



EPA Tools & Resources Training Webinar: PFAS Removal in Drinking Water Treatment Systems

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Center for Environmental Solutions and Emergency Response US EPA Office of Research and Development **EPA Contemporal Stronger Future**

August 6, 2020



- Background of EPA PFAS (per- and polyfluoroalkyl substances) Treatment Research
- EPA Models
 - Treatability database
 - Performance models
 - Cost models
- Demonstration of Performance Models
 - How to download
 - Example runs



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EPA's PFAS Drinking Water Research

Problem: Utilities lack treatment technology cost data for PFAS removal **Actions**:

- Gather performance and cost data from available sources (DOD, utilities, industry, etc.)
- Conduct EPA research on performance of treatment technologies including home treatment systems
- Update EPA's Treatability Database, Treatment Models and Unit Cost Models
- Connect EPA's Treatability Database to EPA's Unit Cost Models for ease of operation
- Model performance and cost, and then extrapolate to other scenarios
- Address treatment impact on corrosion
- Evaluate reactivation of granular activated carbon

Impact: Enable utilities to make informed decisions about cost-effective treatment strategies for removing PFAS from drinking water

Model Scenarios

- Variable source waters
- Variable PFAS concentrations in source water
- Alternate treatment goals
- Changing production rates
- Different reactivation/disposal options
- Document secondary benefits





To provide tools to accurately predict the performance and cost of treating PFAS in drinking waters



Suite of Models

Compatibility

Suite of Models

ETDOT was developed by National Center for Clean Industrial and Treatment Technologies at Michigan Technological University (MTU). In 2019, EPA signed an agreement with MTU to make this suite of adsorption models available to the public at no cost.

Related FPA Resources

Applications



Treatment Information

Publicly Available Drinking Water Treatability Database

Interactive literature review database that contains 123 regulated and unregulated contaminants and covers 35 treatment processes commonly employed or known to be effective (thousands of sources assembled on one site)

Currently available:

PFOA, PFOS, PFTriA, PFDoA, PFUnA, PFDA, PFNA, PFHpA, PFHxA, PFPeA, PFBA, PFDS, PFHpS, PFHxS, PFBA, PFBS, PFOSA, FtS 8:2, FtS 6:2, N-EtFOSAA, N-MeFOSAA and GenX

epa.gov/water-research/drinking-water-treatability-database-tdb Search: EPA TDB



Contains treatment information to be used in performance or cost models





Treatability Database

Agency Landing Page





epa.gov/water-research/drinking-water-treatability-database-tdb





PFAS Treatment

Per- and Polyfluoroalkyl Substances



CAS Number:

Synonyms:

Heptafluoropropyl 1,2,2,2-tetrafluoroethyl ether (E1),2,3,3,3-tetrafluoro-2-(heptafluoropropoxy)propanoate (FRD-902),2,3,3,3-tetrafluoro-2-(heptafluoropropoxy)propanoic acid (FRD-903),Ammonium perfluoro-2-methyl-3-oxahexanoate (GenX),Heptadecafluorononanoic acid,Heptafluorobutyric acid,Nonadecafluorocapric acid,Nonadecafluorodecanoic acid,Pefluorobutane sulfonate,Perfluorobutyric acid,Perfluorobexanesulfonic acid potassium salt,Potassium tridecafluoro-1-hexanesulfonate,Tridecafluorohexane-1-sulfonic acid potassium salt

Contaminant Type: Chemical

Description:

Per- and polyfluoroalkyl substances (PFASs) are fluorinated aliphatic substances with unique properties, such as being both hydrophobic, lipophobic, and extremely stable due to the strength of the C-F bond [2539]. Their properties have led to their extensive use as surface active agents in products like stain repellants and fire-fighting foams [2527, 2539]. The two most frequently studied PFASs, perfluorooctane sulphonate (PFOS) and perfluorooctanoic acid (PFOA), have their own, separate entries in this treatability database. Both PFOS and PFOA are compounds with eight carbon atoms. This group entry covers

