV. Susceptibility of p,p'-DDT-resistant strain to some fumigants. *Indian Journal of Entomology* 33: 45-49.

Black, JA; Birge, WJ; McDonnell, WE; Westerman, AG; Ramey, BA; Bruser, DM. (1982). The aquatic toxicity of organic compounds to embryo-larval stages of fish and amphibians. (Research Report No. 133). Lexington, KY: University of Kentucky.

Borisover, MD; Graber, ER. (1997). Specific interactions of organic compounds with soil organic matter. *Chemosphere* 34: 1761-1776. [http://dx.doi.org/10.1016/S0045-6535\(97\)00032-5](http://dx.doi.org/10.1016/S0045-6535(97)00032-5)

Buccafusco, RJ; Ells, SJ; LeBlanc, GA. (1981). Acute toxicity of priority pollutants to bluegill (*Lepomis macrochirus*). *Bulletin of Environmental Contamination and Toxicology* 26: 446-452. <http://dx.doi.org/10.1007/BF01622118>

Budavari, S; O'Neil, MJ; Smith, A; Heckelman, PE. (1989). *The Merck index: An encyclopedia of chemicals, drugs, and biologicals* (11th ed.). Rahway, NJ: Merck Research Laboratories Division of Merck & Co., Inc.

CARB (California Air Resources Board). (2005). California Air Resources Board (CARB): Indoor air pollution in California [Database]. Retrieved from <https://www.arb.ca.gov/research/apr/reports/13041.pdf>

CalEPA (California Environmental Protection Agency). (2005). Human-exposure-based screening numbers developed to aid estimation of cleanup costs for contaminated soil. California: California Environmental Protection Agency, Office of Environmental Health Hazard Assessment, Integrated Risk Assessment Section. <https://oehha.ca.gov/media/downloads/crn/screenreport010405.pdf>

Call, DJ; Brooke, LT; Ahmad, N. (1980). Toxicity, bioconcentration, and metabolism of selected chemicals in aquatic organisms: Fourth quarterly progress report to EPA (1 January - 31 March 1980) (pp. 80). (U.S. EPA Cooperative Agreement No. CR 806864020).

Capel, PD; Larson, SJ. (1995). A chemodynamic approach for estimating losses of target organic chemicals from water during sample holding time. *Chemosphere* 30: 1097-1107. [http://dx.doi.org/10.1016/0045-6535\(94\)00004-E](http://dx.doi.org/10.1016/0045-6535(94)00004-E)

CDC (Centers for Diseases Control and Prevention). (2013). National Health and Nutrition Examination Survey Data (NHANES) [Database]. Atlanta, GA: CDC, National Center for Health Statistics. Retrieved from <https://www.cdc.gov/nchs/nhanes/index.htm>

Chiou, CT; Freed, VH; Peters, LJ; Kohnert, RL. (1980). Evaporation of solutes from water. *Environment International* 3: 231-236. [http://dx.doi.org/10.1016/0160-4120\(80\)90123-3](http://dx.doi.org/10.1016/0160-4120(80)90123-3)

Crebelli, R; Andreoli, C; Carere, A; Conti, L; Crochi, B; Cotta-Ramusino, M; Benigni, R. (1995). Toxicology of halogenated aliphatic hydrocarbons: Structural and molecular determinants for the disturbance of chromosome segregation and the induction of lipid peroxidation. *Chemico-Biological Interactions* 98: 113-129. [http://dx.doi.org/10.1016/0009-2797\(95\)03639-3](http://dx.doi.org/10.1016/0009-2797(95)03639-3)

Crebelli, R; Benigni, R; Franekic, J; Conti, G; Conti, L; Carere, A. (1988). Induction of chromosome malsegregation by halogenated organic solvents in *Aspergillus nidulans*: Unspecific or specific mechanism? *Mutation Research* 201: 401-411. [http://dx.doi.org/10.1016/0027-5107\(88\)90027-9](http://dx.doi.org/10.1016/0027-5107(88)90027-9)

Crebelli, R; Carere, A; Leopardi, P; Conti, L; Fassio, F; Raiteri, F; Barone, D; Ciliutti, P; Cinelli, S; Vericat, JA. (1999). Evaluation of 10 aliphatic halogenated hydrocarbons in the mouse bone marrow micronucleus test. *Mutagenesis* 14: 207-215. <http://dx.doi.org/10.1093/mutage/14.2.207>

Daubert, TE; Danner, RP. (1989). Physical and thermodynamic properties of pure chemicals: Data compilation. Washington, DC: Taylor & Francis.

