



**Support Document for the Fourth Six-Year
Review of Drinking Water Regulations for
Acrylamide and Epichlorohydrin**

Office of Water (4607M)
EPA-815-R-24-023
February 2024
www.epa.gov/safewater

Disclaimer

This report is in support of the revise/take no action decisions for EPA's Fourth Six-Year Review of Existing Drinking Water Standards Federal Register Notice. This report is intended to provide technical background for the fourth Six-Year Review.

This document is not a regulation itself and it does not substitute for the Safe Drinking Water Act (SDWA) or EPA's regulations. Mention of trade names or commercial products does not constitute endorsement or recommendation for use.

Table of Contents

Table of Exhibits.....	ii
Abbreviations and Acronyms	iii
1 Introduction.....	1-1
2 Regulatory Background	2-1
2.1 Polymer Chemistry	2-1
2.2 Polymer Use in Water Treatment	2-1
2.3 Current Regulatory Framework	2-1
2.4 Regulatory Basis	2-2
3 Supporting Information for Potential Regulatory Revision to Increase Public Health Protection.....	3-1
3.1 Improvements in Manufacturing – NSF Data on Residual Monomer Content.....	3-1
3.2 Regulations and Guidelines in Other Countries – European Union (EU), United Kingdom (UK), Canada, World Health Organization (WHO)	3-3
3.3 Food and Drug Administration (FDA) Regulations.....	3-3
3.4 Acrylamide and Epichlorohydrin Occurrence in Drinking Water	3-4
3.4.1 Polymer Uses in Water Treatment.....	3-5
3.4.2 Frequency of Polymer Use.....	3-5
4 Six Year Review (SYR) Recommendation.....	4-1
5 References.....	5-1

Table of Exhibits

Exhibit 3-1. Summary of NSF International Product Testing Results for Acrylamide and Epichlorohydrin, 2019-2021	3-1
Exhibit 3-2. Summary of NSF International Product Testing Results for Acrylamide and Epichlorohydrin, 2013-2016	3-2
Exhibit 3-3. Summary of NSF International Product Testing Results for Acrylamide and Epichlorohydrin, 2005-2007	3-2
Exhibit 3-4. Comparison of Acrylamide and Epichlorohydrin Drinking Water Guidelines	3-3
Exhibit 3-5. FDA Regulations on Acrylamide and Epichlorohydrin Content in Food Additives	3-4

Abbreviations and Acronyms

AWWA	American Water Works Association
CFR	Code of Federal Regulations
CWSS	Community Water System Survey
DL	detection limit
DWI	Drinking Water Inspectorate
EPA	U.S. Environmental Protection Agency
epi-DMA	epichlorohydrin-dimethylamine copolymer
EU	European Union
FDA	Food and Drug Administration
FR	Federal Register
GW	ground water
MCL	maximum contaminant level
MCLG	maximum contaminant level goal
MDL	method detection limit
mg/kg	milligrams per kilogram
mg/L	milligrams per liter
µg/L	micrograms per liter
MWH	Montgomery Watson Harza
NHMRC	National Health and Medicine Research Council
NPDWR	National Primary Drinking Water Regulation
NSF	NSF International
poly-DADMAC	polydiallyldimethyl ammonium chloride
ppm	parts per million
SDWA	Safe Drinking Water Act
SW	surface water
SYR	Six-Year Review
SYR 1	First Six-Year Review
SYR 2	Second Six-Year Review
SYR 3	Third Six-Year Review
SYR 4	Fourth Six-Year Review
UK	United Kingdom
US	United States
WHO	World Health Organization

1 Introduction

The U.S. Environmental Protection Agency (EPA) has completed its fourth Six-Year Review (SYR 4) of national primary drinking water regulations (NPDWRs). The 1996 Safe Drinking Water Act (SDWA) Amendments require the U.S. Environmental Protection Agency (EPA or the Agency) to periodically review existing NPDWRs. Section 1412(b)(9) of SDWA reads:

...[t]he Administrator shall, not less than every 6 years, review and revise, as appropriate, each primary drinking water regulation promulgated under this title. Any revision of a national primary drinking water regulation shall be promulgated in accordance with this section, except that each revision shall maintain, or provide for greater, protection of the health of persons.