Per- and Polyfluoroalkyl Substances



The following processes were found to be effective for the removal of PFASs: granular activated carbon (GAC) (up to > 98 percent), membrane separation (up to > 99 percent), and ion exchange (up to > 99 percent). These results cover the removal of specific PFASs including PFTriA, PFDoA, PFUA, PFNA, PFHPA, PFHPA, PFPAA, PFPAA, PFPAA, PFPAB, PFBS, PFPS, PFOSA, PFNOAA, PFO3OA, PFO2HXA, FtS 8:2, FtS 6:2, N-EtFOSAA, N-MeFOSAA, and GenX. For results on the removal of PFOS and PFOA, see the separate treatability database entries for those specific contaminants.

Studies were identified evaluating the following treatment technologies for the removal of PFASs:

Adsorptive Media

A bench-scale study conducted batch tests of adsorption using magnetic nanoparticles with different polymer coatings. In ultrapure water, the best performing of these achieved high removals of long chain and sulfonated PFASs (e.g., >90 percent ...

See more

Aeration and Air Stripping

At a full-scale site, packed tower aeration was not effective for removing PFASs [2441].

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PFAS Treatment: Activated Carbon

Matrix of conditions and results from treatment references that can be downloaded into a spreadsheet

				Removal	Contaminant	Contaminant	Contamin	Contamin		Design				Manufact	Product
Ref #	Author	Year	Log or Percent Removal	Туре	Influent	Effluent	ant Units	ant	Scale	Flow	Water	Location Studied	GAC Type	urer	Name
2441	Dickenson,	2016	-10.5 to 13.7#	Percent	4.4 to 5.1#	5.7 to 6.3#	ng/L	PFHpA	F	5	SW	New Jersey	В	Calgon	F300
2441	Dickenson,	2016	-11 to 5#	Percent	3.6 to 5.8#	4.0 to 5.5#	ng/L	PFHxS	F	5	SW	New Jersey	В	Calgon	F300
2441	Dickenson,	2016	-13 to 6#	Percent	1.8 to 2.4#	1.7 to 2.7#	ng/L	PFNA	F	5	SW	New Jersey	В	Calgon	F300
2441	Dickenson,	2016	-19 to 10#	Percent	6.8 to 7.3#	6.1 to 8.7#	ng/L	PFHxA	F	5	SW	New Jersey	В	Calgon	F300
2441	Dickenson,	2016	-26#	Percent	<5.0#	6.3#	ng/L	PFBA	F	5	SW	New Jersey	В	Calgon	F300
2441	Dickenson,	2016	-34 to 8#	Percent	0.59 to 0.97#	0.54 to 1.3#	ng/L	PFDA	F	5	SW	New Jersey	В	Calgon	F300
2441	Dickenson,	2016	-66 to 70#	Percent	1.23 to 1.81#	0.537 to 2.48#	ng/L	PFBA	F	0.5472 to	GW	Minnesota	В	Calgon	F600
2441	Dickenson,	2016	0 to 19#	Percent	<0.05 to 0.085	<0.05 to 0.069#	ng/L	PFPeA	F	0.5472 to	GW	Minnesota	В	Calgon	F600
2441	Dickenson,	2016	0 to 76#	Percent	<0.05 to 0.210	<0.05#	ng/L	PFHxS	F	0.5472 to	GW	Minnesota	В	Calgon	F600
2441	Dickenson,	2016	33#	Percent	15#	10#	ng/L	PFBA	F	5#	SW	Colorado	В	Norit	GAC 300
2441	Dickenson,	2016	46 to 60#	Percent	0.127 to 0.192	<0.05 to 0.1023	ng/L	PFHxA	F	0.5472 to	GW	Minnesota	В	Calgon	F600
2441	Dickenson,	2016	5 to 6#	Percent	2.1 to 3.6#	2.0 to 3.4#	ng/L	PFBS	F	5	SW	New Jersey	В	Calgon	F300
2441	Dickenson,	2016	7.2 to 12.7#	Percent	4.8 to 5.5#	6.4 to 6.9#	ng/L	PFPeA	F	5	SW	New Jersey	В	Calgon	F300
2441	Dickenson,	2016	74#	Percent	17#	4.4#	ng/L	PFPeA	F	5#	SW	Colorado	В	Norit	GAC 300
2441	Dickenson,	2016	91#	Percent	11#	0.97#	ng/L	PFNA	F	5#	SW	Colorado	В	Norit	GAC 300
2441	Dickenson,	2016	>89#	Percent	4.5#	<0.50#	ng/L	PFHpA	F	5#	SW	Colorado	В	Norit	GAC 300
2441	Dickenson,	2016	>96#	Percent	5.8#	<0.25#	ng/L	PFHxS	F	5#	SW	Colorado	В	Norit	GAC 300
2441	Dickenson,	2016	>96#	Percent	6.4#	<0.25#	ng/L	PFBS	F	5#	SW	Colorado	В	Norit	GAC 300
2505	Cummings,	2015	>72 to >93#	Percent	18 to 72	<5	ng/L	PFNA	F		SW	Logan System Birch	В	Calgon	F-400