Dietz, AC; Schnoor, JL. (2001). Phytotoxicity of chlorinated aliphatics to hybrid poplar (*Populus deltoides* x *nigra* DN34). *Environmental toxicology and chemistry* 20: 389-393. [http://dx.doi.org/10.1897/1551-5028\(2001\)020<0389:POCATH>2.0.CO;2](http://dx.doi.org/10.1897/1551-5028(2001)020<0389:POCATH>2.0.CO;2)

ECHA (European Chemicals Agency). (2019). Registration dossier: 1,2 dichloroethane (107-06-2). Available online at <https://echa.europa.eu/registration-dossier/-/registered-dossier/15430> (accessed January 8, 2018).

EC (European Commission). (2018). Information Platform for Chemical Monitoring Data (IPCHEM) [Database]. Retrieved from <https://ipchem.jrc.ec.europa.eu/RDSIdiscovery/ipchem/index.html>

Environment Canada. (1994). Priority substances list assessment report: 1,2-dichloroethane. Ottawa, Ontario: Government of Canada, Environment Canada, Health Canada. https://www.canada.ca/content/dam/hc-sc/migration/hc-sc/ewh-semt/alt_formats/hecs-sesc/pdf/pubs/contaminants/psl1-lsp1/1_2_dichloroethane/1_2_dichloroethane-eng.pdf

FDA (U.S. Food and Drug Administration). (1991). FDA Total Diet Study [Database]. Retrieved from <http://www.fda.gov/Food/FoodScienceResearch/TotalDietStudy/ucm184293.htm>

Foster, G; Tullis, RE. (1984). A quantitative structure-activity relationship between partition coefficients and the acute toxicity of naphthalene derivatives in *Artemia salina* nauplii. *Aquatic Toxicology* AMST: 245-254. [http://dx.doi.org/10.1016/0166-445X\(84\)90023-7](http://dx.doi.org/10.1016/0166-445X(84)90023-7)

Foster, GD; Tullis, RE. (1985). Quantitative structure-toxicity relationships with osmotically stressed *Artemia salina* nauplii. *Environmental Pollution Series A, Ecological and Biological* 38: 273-281. <http://www.sciencedirect.com/science/article/pii/0143147185901321>

Freitag, D; Ballhorn, L; Behechti, A; Fischer, K; Thumm, W. (1994). Structural configuration and toxicity of chlorinated alkanes. *Chemosphere* 28: 253-259. [http://dx.doi.org/10.1016/0045-6535\(94\)90122-8](http://dx.doi.org/10.1016/0045-6535(94)90122-8)

Geiger, DL; Northcott, CE; Call, DJ; Brooke, LT. (1985). Acute toxicities of organic chemicals to fathead minnows (*Pimephales promelas*): Volume II. Superior, WI: University of Wisconsin-Superior, Center for Lake Superior Environmental Studies.

Great Lakes Environment Center. (2005). Final Report on Acute Toxicity of Selected Chemicals in Support of the Great Lakes Water Quality Guidance Final Rep.Gt.Lakes Environ.Ctr., Traverse City, MI:35.

Hansch, C; Leo, A; Hoekman, D. (1995). Exploring QSAR: Hydrophobic, electronic, and steric constants. In C Hansch; A Leo; DH Hoekman (Eds.), ACS Professional Reference Book. Washington, DC: American Chemical Society.