The primary goal of the Six-Year Review (SYR) process is to identify NPDWRs for possible regulatory revision. Although the statute does not define when a revision is “appropriate,” as a general benchmark, EPA considered a possible revision to be “appropriate” if, at a minimum, it presents a meaningful opportunity to:

- improve the level of public health protection, and/or
- achieve cost savings while maintaining or improving the level of public health protection.

For SYR 4, EPA implemented the NPDWR review protocol that it developed for the first Six-Year Review (SYR 1) (USEPA, 2003), including minor revisions developed during the second review process (SYR 2) (USEPA, 2009b) and the third review process (SYR 3) (USEPA, 2016). Following the review method in the protocol, EPA sought new information that might affect the following NPDWR components:

- **Maximum Contaminant Level Goals (MCLGs; the health goal)** – for some contaminants new health effects assessments completed since the MCLG was promulgated or last revised provide a revised reference dose and/or cancer classification.
- **Maximum Contaminant Levels (MCLs; the enforceable standard)** – for some contaminants, the MCL is equal to the MCLG, and the health effects assessment indicates potential to revise the MCLG. Improvements in analytical feasibility as indicated by the practical quantitation limit may also indicate feasibility to set the MCL closer to the MCLG.
- **Treatment Technique** (sometimes established in lieu of an MCL) – new information on health effects, analytical feasibility, or treatment feasibility may suggest a possibility to revise treatment technique.
- **Other Regulatory Requirements (Monitoring)** – Other regulatory revisions may be appropriate if information suggest that changes in monitoring standards (e.g., frequency) could reduce health risks or costs while maintaining or improving the level of public health protection.

As part of its SYR 4, EPA obtained and evaluated new information pertaining to the NPDWRs for acrylamide and epichlorohydrin, which are regulated by treatment techniques. This document provides background information on these contaminants and the current NPDWR in section 2. Section 3 provides a description of the new information that EPA obtained and analyzed during SYR 4. EPA’s determination is in section 4.

2 Regulatory Background

Acrylamide and epichlorohydrin occur in drinking water as treatment impurities. They are primarily introduced as residuals in polymers and copolymers used for water treatment, although they can also be present in contact surfaces used in storage and distribution systems.

2.1 Polymer Chemistry

Polymers are long-chained molecules made up of units called monomers. If a polymer contains a single type of monomer, it is called a homopolymer; if it contains two or more types of monomers, it is called a copolymer.

Polymers used in water treatment are characterized by their molecular weight, the predominant sign of their charged sites (anionic, cationic, or nonionic), and their charge density. A simple polymer is nonionic polyacrylamide, the homopolymer formed from acrylamide monomer. More complex polymers may have varying patterns of copolymerization or cross-linked structures.

Acrylamide monomer is an input used to make anionic and cationic copolymers, as well as nonionic polyacrylamide. Nonionic polyacrylamide may also be hydrolyzed to form an anionic polymer. Epichlorohydrin is an input used to make various cationic copolymers, notably epichlorohydrin-dimethylamine copolymer (epi-DMA). Epichlorohydrin polymers that contain amine monomers are known as polyamines; however, polyamines in general do not necessarily contain epichlorohydrin.

2.2 Polymer Use in Water Treatment

During polymer manufacture, a small fraction of the monomer units do not polymerize and remain in the commercial polymer as an impurity. Thus, there is a potential for exposure to these contaminants as residual monomers in polymers following their addition to water being treated for drinking water use.

The polymers most often used in drinking water treatment are polyacrylamides (anionic, nonionic, and less commonly cationic), epi-DMA (cationic), and polydiallyldimethyl ammonium chloride (poly-DADMAC; cationic). Thus, anionic and nonionic polymers used in drinking water treatment are primarily polyacrylamides, while cationic polymers vary in their composition (Levine et al., 2004; AWWA, 1999). Section 3.4.1 provides more detailed discussion of polymer use in water treatment.

EPA reviewed new information on monomer residuals as part of SYR 4 to determine if improvements in the technology or manufacturing process now allow production of the polymer with lower residual monomer content, and to reassess the appropriateness of the maximum allowable dosage of the polymers and copolymers.

2.3 Current Regulatory Framework

EPA proposed drinking water regulations for acrylamide and epichlorohydrin on May 22, 1989 (54 *FR* 22062, USEPA, 1989) and promulgated final drinking water regulations on January 30,

1991 (56 FR 3526, USEPA, 1991). The NPDWR for epichlorohydrin contains an MCLG of zero based on a cancer classification of B2, probable human carcinogen (56 FR 3526, USEPA, 1991). Similarly, EPA established an MCLG of zero for acrylamide based on a B2 cancer classification (56 FR 3526, USEPA, 1991). In an updated health effects assessment, EPA concluded that acrylamide remains carcinogenic (USEPA, 2010).