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Drinking Water Treatment for PFOS

Ineffective Treatments

Conventional Treatment Low Pressure Membranes Biological Treatment (including slow sand filtration) Disinfection Oxidation Advanced oxidation

Effective Treatments

Anion Exchange Resin (IEX) High Pressure Membranes Powdered Activated Carbon (PAC) Granular Activated Carbon (GAC)

Extended Run Time Designed for PFAS Removal

Percent Removal

> 89 to > 98

90 to 99	-	Effective
93 to 99	-	Effective
10 to 97	-	Effective for only select applications
0 to 26	_	Ineffective

- Effective

PAC Dose to Achieve

 50% Removal
 16 mg/l

 90% Removal
 >50 mg/L

 Dudley et al., 2015
 >50 mg/L





To provide tools to accurately predict the performance and cost of treating PFAS in drinking waters

Treatability Databas	e			Cost Models
SEPA United States Environmental Protection Agency			SE	PAR Environmental Protection
Environmental Topics Laws & Regulations About EPA	Search EPA.gov 9.	1	Env	vironmental Topics Laws & Regulations About EPA Search EPA.gov ۹
telated Topics: Water Research	CONTACT US SHARE 👔 🕑 🖾		Relate	d Topics: Safe Drinking Water Act CONTACT US SHARE 🔶 🎯 🔘
Drinking Water Treatability Data	base (TDB)		Dr	inking Water Treatment Technology Unit
Provides information on the control of contaminants PA's <u>Drinking Water Treatability Database (TBD)</u> is an easy to use tool that provides eferenced information on the control of contaminants in drinking water. It was designed for use by utilities, first responders to spills or emergencies, consultants and technical assistance iroviders, treatment process designers, and researchers. Information in the TDB is gathered from thousands of literature sources and assembled on use the Notamente researce.	Quick Start Find a Contaminant Find a Treatment Process	Environmental Topics Laws & Regulations About EPA Related Topics: Water Research	Search EPA.gov	St Models and Overview of Technologies ing Water Treatment Technology Unit Cost Models and executive orders require EPA to estimate compliance costs for new drinking water standards. The three major components ce costs are: Int ng rative costs
Navigating the TDB. Capabilities Future Updates Support		Environmental Technologies Des Tool (ETDOT)	ign Option	echnologies remove or destroy pollutants (such as arsenic, disinfection byproducts, and waterborne pathogens). treatment costs, EPA developed several engineering models using a bottom-up approach known as work breakdown structure VBS models:
		Adsorption treatment modeling for contaminant removal from drinking water and wastewater The Environmental Technologies Design Option Tool (ETDOT) is a suite of software models that provides engineers with the capability to evaluate and design systems that use granular activated carbon or ion exchange resins for the removal of contaminants, including PFAS, from drinking water and wastewater.	Access ETDOT Access the FIDOT software, manuals, and more at FIDOT GitHub site. EXIT	

Suite of Models

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ETDOT was developed by National Center for Clean Industrial and Treatment Technologies at Michigan Technological University (MTU). In 2019, EPA signed an agreement with MTU to make this suite of adsorption models available to the public at no cost.