Heitmuller, PT; Hollister, TA; Parrish, PR. (1981). Acute toxicity of 54 industrial chemicals to sheepshead minnows (*Cyprinodon variegatus*). *Bulletin of Environmental Contamination and Toxicology* 27: 596-604. <http://dx.doi.org/10.1007/BF01611069>

Horvath, AL; Getzen, FW; Maczynska, Z. (1999). IUPAC-NIST solubility data series 67: Halogenated ethanes and ethenes with water. *Journal of Physical and Chemical Reference Data* 28: 395-627. <http://dx.doi.org/10.1063/1.556039>

HSDB (Hazardous Substances Data Bank). (2010). 1,2-dichloroethane (CASRN: 107-06-2). Available online at <https://toxnet.nlm.nih.gov/cgi-bin/sis/search/a?dbs+hsdb:@term+@DOCNO+65>

IARC (International Agency for Research on Cancer). (1999). Re-evaluation of some organic chemicals, hydrazine and hydrogen peroxide. In IARC Monographs on the Evaluation of Carcinogenic Risks to Humans. Lyon, France: World Health Organization. <https://monographs.iarc.fr/wp-content/uploads/2018/06/mono71.pdf>

ICES (International Council for the Exploration of the Sea). (2018). ICES-Dome [Database]. Retrieved from <http://www.ices.dk/marine-data/data-portals/Pages/DOME.aspx>

Jafvert, CT; Lee Wolfe, N. (1987). Degradation of selected halogenated ethanes in anoxic sediment-water systems. *Environmental toxicology and chemistry* 6: 827-837.

Jeffers, PM; Ward, LM; Woytowitch, LM; Wolfe, NL. (1989). Homogeneous hydrolysis rate constants for selected chlorinated methanes, ethanes, ethenes, and propanes. *Environmental Science and Technology* 23: 965-969. <http://dx.doi.org/10.1021/es00066a006>

Jefferson, RN. (1942) The influence of carbon tetrachloride on the toxic efficiency of certain volatile organic compounds. (Doctoral Dissertation). Iowa State University, Ames, IA.

Kitchin, KT; Brown, JL; Kulkarni, AP. (1993). Predicting rodent carcinogenicity of halogenated hydrocarbons by in vivo biochemical parameters. *Birth Defects Research, Part B: Developmental and Reproductive Toxicology* 13: 167-184.

Kramer, VC; Schnell, DJ; Nickerson, KW. (1983). Relative toxicity of organic solvents to *Aedes aegypti* larvae. *Journal of Invertebrate Pathology* 42: 285-287. [http://dx.doi.org/10.1016/0022-2011\(83\)90076-9](http://dx.doi.org/10.1016/0022-2011(83)90076-9)

Kwok, ESC; Atkinson, R. (1994). Estimation of hydroxyl radical reaction rate constants for gas-phase organic compounds using a structure-reactivity relationship: An update. (CMA Contract No. ARC-8.0-OR). Riverside, CA: University of California.

Lawrence, S. (2006). Description, properties, and degradation of selected volatile organic compounds detected in ground water - a review of selected literature. Atlanta, GA: United States Geological Survey. <http://pubs.usgs.gov/>

LeBlanc, GA. (1980). Acute toxicity of priority pollutants to water flea (*Daphnia magna*). *Bulletin of Environmental Contamination and Toxicology* 24: 684-691. <http://dx.doi.org/10.1007/BF01608174>

Leesch, JG. (1984). Fumigation of lettuce: Efficacy and phytotoxicity. *Journal of Economic Entomology* 77: 142-150.

Leighton, DT, Jr; Calo, JM. (1981). Distribution coefficients of chlorinated hydrocarbons in dilute air-water systems for groundwater contamination applications. *Journal of Chemical and Engineering Data* 26: 382-585. <http://dx.doi.org/10.1021/jc00026a010>

Lewis, JA; Papavizas, GC; O'Neill, NR. (1979). Effect of seed immersion in organic solvents on germinability. *Journal of Agricultural Science* 92: 563-570. <http://dx.doi.org/10.1017/S0021859600053806>

Lide, DR. (1998). *CRC handbook of chemistry and physics: A ready-reference book of chemical and physical data* (79th ed.). Boca Raton, FL: CRC Press.

indgren, D; Vincent, L; Krohne, H. (1954). Relative effectiveness of ten fumigants to adults of eight species of stored-product insects. *Journal of Economic Entomology* 47: 923-926.