EPA regulated these contaminants using a treatment technique requirement—in lieu of an MCL—because of the absence of standardized analytical methods for their measurement in water. The treatment technique requirement for these contaminants limits the allowable monomer levels in polymers and copolymers, as well as the polymer dose in treatment. EPA selected this option because methods are available for measurement of residual monomer in polymer products and these levels are routinely measured by manufacturers. These limits are:

- Acrylamide: 0.05 percent acrylamide in polymers/copolymers (e.g., 500 milligrams per kilogram or mg/kg) and maximum dosage of 1 part per million (ppm) (e.g., 1 milligram per liter or mg/L); and
- Epichlorohydrin: 0.01 percent residual epichlorohydrin concentration in polymers/copolymers and a maximum dosage of 20 ppm.

Under EPA’s regulation, each water system is required to certify, in writing, to the Primacy authority (e.g., a state or EPA Region) that the product of the polymer dose and the residual monomer level do not exceed the specifications in the NPDWR. A system may use third-party or manufacturer’s certification in lieu of testing for the residual monomer level.

The NSF International (NSF), a third party organization, tests and certifies water treatment chemicals. Chemicals must meet NSF/ANSI Standard 60, *Drinking Water Treatment Chemicals – Health Effects*, which sets out requirements for treatment chemicals based on human health protection (NSF, 2016). The requirements for acrylamide and epichlorohydrin based polymers in Standard 60 are based on EPA’s treatment technique requirements. Thus, NSF 60 certification of a polymeric coagulant aid containing acrylamide or epichlorohydrin indicates that users are in compliance of EPA’s regulation when a product is used as specified (i.e., for the intended purpose and up to the maximum usage level indicated by NSF).

2.4 Regulatory Basis

In setting the treatment technique requirement for acrylamide, EPA used a level of 0.05 percent residual acrylamide monomer and a polymer dose of 1 ppm, based on the maximum acceptable levels in EPA’s Drinking Water Additives Advisory Program. This program was operational at the time of the rulemaking but was later terminated. The residual monomer level was considered to be the lowest level that manufacturers could feasibly achieve at the time the regulation was promulgated and corresponded to similar requirements in Food and Drug Administration (FDA) regulations governing polyacrylamide as a secondary direct food additive (54 FR 22062, USEPA, 1989). The dose was based on typical doses of polyacrylamide used in drinking water treatment. Polymers may be sold as dry powder, emulsions, solutions, or dispersions with less than 100 percent active polymer (AWWA, 2006a, 2006b). The doses specified in the rule are on an active polymer basis.

The 1991 rule similarly limited residual epichlorohydrin and polymer dose, using a level of 0.01 percent residual epichlorohydrin dosed at 20 ppm. The monomer level and dose were those accepted for epichlorohydrin-based polymers within the framework of the EPA's Drinking Water Additives Advisory Program. As with acrylamide, the monomer level was considered the lowest feasible level for manufacturers, and the dose was based on typical doses of polymers containing epichlorohydrin (54 *FR* 22062, USEPA, 1989).

In estimating exposure from the allowable levels of acrylamide and epichlorohydrin, EPA used the worst-case assumption that all residual monomer carries over to the finished water, resulting in finished water concentrations from polymer use of 0.5 micrograms per liter ($\mu\text{g/L}$) and 2 $\mu\text{g/L}$, respectively (54 *FR* 22062, USEPA, 1989). EPA assumed that an additional 10 percent of acrylamide and epichlorohydrin—that is, up to 0.05 $\mu\text{g/L}$ of acrylamide and 0.2 $\mu\text{g/L}$ of epichlorohydrin—enter drinking water via leaching from these other materials or from raw water.

Thus, taking into account exposure as residual monomer in water treatment chemicals and through leaching from surfaces in contact with water, total human exposure to acrylamide via drinking water will at a maximum be approximately 0.55 $\mu\text{g/L}$. Similarly, total human exposure to epichlorohydrin via drinking water will at maximum be 2.2 $\mu\text{g/L}$.