Related EPA Resources

Applications



Search: EPA ETDOT

Available Modeling Tools



https://www.epa.gov/water-research/environmental-technologies-design-option-tool-etdot

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Available Modeling Tools

Models available at the GitHub site:

- Adsorption Design Software for Windows (AdDesignS) Version 1.0
- Advanced Oxidation Process Software (AdOx) Version 1.0.2
- Aeration System Analysis Program (ASAP) Version 1.0
- Biofilter Design Software Version 1.0.27
- Continuous Flow Pore Surface Diffusion Model for Modeling • Powdered Activated Carbon Adsorption Version 1.0
- Dye Study Program (DyeStudy) Version 1.0.0
- Predictive Software for the Fate of Volatile Organics in Municipal Wastewater Treatment Plants (FaVOr) Version 1.0.11
- Ion Exchange Design Software (IonExDesign) Version 1.0.0
- Software to Estimate Physical Properties (StEPP) Version 1.0





Adsorption Design Softwa	re (AdDesignS™)
	Version 1.0 Commercial Version Copyright 1994-1999
National Center for Clean Industrial and Treatment Technologies Michigan Technological University Houghton, Michigan	David R. Hokanson David W. Hand John C. Crittenden
MichiganTech	Tony N. Rogers Eric J. Oman

The engines are written in FORTRAN with a Visual Basic front end



Expected Interest/Users

ETDOT Modeling Tool

It is expected that these established and new models will be useful to the following:

- State primacy personnel interested in evaluating data sets
- Water utilities with experience in running models
- Consulting engineers
- University academicians

Incorporation of Complex Mechanisms



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EPA's Drinking Water Cost Models

Adsorptive media Anion exchange Biological treatment Cation exchange GAC Greensand filtration Microfiltration / ultrafiltration Multi-stage bubble aeration



Non-treatment Packed tower aeration Point of Use (POU)/ Point of Entry (POE)* Reverse Osmosis / Nanofiltration UV disinfection

<u>epa.gov/dwregdev/drinking-water-treatment-technology-unit-cost-models-and-overview-technologies</u>
Search: EPA WBS

*POU/POE temporarily taken off web. Please contact Rajiv Khera

Set EPA

Costs for PFAS Treatment: One GAC Example



Costs can be generated for various sizes, contaminants, and even POU scenarios

Primary Assumptions:

- Two vessels in series
- 20 min Empty Bed Contact Time (EBCT) Total
- Bed Volumes Fed

 1,1-DCA = 5,560 (7.5 min EBCT)
 Shorter Chain PFCA = 4,700
 Gen-X = 7,100
 Shorter Chain PFS = 11,400
 PFOA = 31,000
 PFOS = 45,000
 7% Discount rate
- Mid Level Cost



Costs for PFAS Treatment: One IEX Example



Costs can be generated for various sizes, contaminants, and even POU scenarios

Primary Assumptions:

- Two vessels in series
- 3 min EBCT Total
- Bed Volumes Fed:
 - Shorter Chain PFCA = 3,300

Gen-X = 47,600

- Shorter Chain PFS = 34,125
- PFOA = 112,500

PFOS = 191,100

- 7% Discount rate
- Mid Level Cost

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Future Plans

Treatability Database

• Further update treatability database with new references

Performance Models

- Update Graphical User Interface to work with Windows 10
- Provide Python code for pore surface diffusion model (PSDM GAC) to automate the optimization routines for:
 - Specific throughput and carbon use rate calculations for multiple scenarios
 - Automated fitting of parameters
 - Automated optimal bed configuration
 - Automated optimal Empty Bed Contact Time (EBCT) selection
 - Automated evaluation of bed replacement frequency
 - Evaluation of multiple feed conditions
 - Evaluation of multiple flow conditions
 - Automated fitting and predicting lead/lag operations