Lyman, WJ; Reehl, WF; Rosenblatt, DH. (1990). Handbook of chemical property estimation methods: Environmental behavior of organic compounds. In WJ Lyman; WF Reehl; DH Rosenblatt (Eds.). Washington, DC: American Chemical Society.

Mackay, D; Shiu, WY; Ma, KC; Lee, SC. (2006). Handbook of physical-chemical properties and environmental fate for organic chemicals. Boca Raton, FL: CRC Press.

Mayer, FL; Ellersieck, MR. (1986). Manual of acute toxicity: Interpretation and data base for 410 chemicals and 66 species of freshwater animals. Washington, DC: U.S. Department of the Interior, Fish and Wildlife Service.

McGovern, EW. (1943). Chlorohydrocarbon solvents. The Journal of Industrial and Engineering Chemistry 35: 1231-1239.

MDI (MDI Biological Laboratory). (2002). Comparative Toxicogenomics Database (CTD) [Database]. Retrieved from <http://ctdbase.org>

Mudder, TI; Musterman, JL. (1982). Development of empirical structure biodegradability relationships and biodegradability testing protocol for volatile and slightly soluble priority pollutants. In Abstracts of Papers of the American Chemical Society. Kansas City, MO: ACS.

Mumford, SA; Phillips, JWC. (1950). The physical properties of some aliphatic compounds [10.1039/JR9500000075]. Journal of the Chemical Society (Resumed)75-84.
<http://dx.doi.org/10.1039/JR9500000075>

NCI (National Cancer Institute). (1978). Bioassay of 1,2-dichloroethane for possible carcinogenicity, CAS no. 107-06-2. (NCI-CG-TR-55). Bethesda, MD: National Institutes of Health, National Cancer Institute, Division of Cancer Cause and Prevention.
https://ntp.niehs.nih.gov/ntp/htdocs/lt_rpts/tr055.pdf

Neuhauser, EF; Loehr, RC; Malecki, MR; Milligan, DL; Durkin, PR. (1985). The toxicity of selected organic chemicals to the earthworm *Eisenia fetida*. Journal of Environmental Quality 14: 383-388.

NICNAS (National Industrial Chemicals Notification and Assessment Scheme). (2013). Ethane, 1,2-dichloro-: Human health tier II assessment. Sydney, Australia: Australian Department of Health, National Industrial Chemicals Notification and Assessment Scheme.
https://www.nicnas.gov.au/chemical-information/imap-assessments/imap-assessment-details?assessment_id=72#cas-A_107-06-2

NICNAS (National Industrial Chemicals Notification and Assessment Scheme). (2014). Ethane, 1,2-dichloro-: Environment tier II assessment. Sydney, Australia: Australian Department of Health, National Industrial Chemicals Notification and Assessment Scheme.
<https://www.nicnas.gov.au/chemical-information/imap-assessments/imap-assessments/tier-ii-environment-assessments/1,2-dichloroethane>

NITE (National Institute of Technology and Evaluation). (2010). Japan CHEMicals Collaborative Knowledge database (J-CHECK). CASRN: 107-06-2. Available online at https://www.nite.go.jp/chem/jcheck/detail.action?cno=107-06-2&mno=2-0054&request_locale=en (accessed July 5, 2018).