3 Supporting Information for Potential Regulatory Revision to Increase Public Health Protection

3.1 Improvements in Manufacturing – NSF Data on Residual Monomer Content

In 2022, NSF provided EPA with results of NSF analyses of acrylamide monomer in polyacrylamides and free epichlorohydrin in polyamines.¹ NSF performed the analyses for approval of these products against NSF/ANSI Standard 60. The NSF data provided to EPA included 307 analytical results for acrylamide (in dry and emulsion forms) and 105 analytical results for epichlorohydrin. NSF conducted the analyses between 2019 and 2021. **Exhibit 3-1** provides summary statistics.

Based on data provided by NSF, EPA determined that the free epichlorohydrin residual levels in the polyamines products tested and certified are consistently and substantially less than the residual level in the treatment technique. All analyses for residual epichlorohydrin have non-detection results. The detection limit (DL; 20 mg/kg) is one-fifth the residual level in the treatment technique (0.01 percent or 100 mg/kg).

Product analysis results for acrylamide are generally less than the residual level in the treatment technique. The mean concentration acrylamide in the polymer is about one-fifth the residual level in the treatment technique, and the 90th percentile result is nearly one-half the residual level in the treatment technique. The analysis solution for 1 of the 307 samples exceeds the treatment technique residual level (0.05 percent or 500 mg/kg). That sample had a residual level of 760 mg/kg. The corresponding maximum use level of 0.6 mg/L is lower than the treatment technique, however. Thus, the maximum concentration would not exceed 0.5 µg/L.

Exhibit 3-1. Summary of NSF International Product Testing Results for Acrylamide and Epichlorohydrin, 2019-2021

Contaminant	Number of Analyses and Detections ¹	Detection Limit (DL) (mg/kg)	Maximum (mg/kg)	90 th Percentile (mg/kg)	Mean ² (mg/kg)	Median ² (mg/kg)	Minimum (mg/kg)	Treatment Technique (mg/kg) ³
Acrylamide	307 (216)	10	760	260	104	49	Non-detect	500
Epichlorohydrin	105 (0)	20	NA	NA	NA	NA	NA	100

Source: EPA analysis of data provided by Purkiss (2022)

NA = not applicable because all results are below the DL.

1. Number of detection results shown in parenthesis.

2. Includes non-detection results for acrylamide at the reported DL.

3. Treatment technique residual monomer content converted from percent to mg/kg (e.g., 100 mg/kg = 100/10⁶ = 0.0001 = 0.01%).

For comparison purposes, **Exhibit 3-2** and **Exhibit 3-3** provide the same type of results from the SYR 3 and SYR 2 cycles, respectively. Trends over time indicate increased product sample

¹ Purkiss (2022). NSF did not provide any confidential business information such as which manufacturers were included in the analyses. NSF only provided vectors of testing results.

testing from 150 total samples during the second cycle to 412 during SYR 4. The DLs remain constant over time. Non-detection of epichlorohydrin is also consistent over time. The median acrylamide concentration has declined somewhat from 60 mg/kg during SYR 2 to 48 mg/kg in SYR 4.

Exhibit 3-2. Summary of NSF International Product Testing Results for Acrylamide and Epichlorohydrin, 2013-2016

Contaminant	Number of Analyses and Detections ¹	Detection Limit (DL) (mg/kg)	Maximum (mg/kg)	90 th Percentile (mg/kg)	Mean ² (mg/kg)	Median ² (mg/kg)	Minimum (mg/kg)	Treatment Technique (mg/kg) ³
Acrylamide	244 (81)	10	490	270	105	55	Nondetect	500
Epichlorohydrin	90 (0)	20	NA	NA	NA	NA	NA	100

Source: EPA analysis of data provided by Purkiss (2016)

NA = not applicable because all results are below the DL.

1. Number of detection results shown in parenthesis. Excludes analyses of well-drilling aids.

2. Includes non-detection results for acrylamide at the reported DL of 10 mg/kg.

3. Treatment technique residual monomer content converted from percent to mg/kg (e.g., 100 mg/kg = 100/10⁶ = 0.0001 = 0.01%.

Exhibit 3-3. Summary of NSF International Product Testing Results for Acrylamide and Epichlorohydrin, 2005-2007

Contaminant	Number of Analyses and Detections	Detection Limit (DL) (mg/kg)	Maximum (mg/kg)	90 th Percentile (mg/kg)	Mean ² (mg/kg)	Median ² (mg/kg)	Minimum (mg/kg)	Treatment Technique (mg/kg) ³
Acrylamide	66 (45)	10	420	250	98	60	10	500
Epichlorohydrin	84 (0)	20	NA	NA	NA	NA	NA	100

Source: USEPA, 2009c

NA = not applicable because all results are below the DL.