Future Plans

Performance Models (continued)

- Provide code for ion exchange models for
 - Include competition (e.g., inorganic ions and PFAS)
 - Continuous flow (columns) and batch (isotherm and kinetics)
 - Gel (HSDM) and macro porous (PSDM) resins
 - with automation features

Cost Models

• Further updates to the cost models

Combined Models

- Further merge Treatability Database, performance models, and cost models
- Further merge the Treatability Database with <u>EPA's</u> <u>CompTox Chemicals Dashboard</u>









Performance Model Demonstration

To provide tools to accurately predict the performance and cost of treating PFAS in drinking waters

Treatability Databas	e			Cost Models
SEPA United States Environmental Protection Agency			\$	EPA United States
Environmental Topics Laws & Regulations About EPA	Search EPA.gov		E	nvironmental Topics Laws & Regulations About EPA Search EPA.gov Q
Related Topics: Water Research	CONTACT US SHARE (f) 🕑 🖾		Rela	ted Topics: Safe Drinking Water Act
Drinking Water Treatability Data	base (TDB)		D	rinking Water Treatment Technology Unit
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Related EPA Resources

Applications

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ETDOT Software



https://www.epa.gov/water-research/environmental-technologies-design-option-tool-etdot Search: EPA ETDOT

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GitHub: ETDOT Software

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	ucchejbb committed 4548d46 c	n Jan 31	🕤 7 commits 🛭 🖓 1 branch 💿 1 tag	No description, website, or topics provided.	
	code	Loading all files	5 months ago	🛱 Readme	4
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	in manuals	adding manuals folder	5 months ago	on Mar 23	Current Release
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	Version 1.0: Copyright 1994–2	2005			
	David R. Hokanson				
	 David W. Hand John C. Crittenden 				
	Tony N. Rogers				
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https://github.com/USEPA/Environmental-Technologies-Design-Option-Tool



https://github.com/USEPA/Environmental-Technologies-Design-Option-Tool/releases/tag/1.0

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Readme.md

Disclaimer:

The United States Environmental Protection Agency (EPA) GitHub project code is provided on an "as is" basis and the user assumes responsibility for its use. EPA has relinquished control of the information and no longer has responsibility to protect the integrity, confidentiality, or availability of the information. Any reference to specific commercial products, processes, or services by service mark, trademark, manufacturer, or otherwise, does not constitute or imply their endorsement, recommendation or favoring by EPA. The EPA seal and logo shall not be used in any manner to imply endorsement of any commercial product or activity by EPA or the United States Government.

Installation Instructions

This software requires *Administrator Rights* to a computer to install and to run. Files are installed directly to a folder X:\ETDOT10... where X is the system main drive.

- 1. Download zip file in the release tab.
- 2. Unzip/Unpack zip file
- 3. Run setup.exe and follow prompts
- 4. When prompted enter license key: CAADV0-R74JM-QXCNP-7EER9-1AT72
- 5. To run each module in Windows 7 or newer: Edit *properties* of the program to be run and select Compatibility Tab and "run in compatibility mode". Select Windows 98/Me from the Compatibility Mode dropdown menu.

Available Users manuals will be located in the modules subfolder within the help folder.

Notes on current software

The **ETDOT** suite of software packages consists of a FORTRAN engine with a Visual Basic (version 6) graphical user interface. The VB6 portion of the code relies on ActiveX control files which are located in the repository, however, these are an older coding standard and no longer supported with current versions of Visual Studio 20##. Precompiled engine files are included.

https://github.com/USEPA/Environmental-Technologies-Design-Option-Tool

Accessing ETDOT Software



Start Menu: 'ETDOT Programs'