NTP (National Toxicology Program). (1991). NTP technical report on the toxicity studies of 1,2-dichloroethane (ethylene bichloride) in F344/N rats, Sprague Dawley rats, Osborne-Mendel rats, and B6C3F1 mice (drinking water and gavage studies) (NTP TOX 4; NIH Publication No. 91-3123). Research Triangle Park, NC: U.S. Department of Health and Human Services, National Institutes of Health, National Toxicology Program.
https://ntp.niehs.nih.gov/ntp/htdocs/st_rpts/tox004.pdf

NTP (National Toxicology Program). (1993). NTP technical report on the toxicity studies of a chemical mixture of 25 groundwater contaminants administered in drinking water to F344/N rats and B6C3F1 mice (NTP TR 35; NIH Publication 93-3384). Research Triangle Park, NC: U.S. Department of Health and Human Services, National Institutes of Health, National Toxicology Program. https://ntp.niehs.nih.gov/ntp/htdocs/st_rpts/tox35.pdf

NTP (National Toxicology Program). (2016). Report on Carcinogens, Fourteenth Edition: 1,2-dichloroethane. Research Triangle Park, NC: U.S. Department of Health and Human Services, National Institutes of Health, National Toxicology Program.
<https://ntp.niehs.nih.gov/ntp/roc/content/profiles/dichloroethane.pdf>

NTP (National Toxicology Program). (2016). Report on carcinogens, fourteenth edition: 1,2-dichloroethane. Research Triangle Park, NC: U.S. Department of Health and Human Services, National Institutes of Health, National Toxicology Program.
<https://ntp.niehs.nih.gov/ntp/roc/content/profiles/dichloroethane.pdf>

OECD (Organisation for Economic Co-operation and Development). (2002). SIDS initial assessment report: 1,2-dichloroethane (pp. 203). Paris, France.
<https://hpvchemicals.oecd.org/ui/handler.axd?id=95F8D194-732A-4CC9-B59B-839ED3B18732>

OECD (Organisation for Economic Co-operation and Development). (2018). OECD Monitoring Database [Database].

Peijnenburg, W; Eriksson, L; De Groot, A; Sjöström, M; Verboom, H. (1998). The kinetics of reductive dehalogenation of a set of halogenated aliphatic hydrocarbons in anaerobic sediment

slurries. *Environmental Science and Pollution Research* 5: 12-16.

<http://dx.doi.org/10.1007/BF02986368>

Price, KS; Waggy, GT; Conway, RA. (1974). Brine shrimp bioassay and seawater BOD of petrochemicals. *Water Environment and Technology* 46: 63-77.

Punj, GK. (1970). The effect of nutrition on the susceptibility of larvae of *Trogoderma granarium* Everts (Coleoptera, Dermestidae) to certain fumigants. *Journal of Stored Products Research* 6: 181-185.

Qureshi, A; Flood, K; Thompson, S; Janhurst, S; Inniss, C; Rokosh, D. (1982). Comparison of a luminescent bacterial test with other bioassays for determining toxicity of pure compounds and complex effluents. In *Aquatic Toxicology and Hazard Assessment*. West Conshohocken, PA: ASTM International.

RIVM (National Institute for Public Health and the Environment - Netherlands). (2001). Re-evaluation of human-toxicological maximum permissible risk levels (pp. 297). (711701025). Bilthoven, Netherlands: National Institute for Public Health and the Environment (RIVM).
<https://www.rivm.nl/bibliotheek/rapporten/711701025.pdf>

Rumble, JR. (2018). CRC handbook of chemistry and physics. In JR Rumble (Ed.), (99th ed.). Boca Raton, FL: CRC Press.

Sabljić, A; Güsten, H; Verhaar, H; Hermens, J. (1995). QSAR modelling of soil sorption. Improvements and systematics of log KOC vs. log KOW correlations. *Chemosphere* 31: 4489-4514. <http://www.sciencedirect.com/science/article/pii/0045653595003275>

Sanchez-Fortun, S; Sanz, F; Santa-Maria, A; Ros, JM; De Vicente, ML; Encinas, MT; Vinagre, E; Barahona, MV. (1997). Acute sensitivity of three age classes of *Artemia salina* larvae to seven chlorinated solvents. *Bulletin of Environmental Contamination and Toxicology* 59: 445-451.
<http://dx.doi.org/10.1007/s001289900498>