1. Number of detection results shown in parenthesis. Excludes analyses of well-drilling aids.

2. Includes non-detection results for acrylamide at the reported DL of 10 mg/kg.

3. Treatment technique residual monomer content converted from percent to mg/kg (e.g., 100 mg/kg = 100/10⁶ = 0.0001 = 0.01%.

The NSF data indicate potential to lower the residual monomer limits for acrylamide and epichlorohydrin. Given consistent non-detection results for epichlorohydrin in all three data sets (2005-2007, 2013-2016, and 2019-2021), EPA concludes that a value equal to or slightly greater than the DL is feasible. The feasible limit for acrylamide is more difficult to identify. Although acrylamide was not detected in 30 percent of samples taken in 2019-2021 and two-thirds of samples taken from 2013-2016, detected quantities ranged as high as 760 mg/kg, which exceeds the current treatment technique limit of 0.05 percent or 500 mg/kg. Therefore, during SYR 3, EPA reviewed drinking water limits or guidelines applied elsewhere.

3.2 Regulations and Guidelines in Other Countries – European Union (EU), United Kingdom (UK), Canada, World Health Organization (WHO)

Regulations in other areas of the world are generally more stringent than the current EPA NPDWR for acrylamide and epichlorohydrin in drinking water. **Exhibit 3-4** provides a comparison of recommendations and guidelines used elsewhere to EPA’s current regulations.

Canada does not have drinking water guidelines for acrylamide or epichlorohydrin (Health Canada, 2014). Nine of 13 provinces require that drinking water additives be certified to meet health-based standards such as NSF/ANSI Standard 60 (NSF, 2016), which effectively limits the monomer residual content to the same concentration as the United States (U.S.) standard. Areas under federal jurisdiction have the same requirements (Health Canada, 2016).

Exhibit 3-4. Comparison of Acrylamide and Epichlorohydrin Drinking Water Guidelines

Country/Region	Regulation or Guideline	Acrylamide	Epichlorohydrin
EPA	Residual Monomer [a]	0.05%	0.01%
EPA	Maximum Dosage [b]	1 mg/L	20 mg/L
EPA	Water concentration ([a] x [b]) x 1000	0.5 µg/L	2 µg/L
Canada	NSF/ANSI Standard 60 certification required in 9 of 13 provinces	NSF/ANSI Standard 60 certification required in 9 of 13 provinces	NSF/ANSI Standard 60 certification required in 9 of 13 provinces
UK ¹	Residual Monomer	0.02%	0.002%
UK ¹	Dosage	0.25 mg/L (average) 0.5 mg/L (maximum)	2.5 mg/L (average) 5 mg/L (maximum)
UK ¹	Concentration in Water	0.1 µg/L (maximum)	0.1 µg/L (maximum)
EU ²	Concentration in Water	0.1 µg/L	0.1 µg/L
WHO ³	Concentration in Water	0.5 µg/L	0.4 µg/L
Australia ⁴	Concentration in Water	0.2 µg/L	0.5 µg/L

1. DWI (2010 and 2016). Residual monomer for epichlorohydrin is inferred from concentration limit and dosage limits.

2. EU (2007). The enforceable parameter or limit is the residual monomer concentration in the water based on the maximum release from polymer.

3. WHO (2011). The WHO’s guideline for epichlorohydrin is a provisional guideline value because of uncertainties in the health database.

4. NHMRC (2016). The guideline value for epichlorohydrin is below the limit of determination.

3.3 Food and Drug Administration (FDA) Regulations

When EPA set the residual monomer level for acrylamide, it considered the monomer levels specified in FDA regulations governing polyacrylamides as secondary direct food additives. During SYR 4, EPA reviewed current FDA regulations on acrylamide and epichlorohydrin to ensure that monomer limits have not changed.

FDA regulates polyacrylamides for several applications, including uses as secondary direct food additives (e.g., in resins used for sugar clarification), indirect food additives (e.g., in the manufacture of food container in contact with aqueous and fatty foods), and, in one case, a direct

food additive. The cases where FDA regulates residual monomer content are summarized in Exhibit 3-5.