Software & Instruction Manuals are listed



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AdDesignSTM Software

AdDesignS - (Untitled)			Results for the PSDM (No Reaction	ns Present)			
File Phase Run Results Options Databases Hel	p		Results for:				Close
Water Properties:	Fixed Bed Properties:		New Component	-	Length of th	he MTZ (cm): 53.773	
Pressure 1.00 atm	Adsorber Database	1		Time (days)	BVT(m³/m³)	VTM(m³/kg) C (mg/L)	
Temperature 15.0 C	Red Length 2.77		5% of influent conc.	28.64	4.39E+03	9.76 2.50	
Component Properties:	Bed Diameter 3.05		95% of influent conc.	34.59	4.83E+03	11.78 47.50	
New Component	Bed Mass 9072		Treatment Objective	28.64	4.39E+03	9.76 2.50	
	Elowrate 0.0358	n ³ /e v		,			
	EBCI 564						Calif Chalm
New Commente	Bed Density 0.4497	ía/mL)	50059.4			_	
	Bed Porosity 0.440	-	50059.4				Both 💌
<u>Add</u> <u>Delete</u> <u>Edit Properties</u>	Superficial Velocity 17.648	(m/hr)					
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Water Treatment Models

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		🗅 Readme.md	Update Readme.md	2 days ago	No releases published Create a new release	
				P	Packages No packages published	
		READIVIE for wate	er Treatment Models		Publish your first package	
		Tools in this repository: 1. Adsorption Model for Granular 2. Ion Exchange Model (IonExchar	Activated Carbon (PSDM Folder) igeModel Folder)		Contributors 3	
		Both tools are programmed in Pytho their respective folders. These tools well treatment technologies (Granul	on. Additional resources and information a focus on predicting water treatment unit c ar Activated Carbon and Ion Exchange Res	ssociated with either can be found in operation effectiveness, specifically how is) will work for removing contaminants.	etter in the second sec	
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https://github.com/USEPA/Water Treatment Models

Water Treatment Models

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		Both tools are programmed their respective folders. Thes well treatment technologies	in Python. Additional resources and information e tools focus on predicting water treatment unit (Granular Activated Carbon and Ion Exchange Re	associated with either can be found in operation effectiveness, specifically how sis) will work for removing contaminants.	aatsov	vo dalsovo		
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https://github.com/USEPA/Water_Treatment_Models



https://github.com/USEPA/Water Treatment Models

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PSDM Model Examples



Single Compound





Modeling Fouling





Multicomponent competitive modeling

Models can be used to perform a variety of different analyses or applications



Applying the PSDM Mode

Fitting Pilot / Full-scale Data



Predicting Results for Consistent Design

Allows for comparison across technologies by cost

Allows for Predicting other Scenarios

- Other designs: number of contactors, contactor Empty Bed Contact Times (EBCTs), different treatment goals, etc.
- Other influent conditions: Changing concentrations of PFAS or background constituents, changing demand, etc.



Impact of Bed Size for Other Conditions

Relationship Between Specific Throughput and EBCT



Example: What is the optimal Empty Bed Contact Time (EBCT) choice in the future?

The models can evaluate EBCT for various PFAS for different PFAS at different influent concentrations, effluent goals, etc.

Modeled Specific Throughput can help determine EBCT or expected treatment volumes for single component treatment objectives



Design/Operation Evaluations

Example: How should the beds be deployed and replaced?

> The models can evaluate how many and how often beds need to be brought online or replaced





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Lead/Lag Operation Evaluations

Example: What advantage can be gained with a lead/lag configuration?

- > The models can predict this increased carbon use rate.
- Note: Lead/lag operations can complicate modeling efforts, especially for those that take Natural Organic Matter (NOM) fouling/preloading into effect.



In an ideal situation, the first column would be completely saturated to maximize the use of the carbon.

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Example: Lead/Lag Operation Evaluations

Single use:

Time to 70 ppt breakthrough (PFOA & PFOS)

> 154 days (based on 1st column)



Lead/Lag:

Time to 70 ppt breakthrough (PFOA & PFOS)

> 236 days (based on 1st column)





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