Sasaki, YF; Saga, A; Akasaka, M; Ishibasi, S; Yoshida, K; Su, QY; Matsusaka, N; Tsuda, S. (1998). Detection of in vivo genotoxicity of haloalkanes and haloalkenes carcinogenic to rodents by the alkaline single cell gel electrophoresis (comet) assay in multiple mouse organs. *Mutation Research: Genetic Toxicology and Environmental Mutagenesis* 419: 13-20.
[http://dx.doi.org/10.1016/S1383-5718\(98\)00114-4](http://dx.doi.org/10.1016/S1383-5718(98)00114-4)

Sauvant, MP; Pepin, D; Bohatier, J; Groliere, CA. (1995a). Comparison of six bioassays for assessing in vitro acute toxicity and structure-activity relationships for vinyl chloride monomer, its main metabolites and derivatives. *Science of the Total Environment* 172: 79-92.
[http://dx.doi.org/10.1016/0048-9697\(95\)04782-4](http://dx.doi.org/10.1016/0048-9697(95)04782-4)

Sauvant, MP; Pepin, D; Bohatier, J; Groliere, CA. (1995b). Microplate technique for screening and assessing cytotoxicity of xenobiotics with *Tetrahymena pyriformis*. *Ecotoxicology and Environmental Safety* 32: 159-165.

<http://www.sciencedirect.com/science/article/pii/S0147651385710974>

Sauvant, MP; Pépin, D; Grolière, CA; Bohatier, J. (1995c). Effects of organic and inorganic substances on the cell proliferation of L-929 fibroblasts and *Tetrahymena pyriformis* GL protozoa used for toxicological bioassays. *Bulletin of Environmental Contamination and Toxicology* 55: 171-178. <http://dx.doi.org/10.1007/BF00203006>

Shivanandappa, T; Rajendran, S. (1987). Induction of glutathione S-transferase by fumigants in larvae of the Khapra beetle, *Trogoderma granarium* (E.). *Pesticide Biochemistry and Physiology* 28: 121-126.

Snedecor, G; Hickman, JC; Mertens, J, a. (2004). Chloroethylenes. In Kirk-Othmer Encyclopedia of Chemical Technology. [online]: John Wiley & Sons, Inc.

Staudinger, J; Roberts, PV. (1996). A critical review of Henry's law constants for environmental applications. *Critical Reviews in Environmental Science and Technology* 26: 205-297.

Syracuse Research Corp. (1978). Results of continuous exposure of fathead minnow embryo to 21 priority pollutants (pp. 47). (EPA/OTS 40-7848049). Springfield, VA.

Tabak, HH; Quave, SA; Mashni, CI; Barth, EF. (1981). Biodegradability studies with organic priority pollutant compounds. *Journal of Water Pollution Control Federation* 53: 1503-1518.

Tsai, KP; Chen, CY. (2007). An algal toxicity database of organic toxicants derived by a closed-system technique. *Environmental toxicology and chemistry* 26: 1931-1939.

<http://dx.doi.org/10.1897/06-612R.1>

U.S. EPA (U.S. Environmental Protection Agency). (1987). Integrated Risk Information System (IRIS) chemical assessment summary: 1,2-dichloroethane; CASRN: 107-06-2. Washington, DC: U.S. Environmental Protection Agency, National Center for Environmental Assessment.

https://cfpub.epa.gov/ncea/iris/iris_documents/documents/subst/0149_summary.pdf

U.S. EPA (U.S. Environmental Protection Agency). (1990). EPA Ambient Monitoring Technology Information Center (AMTIC): Air toxics data [Database]. Retrieved from

<http://www3.epa.gov/ttnamti1/toxdat.html>

U.S. EPA (U.S. Environmental Protection Agency). (1993). A literature review of atmospheric transformation products of Clean Air Act Title III Hazardous Air Pollutants. (600R94088). Research Triangle Park, NC: U.S. Environmental Protection Agency.