Exhibit 3-5. FDA Regulations on Acrylamide and Epichlorohydrin Content in Food Additives

Contaminant	Type of additive	Application	Monomer limit	Code of Federal Regulation (CFR)
Acrylamide (in several polymers)	Secondary direct	Sugar clarification	0.05%	21 CFR 173.5; 21 CFR 173.10 21 CFR 173.310 21 CFR 173.315 21 CFR 173.357
		Boiler water additive	0.05%	
		Fruit/vegetable washing	0.2%	
		Corn syrup manufacture	0.05%	
Acrylamide (in several polymers)	Indirect	Paper for containers	0.2%	21 CFR 176.110 21 CFR 176.170
		Paper food contact surfaces	0.2%	
Acrylamide (in several polymers)	Direct	Film former in imprinting of soft shell gelatin capsules	0.2%	21 CFR 172.255
Epichlorohydrin (in epi-DMA copolymer)	Secondary direct	Sugar clarification	0.001%	21 CFR 173.60 21 CFR 173.357
		Corn syrup manufacture	0.001%	

Exhibit 3-5 shows that FDA’s regulations on acrylamide content are the same as EPA’s current treatment technique requirement, or less stringent than EPA’s requirement. This may be because the potential human exposure to acrylamide from food additives is expected to be lower than exposure to acrylamide via drinking water. FDA’s requirement for epichlorohydrin in epi-DMA copolymer is more stringent, suggesting that there may be an opportunity for EPA to revise the drinking water monomer residual limit downward.

FDA (2007) also reports acrylamide formation as a reaction between asparagine and reducing sugars. The formation occurs during cooking or thermal processing of foods such as potato products (French fries and potato chips) and cereal products (such as cookies, crackers, and toasted bread), and coffee. The estimated mean daily intake ranged from 21 to 60 µg, which is substantially higher than the daily exposure from drinking water at the maximum allowable concentration of 0.5 µg (e.g., daily exposure would be 1 µg for an adult consumption rate of 2 liters per day). FDA published *Guidance for Industry Acrylamide in Foods* with recommendations for growers, manufactures, and food service operators to reduce acrylamide-formation precursors as well as acrylamide formation during food processing (FDA, 2016).

3.4 Acrylamide and Epichlorohydrin Occurrence in Drinking Water

There is a potential for exposure to these contaminants as residual monomers in polymers when they are used in water treatment (as direct additive). Finished water may also contain acrylamide and epichlorohydrin because of raw water contamination and because of leaching from components and materials used in drinking water treatment, storage and distribution systems (indirect additives). These components and materials may contain polymers based on acrylamide or epichlorohydrin. Although there is little actual data on the occurrence of acrylamide and epichlorohydrin in raw water, EPA’s occurrence estimates in finished water have been based

largely on the release of residue or impurity from their use as direct additives in drinking water treatment.

3.4.1 Polymer Uses in Water Treatment

The polymers most often used in drinking water treatment are polyacrylamides (anionic, nonionic, and less commonly cationic), epi-DMA, and another cationic polymer - poly-DADMAC. Thus, anionic and nonionic polymers used in drinking water treatment are primarily polyacrylamides, while cationic polymers vary in their composition (Levine et al., 2004; AWWA, 1999). Polymer type and dose vary by treatment objective and source water quality. The following discussion describes different applications.

Cationic polymers such as epi-DMA are often used in combination with metal ion coagulants at the time of coagulation, operating primarily by charge neutralization. Under some conditions, when the coagulated particles have positive surface charges, anionic polymer may be used instead. As coagulant aids, polymers can permit reductions of 40-80 percent in the dose of metal ion coagulants, thus reducing sludge volumes. Cationic polymers are also sometimes used by themselves as coagulants for direct filtration, reducing solids volumes in comparison to inorganic coagulants. However, polymers by themselves are ineffective at removing dissolved material (Levine et al., 2004; MWH, 2005).

Polymers may also be added after coagulation or flocculation, as flocculation or filter aids. In these applications, they are intended to produce larger, denser flocs that settle faster, or to strengthen the floc so that filtration can more effectively remove particulate and organic matter. These applications rely primarily on particle bridging, where a single polymer chain adsorbs on the surfaces of different particles. Bridging takes place with high molecular weight (i.e., long chain) polymers that are nonionic or have low charge densities (Levine et al., 2004; MWH, 2005).