U.S. EPA (U.S. Environmental Protection Agency). (1996). EPA Unregulated Contaminant Monitoring Rule (UCMR) [Database]. Retrieved from <https://www.epa.gov/dwucmr>

U.S. EPA (U.S. Environmental Protection Agency). (2000). Memorandum within U.S. EPA concerning a fathead minnow early life stage toxicity test with attachment (pp. 13). (EPA/OTS Doc.#40-8247020).

U.S. EPA (U.S. Environmental Protection Agency). (2006). Targeted National Sewage Sludge Survey (TNSSS) [Database]. Retrieved from <https://www.epa.gov/biosolids/sewage-sludge-surveys>

U.S. EPA (U.S. Environmental Protection Agency). (2007). EPA Discharge Monitoring Report Data (EPA DMR) [Database]. Retrieved from <https://cfpub.epa.gov/dmr/>

U.S. EPA (U.S. Environmental Protection Agency). (2010). Provisional peer-review toxicity values for 1,2-dichloroethane (CASRN 107-06-2). Cincinnati, OH: U.S. Environmental Protection Agency, National Center for Environmental Assessment, Superfund Health Risk Technical Support Center. https://hhpprtv.ornl.gov/issue_papers/Dichloroethane12.pdf

U.S. EPA (U.S. Environmental Protection Agency). (2012). Estimation Programs Interface Suite for Microsoft Windows, v. 4.11. Washington, DC: U.S. Environmental Protection Agency. <https://www.epa.gov/tsca-screening-tools/download-epi-suitetm-estimation-program-interface-v411>

U.S. EPA (U.S. Environmental Protection Agency). (2012). Non-Confidential 2012 Chemical Data Reporting (CDR) (August 18, 2014 ed.). Environmental Protection Agency, Chemical Data Reporting. <http://www.epa.gov/cdr/>

U.S. EPA (U.S. Environmental Protection Agency) (2012). TSCA Work Plan Chemicals: Methods Document. Washington, DC. U.S. Environmental Protection Agency, Office of Pollution Prevention and Toxics. February 2012. <https://www.epa.gov/assessing-and-managing-chemicals-under-tsca/tsca-work-plan-methods-document>

U.S. EPA (U.S. Environmental Protection Agency) (2013). 1986-2002 Inventory Update Reporting rule data (Non-confidential Production Volume in Pounds. Washington, DC. U.S. Environmental Protection Agency, Office of Pollution Prevention and Toxics. Retrieved: August 9, 2013.

U.S. EPA (U.S. Environmental Protection Agency). (2012). PhysProp database. Estimation Programs Interface Suite™ for Microsoft® Windows, v 4.11: CASRN 107-06-2 [Fact Sheet]. Washington, DC: U.S. Environmental Protection Agency. <https://www.epa.gov/tsca-screening-tools/epi-suitetm-estimation-program-interface>

U.S. EPA (U.S. Environmental Protection Agency). (2014). TSCA work plan for chemical assessments: 2014 update. Washington, D.C.: U.S. Environmental Protection Agency, Office of Pollution Prevention and Toxics. https://www.epa.gov/sites/production/files/2015-01/documents/tsca_work_plan_chemicals_2014_update-final.pdf

U.S. EPA (U.S. Environmental Protection Agency) (2019a). Chemical Data Reporting (2012 and 2016 CBI CDR database). Washington, DC. U.S. Environmental Protection Agency, Office of Pollution Prevention and Toxics. Retrieved: April 25, 2019.

U.S. EPA (U.S. Environmental Protection Agency). (2019b). Chemical and Products Database (CPDat). Available online at <https://www.epa.gov/chemical-research/chemical-and-products-database-cpdat>

U.S. EPA (U.S. Environmental Protection Agency). (2019c). Envirofacts Toxics Release Inventory 2017 Updated Dataset (released April 2019) <https://www.epa.gov/enviro/tri-customized-search>

USGS (U.S. Geological Survey). (1991a). USGS Monitoring Data: National Water Quality Monitoring Council [Database]. Retrieved from <https://www.waterqualitydata.us/portal>

USGS (U.S. Geological Survey). (1991b). USGS Monitoring Data: National Water Quality Monitoring Council - Air [Database]. Retrieved from <https://www.waterqualitydata.us/portal/#sampleMedia=Air&mimeType=csv>

USGS (U.S. Geological Survey). (1991c). USGS Monitoring Data: National Water Quality Monitoring Council - Groundwater [Database]. Retrieved from <https://www.waterqualitydata.us/portal/#siteType=Aggregate%20groundwater%20use&sampleMedia=Water&mimeType=csv&dataProfile=activityAll>

USGS (U.S. Geological Survey). (1991d). USGS Monitoring Data: National Water Quality Monitoring Council - Sediment [Database]. Retrieved from <https://www.waterqualitydata.us/portal/#sampleMedia=Sediment&mimeType=csv>

USGS (U.S. Geological Survey). (1991e). USGS Monitoring Data: National Water Quality Monitoring Council - Soil [Database]. Retrieved from <https://www.waterqualitydata.us/portal/#sampleMedia=Soil&mimeType=csv>

USGS (U.S. Geological Survey). (1991f). USGS Monitoring Data: National Water Quality Monitoring Council - Surface Water [Database]. Retrieved from <https://www.waterqualitydata.us/portal/#siteType=Aggregate%20surface-water-use&sampleMedia=Water&mimeType=csv>

USGS (U.S. Geological Survey). (1991g). USGS Monitoring Data: National Water Quality Monitoring Council - Tissue [Database]. Retrieved from <https://www.waterqualitydata.us/portal/#sampleMedia=Tissue&mimeType=csv>

USGS (U.S. Geological Survey). (2006). Description, properties, and degradation of selected volatile organic compounds detected in ground water - a review of selected literature. Atlanta, GA: U.S. Department of the Interior, U.S. Geological Survey. <https://pubs.usgs.gov/of/2006/1338/pdf/ofr2006-1338.pdf>

Verschueren, K. (1996). Handbook of environmental data on organic chemicals (3rd ed.). New York, NY: Van Nostrand Reinhold Company.

Walbridge, CT; Fiantt, JT; Phipps, GL; Holcombe, GW. (1983). Acute toxicity of ten chlorinated aliphatic hydrocarbons to the fathead minnow (*Pimephales promelas*). Archives of Environmental Contamination and Toxicology 12: 661-666.

Weiss, G. (1980). Hazardous chemicals data book. Park Ridge, NJ: Noyes Data Corporation.

WHO IPCS (World Health Organization International Programme on Chemical Safety). (1995). Environmental health criteria 176: 1,2-Dichloroethane. World Health Organization, International Programme on Chemical Safety. <http://www.inchem.org/documents/ehc/ehc/ehc176.htm>

Witt, KL; Knapton, A; Wehr, CM; Hook, GJ; Mirsalis, J; Shelby, MD; MacGregor, JT. (2000). Micronucleated erythrocyte frequency in peripheral blood of B6C3F(1) mice from short-term, prechronic, and chronic studies of the NTP carcinogenesis bioassay program. Environmental and Molecular Mutagenesis 36: 163-194. [http://dx.doi.org/10.1002/1098-2280\(2000\)36:3<163::AID-EM1>3.0.CO;2-P](http://dx.doi.org/10.1002/1098-2280(2000)36:3<163::AID-EM1>3.0.CO;2-P)

Woodward, D; Mauck, W. (1980). Toxicity of five forest insecticides to cutthroat trout and two species of aquatic invertebrates. Bulletin of Environmental Contamination and Toxicology 25: 846-854.

Wu, S; Zhang, H; Yu, X; Qiu, L. (2014). Toxicological responses of *Chlorella vulgaris* to dichloromethane and dichloroethane. Environmental Engineering Science 31: 9-17. <http://dx.doi.org/10.1089/ees.2013.0038>

Zeiger, E. (1987). Carcinogenicity of mutagens: Predictive capability of the Salmonella mutagenesis assay for rodent carcinogenicity. Cancer Research 47: 1287-1296.