3.4.2 Frequency of Polymer Use

Estimates of polymer use for drinking water treatment vary. EPA's 2006 Community Water System Survey (CWSS) indicated that polymer use among surface water (SW) systems ranged from a low of 16 percent among systems serving 100 or fewer people to a high of almost 57 percent among systems serving more than 100,000 people (USEPA, 2009a). Among ground water (GW) systems, polymer use did not exceed 3 percent for any size category (USEPA, 2009a). The American Water Works Association's (AWWA's) WATER:STATS database (cited in Levine, 2004) indicates higher use rates: 66 percent of SW treatment plants surveyed used a polymer, predominantly cationic; 13 percent of GW treatment plants used a polymer, with anionic polymers most often used.

EPA does not have data indicating what percent of the population served by public water systems might be exposed to either acrylamide or epichlorohydrin through drinking water. Based on higher usage frequencies among SW systems—especially larger systems—that tend to use cationic polymers, polyamines are likely to dominate polymer use. The infrequent use of anionic polymers among GW systems suggests potential acrylamide exposure frequency is low by comparison.

EPA also does not have estimates of the actual concentrations to which people might be exposed. Populations exposed to cationic polymers such as epi-DMA are already benefitting from the use of polymers with lower residual monomer levels. Therefore, revising the allowable residual monomer level of epichlorohydrin to equal or nearly equal its detection level will have no benefits or costs. Revising the allowable residual monomer level of acrylamide might result in health risk reductions if public water systems switch to products with lower residual monomer levels. The NSF 2013-2016 data indicate, however, that two-thirds of the products do not have detectable acrylamide levels and almost 90 percent of products have half the allowable acrylamide residual.

4 Six Year Review (SYR) Recommendation

EPA's review of recent NSF analyses of acrylamide and epichlorohydrin impurities in polymers indicates potential to revise the NPDWRs for these contaminants. Specifically, NSF data indicate that it is feasible to reduce the allowable monomer residual levels in water treatment polymers.

The NSF data also indicate that because monomer residuals are almost always less than EPA's treatment technique requirements, the health benefits associated with the lower impurity levels are already being realized by communities throughout the country. In particular, there will be no benefits associated with epi-DMA use because epichlorohydrin residual levels are already below the DL, which is one-fifth of the allowable residual under the NPDWR.

The feasible limit for acrylamide is uncertain. Although acrylamide was not detected in approximately 30 percent of the sample results provided by NSF, concentrations in the remaining samples ranged from 11 mg/kg to 760 mg/kg. Approximately 85 percent of the detected concentrations are less than or equal to 250 mg/kg, which is half the monomer limit in the treatment technique. Although EPA does not have information to characterize the actual distribution of acrylamide monomer levels in polymers used for drinking water treatment, the majority of tested products already pose lower health risks than required under the current treatment techniques. Revising the treatment techniques will not result in health risk reductions for these products. Therefore, EPA concludes that a regulatory revision is not likely to provide a meaningful opportunity to improve public health protection.

Given resource limitations, competing workload priorities, and administrative costs and burden to states to adopt any regulatory changes associated with the rulemaking, the revisions to these NPDWRs are a low priority. EPA does not believe a revision to the NPDWR for acrylamide or epichlorohydrin is appropriate at this time.

5 References

- American Water Works Association (AWWA). 1999. Water Quality and Treatment: A Handbook of Community Water Supplies, 5th edition. New York: McGraw Hill.
- AWWA. 2006a. AWWA Standard B452-06: EPI-DMA Polyamines. Effective date August 1, 2006.
- AWWA. 2006b. AWWA Standard B453-06: Polyacrylamide. Effective date August 1, 2006.
- Drinking Water Inspectorate (DWI), United Kingdom. 2010. What are the drinking water standards? London, UK. Retrieved from <http://dwi.defra.gov.uk/consumers/advice-leaflets/standards.pdf>.
- DWI. 2016. List of Approved Products for Use in the Water Supply in the United Kingdom. Retrieved from <http://www.dwi.gov.uk/drinking-water-products/approved-products/soslistcurrent.pdf>.
- European Union. 2007. Section 2: Standards for Drinking Water Quality. Retrieved from <https://www.epa.ie/pubs/advice/drinkingwater/privatewatersupplieshandbook/Section%202.pdf>.
- Food and Drug Administration (FDA). 2007. “Acrylamide, Furan, and the FDA”. Reprinted from Food Safety Magazine. Retrieved from <http://www.fda.gov/downloads/Food/FoodborneIllnessContaminants/ChemicalContaminants/UCM194483.pdf>.
- FDA. 2016. Guidance for Industry Acrylamide in Foods. Retrieved from <http://www.fda.gov/downloads/Food/GuidanceRegulation/GuidanceDocumentsRegulatoryInformation/ChemicalContaminantsMetalsNaturalToxinsPesticides/UCM374534.pdf>.
- Health Canada. 2014. Guidelines for Canadian Drinking Water Quality Summary Table. Prepared by the Federal-Provincial-Territorial Committee on Drinking Water. Retrieved from http://www.hc-sc.gc.ca/ewh-semt/pubs/water-eau/sum_guide-res_recom/index-eng.php.
- Health Canada. 2016. Guidance for Providing Safe Drinking Water in Areas of Federal Jurisdiction – Version 2. Retrieved from <http://healthycanadians.gc.ca/publications/healthy-living-vie-saine/water-federal-eau/index-eng.php?page=text>.
- Levine, Audrey D., Brian A. Bolto, and David R. Dixon. 2004. Reactions of Polyelectrolytes with Other Water Treatment Chemicals. AWWA Research Foundation Report #2509.
- Montgomery Watson Harza (MWH). 2005. Water Treatment: Principles and Design. Hoboken: John Wiley and Sons.
- National Health and Medicine Research Council (NHMRC), Australia. 2016. Australian Drinking Water Guidelines (2011) – Updated November 2016. Retrieved from <https://www.nhmrc.gov.au/guidelines-publications/eh52>.

- NSF International (NSF). 2016. Water Matters. Retrieved from https://www.nsf.org/newsroom_pdf/water_municipal_water_matters_1606.pdf.
- Purkiss, D. 2016. Personal communication from D. Purkiss (NSF International) to R. Khera, (US EPA), dated December 2, 2016.
- Purkiss, D. 2022. Personal communication from D. Purkiss (NSF International) to R. Khera, (US EPA), dated March 18, 2022.
- U.S. Environmental Protection Agency (USEPA). 1989. National Primary and Secondary Drinking Water Regulations; Proposed Rule. *Federal Register*. Vol. 54, No. 97. p. 22062, May 22, 1989.
- USEPA. 1991. National Primary Drinking Water Regulations—Synthetic Organic Chemicals and Inorganic Chemicals; Monitoring for Unregulated Contaminants; National Primary Drinking Water Regulations Implementation; National Secondary Drinking Water Regulations; Final Rule. *Federal Register*. Vol. 56, No. 30. p. 3526, January 30, 1991.
- USEPA. 2003. EPA Protocol for Review of Existing National Primary Drinking Water Regulations. EPA 815-R-03-002. Retrieved from <https://www.epa.gov/sites/default/files/2014-12/documents/815r03002.pdf>.
- USEPA. 2009a. 2006 Community Water System Survey, Volume II: Detailed Tables and Survey Methodology. EPA 815-R-09-002. Retrieved from <https://nepis.epa.gov/Exe/ZyPDF.cgi?Dockey=P1009USA.txt>.
- USEPA. 2009b. EPA Protocol for the Second Review of Existing National Primary Drinking Water Regulations (Updated). EPA 815-B-09-002. Retrieved from <https://www.epa.gov/sites/default/files/2014-12/documents/815b09002.pdf>.
- USEPA. 2009c. Water Treatment Technology Feasibility Support Document for Chemical Contaminants for the Second Six-Year Review of National Primary Drinking Water Regulations. EPA Report 815-B-09-007. Retrieved from <https://www.epa.gov/sites/default/files/2014-12/documents/815b09007.pdf>.
- USEPA. 2010. Integrated Risk Information System (IRIS) Chemical Assessment Summary: Acrylamide; CASRN 79-06-1. Retrieved from https://cfpub.epa.gov/ncea/iris/iris_documents/documents/subst/0286_summary.pdf.
- USEPA. 2016. EPA Protocol for the Third Review of Existing National Primary Drinking Water Regulations. EPA 810-R-16-007. Retrieved from <https://www.epa.gov/sites/default/files/2016-12/documents/810r16007.pdf>.
- World Health Organization (WHO). 2011. Guidelines for Drinking-water Quality: Incorporating First Addendum, Vol. 1, Recommendations. 4th Edition. World Health Organization. Geneva, Switzerland. Retrieved from http://apps.who.int/iris/bitstream/10665/44584/1/9789241548151_eng.pdf.